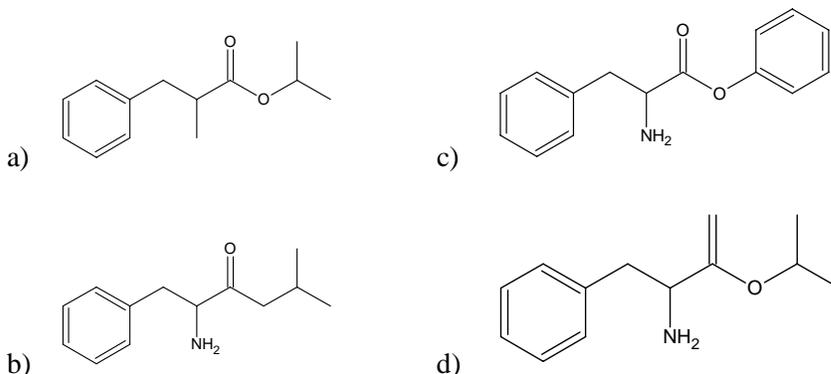
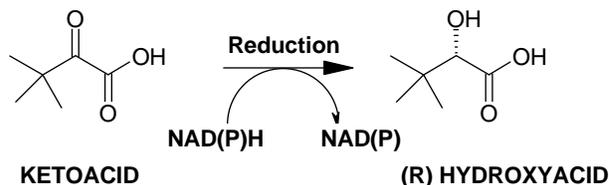




- 4) Look again at the bioconversion shown in Question 3(d). Using the same (unnamed) enzyme, evaluate whether the bioconversion is likely to occur for the following compounds. For each compound, give a brief statement as to why you think it will or will not be acted upon by the enzyme.



- 5) You would like to identify an enzyme that will perform the following reduction, giving a high yield of the *R*-enantiomer.



You have at your disposal 10 different enzyme preparations that are candidate catalysts. For each enzyme, you set up a small reaction, charged with 1 g/L of substrate and let the reaction run at room temperature for 3 hours. You then obtain the following data (each value is the concentration, in g/L, of the compound listed):

Enzyme	Ketoacid	<i>R</i> -hydroxyacid	<i>S</i> -hydroxyacid
Red1	0.48	0.26	0.26
Red2	0.12	0.24	0.64
Red3	0.76	0.19	0.05
Red4	0.75	0.03	0.22
Red5	0.25	0.71	0.04
Red6	0.33	0.66	0.01
Red7	0.20	0.39	0.41
Red8	0.10	0.86	0.04
Red9	0.99	0.003	0.007
Red10	0.62	0.31	0.07

For each enzyme, calculate the conversion and the EE. Which enzymes would *not* be carried forward for process development? Of the remaining, which is your top candidate for process development and why? (Courtesy of Merck & Co., Inc. Used with permission.)

- 6) An enzyme with a  $K_M$  of  $1 \times 10^{-3}$  M was assayed using an initial substrate concentration of  $3 \times 10^{-5}$  M. After 2 min, 5 percent of the substrate was converted. How much substrate will be converted after 10 min, 30 min, 60 min? How long must the reaction be run to achieve 99% conversion? (Assume that the enzyme follows Michaelis-Menten kinetics.)