

Hints for Problem 15.26

Some of the parameters required to solve 15.26 (S&H) need to be found in the literature (e.g. Perry's). We have provided those below to save time:

1. The concentration of water in air at 80°F, 1 atm and 80% relative humidity (from psychrometric chart)

$$= \frac{0.0177 \text{ lb H}_2\text{O}}{\text{lb dry air}}$$

2. The diffusivity of water vapor in air at 1 atm, 80°F

$$D_i = 0.26 \times 10^{-4} \frac{\text{m}^2}{\text{s}}$$

3. The viscosity of air at 80 °F

$$\mu = 1.75 \times 10^{-5} \frac{\text{kg}}{\text{m s}}$$

Additional advice:

- *Pay attention to units throughout the problem.*
- You can use the ideal gas law to calculate the density (in lb/ft³) of the gas entering the bed, which is a mixture of water vapor and air.
- Assume that the cross-sectional area of the bed is 1 ft².
- Use the equation $\rho_p = \frac{\rho_b}{1 - \epsilon_b}$ to calculate ρ_p , the particle density, (lb gel/ ft³ particles) from the density of the silica given in the problem statement $\rho_b = 39 \frac{\text{lb}}{\text{ft}^3}$ and the void fraction $\epsilon_b = 0.47$.
- The units of the equilibrium constant, K, should be in $\frac{\text{lb H}_2\text{O} / \text{ft}^3 \text{ gel}}{\text{lb H}_2\text{O} / \text{ft}^3 \text{ gas}}$ to use Equation 15-106 (S&H).
- If using Excel to solve the Klinkenberg equation, use erf carefully and remember that erf (-z) = - erf (z).