

## Chapter 4 Question #2

A 12 volt battery is given a 20 minute charge at a steady 10amps. During the process, the battery gets hot and loses 10kJ of heat. What is the change in internal energy of the battery?

- 1) Not enough information is given to solve this problem
- 2) +134kJ
- 3) -7.6kJ
- 4) +4.4kJ

**Chapter 4 Question 2 Answer:**

**(2) +134 kJ**

$P=VI = 12 \text{ V} * 10 \text{ A} = 120 \text{ W}$ . The total energy given to the battery is therefore power \* time =  $120 \text{ W} * 1200 \text{ seconds} = 144,000 \text{ J}$ . The battery then loses 10 kJ so the net increase in internal energy is +134 kJ.

Class Response:

Question 3 : Question 3

