

**1.85 WATER AND WASTEWATER TREATMENT ENGINEERING  
HOMEWORK 4 – DUE MARCH 10, 2005**

Question 1 (5 points)

A suspension of three sizes of spherical particles is to be filtered at a rate of 15 m/hr through a 60-cm rapid sand filter. The bed sand has a diameter of 0.5 mm and a porosity of 0.4. Particle counting indicates that there are 5, 25, and 125 mg/L each of 0.1, 1.0, and 10.0-micron diameter particles, respectively. The particles have a density of 1.05 g/cm<sup>3</sup>. The water temperature is 25C.

- a. Assuming discrete particle settling, determine the single collector efficiency,  $\eta$ , for each particle size using the theoretical model of Yao et al. (1971) as given in the notes for Lectures 7 and 8. (2 points)
- b. If the attachment efficiency,  $\alpha$ , is 0.2, what is the concentration of each particle after the water passes through the filter bed? (2 points)
- c. If the attachment efficiency,  $\alpha$ , is 1.0, what is the concentration of each particle after the water passes through the filter bed? (1 point)

Question 2 (5 points)

Viessman and Hammer Problem 11.22.

The water defined by the analysis given below is to be softened by excess-lime (and soda ash) treatment.

- a. Sketch an meq/L bar graph (1 point).
- b. Calculate the softening chemicals required (3 points).
- c. Draw a bar graph for the softened water after recarbonation and filtration, assuming that 80% of the alkalinity is in the bicarbonate form (1 point).

CO <sub>2</sub>	8.8 mg/L		
Ca <sup>2+</sup>	40.0 mg/L	ALK (HCO <sub>3</sub> <sup>-</sup> )	135 mg/L
Mg <sup>2+</sup>	14.7 mg/L	SO <sub>4</sub> <sup>2-</sup>	29.0 mg/L
Na <sup>+</sup>	13.7 mg/L	Cl <sup>-</sup>	17.8 mg/L