

HOMEWORK 2 - 10 pts total

2 pts

$$1. \quad V = 1 \text{ m}^3$$

$$S_R = 0.2 \quad \text{residual saturation}$$

$$n = 0.25 \quad \text{porosity}$$

$$\begin{aligned} \text{Mass balance: original vol. of TCE in soils} \\ = 1 \text{ m}^2 \times 0.25 \times 0.2 = 0.05 \text{ m}^3 \end{aligned}$$

$$\text{Density of TCE} = 1.47 \text{ g/cm}^3 = 1470 \text{ kg/m}^3$$

$$\text{Initial mass TCE} = 1470 \frac{\text{kg}}{\text{m}^3} \cdot 0.05 \text{ m}^3 = 74 \text{ kg TCE}$$

Assume TCE dissolves into rain water to solubility = 1100 mg/L

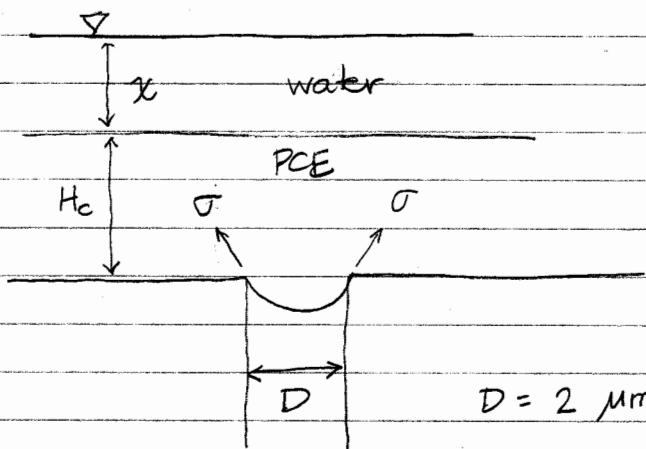
$$\text{Volume of rain} = 1 \text{ m}^2 \times 500 \text{ mm/yr} = 0.5 \text{ m}^3/\text{yr}$$

$$\begin{aligned} \text{Mass carried by rain} &= 0.5 \text{ m}^3/\text{yr} \times 1000 \text{ mg/L} \cdot 10^{-3} \text{ m}^3 \\ &= 0.5 \times 10^6 \text{ mg/yr} = 0.5 \text{ kg/yr} \end{aligned}$$

$$\text{Time to remove TCE} = \frac{74 \text{ kg TCE}}{0.5 \text{ kg TCE/yr}} = 150 \text{ years}$$

3 pts

2.



Properties of PCE

$$\rho_p = 1.63 \text{ g/cm}^3$$

$$\sigma = 44.4 \text{ dyn/cm}$$

$$1 \text{ dyn} = 1 \frac{\text{g cm}}{\text{sec}^2}$$

$$P_{NAPL} = \rho_w x g + \rho_p H_c g$$

$$P_{WATER} = \rho_w (x + H_c) g$$

$$a) \Delta P = (\rho_p - \rho_w) H_c g = \frac{2\sigma}{r} \quad \text{for circular fracture}$$

$$0.63 \frac{\text{g}}{\text{cm}^3} H_c 9.8 \frac{\text{m}}{\text{sec}^2} = 2 \frac{44.4 \frac{\text{g}}{\text{sec}^2}}{1 \mu\text{m}}$$

$$H_c = \frac{2 \cdot 44.4}{0.63 \cdot 9.8} \frac{\text{g/sec}^2}{\mu\text{m g/cm}^3 \text{ m/sec}^2}$$

$$\frac{1}{10^{-6} \text{ m} / (10^{-2})^3 \text{ m}^3 \cdot \text{m}} = \text{m}$$

$$= 14 \text{ m}$$

b) For linear fracture, σ is exerted on both sides of fracture of length L

Hydrostatic pressure is exerted on opening DL

$$\Delta P D L = \sigma 2L$$

$$\Delta P = \Delta P H_c g = (\rho_p - \rho_w) H_c g$$

$$H_c = \frac{20}{\Delta P g D} = 7.2 \text{ m} \quad (\text{one half value for circular fracture})$$

3 pts

3. $i = 0.005 \quad t = 10 \text{ years}$

As first approximation, consider only advective travel. Thus, downstream wells need to extend to distance L :

$$L = ut = \frac{Ki}{H} t$$

For each type soil, use literature values of K and n^*

soil	$K \text{ (cm/s)}$	n_e	L
a. coarse sand	$9 \times 10^{-5} - 6 \times 10^{-1}$	0.2 - 0.35	48 Km
b. fine sand	$2 \times 10^{-5} - 2 \times 10^{-2}$	0.1 - 0.3	1600 m
c. silt	$1 \times 10^{-7} - 2 \times 10^{-3}$	0.34 - 0.61	94 m
d. clay	$1 \times 10^{-9} - 5 \times 10^{-7}$	0.34 - 0.60	2 cm

To be conservative, I used high value of K and low value of n from range in Daniel. L is computed to one-digit accuracy commensurate with accuracy of K .

For finer soils, diffusion could be more important than advection.

$$D_h = D_d^* + \alpha_L u$$

$$D_d^* \approx 10^{-9} \text{ m/s}$$

Based on chart from Gelhar et al. (1992) given in Lecture 2, for 1 to 10 m scale,

$$\alpha_L \approx 1 \text{ m}$$

* Technical Protocol for Evaluating Natural Attenuation of Chlorinated Solvents in Ground Water Section C-3 pg. 206 of 248
<http://www.epa.gov/ada/download/reports/protocol.pdf>

3. (continued)

Velocities and dispersion coeffs are:

	u (m/yr)	$a_L u$ (m^2/s)	D_h (m^2/s)
b. fine sand	7	2×10^{-7}	2×10^{-7}
c. silt	0.5	1×10^{-8}	1×10^{-8}
d. clay	0.004	1×10^{-10}	1×10^{-9}

Evaluate transport with equation given in Lecture 3:

$$C = \frac{C_0}{2} \operatorname{erfc} \left(\frac{R_d x - ut}{\sqrt{4 R_d D_h t}} \right)$$

Assume $\frac{C}{C_0} = 0.01$ is threshold

$R_d = 1$ to be conservative

$$\frac{C}{C_0} = 0.01 = \frac{1}{2} \operatorname{erfc} \left(\frac{x - ut}{\sqrt{4 D_h t}} \right)$$

$$\operatorname{erfc}(z) = 0.02 \text{ at } z \approx 1.2$$

$$\frac{x - ut}{\sqrt{4 D_h t}} = 1.2$$

$$x = ut + 2.4 \sqrt{D_h} \sqrt{t}$$

Evaluate for different soils:

	u (m/yr)	D_h (m^2/s)	D_h (m^2/yr)	x
b. fine sand	7	2×10^{-7}	7	90 m
c. silt	0.5	1×10^{-8}	0.5	10 m
d. clay	0.004	1×10^{-9}	0.004	0.5 m

Note: diffusion dominates advection in clay, is about equal to advection in silt

Problem 3

$i = 0.005$
 $t = 10 \text{ years}$
 $\alpha_L = 1 \text{ m}$
 $D_d^* = 1.00E-09 \text{ m/s}$

Soil	K (cm/s)	n_e	u (m/s)	u (m/yr)	L (m)	$D_h (\text{m}^2/\text{s})$	$D_h (\text{m}^2/\text{yr})$	x (m)
a. coarse sand	1.00E-02	0.24	2.08E-06	6.57E+01	657.00	2.08E-06	6.57E+01	657
b. fine sand	1.00E-05	0.24	2.08E-09	6.57E-02	0.66	3.08E-09	9.72E-02	0.7
c. silt	1.00E-06	0.35	1.43E-10	4.51E-03	0.05	1.14E-09	3.60E-02	0.05
d. clay	1.00E-08	0.4	1.25E-12	3.94E-05	0.0004	1.00E-09	3.16E-02	0.0006

$$L = ut = (Ki/n_e) t$$

$$D_h = D_d^* + \alpha_L u$$

3. (continued)

Benzene computation:

$$K_{oc} = 83 \text{ mg/l}$$

foc - check Technical Protocol

can assume a value of $foc \approx 0.001$ for all

$$\rho_b = 1.65 \text{ g/cm}^3$$

$$n = 0.3$$

$$R_d = 1 + \frac{\rho_b f_{oc} K_{oc}}{n} \approx 1.5$$

Monitoring for benzene would only need to be about $\frac{2}{3}$ the distance of monitoring for MTBE.

2 pts

4.

$$K = 1.6 \times 10^{-4} \text{ cm/s} = 0.45 \text{ ft/day}$$

$$i = 0.035$$

$$L = 400 \text{ ft}$$

$$u = \frac{Kl}{n}$$

$$\text{assume } n \text{ for clay-silt} = 0.3 \quad \therefore u = 0.05 \frac{\text{ft}}{\text{day}}$$

$$\text{Ground-water travel time} = \frac{L}{u} = \frac{400 \text{ ft}}{0.05 \text{ ft/d}} \\ = 8000 \text{ days} = 22 \text{ years}$$

Need to find retardation factors for PCE, TCA, xylene

From Table C.3.3 of Technical Protocol for Evaluating Natural Attenuation of Chlorinated Solvents in Ground Water

$$f_{oc} \approx 0.002 \text{ for glacial till}$$

From Table A-1* K_{oc} values are:

$$\text{PCE} \quad K_{oc} = 364 \text{ L/kg}$$

$$\text{TCA} \quad 152$$

$$\text{X} \quad 240$$

$$\text{TCE} \quad 126$$

$$\text{Bulk density, } \rho_b = 1.65 \frac{\text{g}}{\text{cm}^3} \text{ (typical value)} = 1.65 \frac{\text{kg}}{\text{L}}$$

$$\text{Porosity, } n = 0.3$$

Compound	K _{oc} (L/kg)	R _d = 1 + $\frac{\rho_b K_d}{n}$ = 1 + $\frac{\rho_b K_{oc} f_{oc}}{n}$	Travel time (yrs)
PCE	364	3.0	66
TCA	152	1.8	40
X	240	2.3	51
TCE	126	1.7	37

$$\text{Travel time} \quad \frac{L}{u/R_d} = R_d \frac{L}{u}$$

* From Chemical reference information in course reader

○ 4. (cont)

Whether compounds reach brook depends on chemistry
PCE and TCE/A will degrade if conditions
are anaerobic
Xylene will degrade in aerobic

Given mix of chemicals at site, both anaerobic
and aerobic zones are possible - I would
expect chemicals to degrade over these long
travel times, but not necessarily from
areas of the site nearer the brook.