

12.480 Handout #3

Reading

Stormer (1975) *Am. Mineral* **60**, 667-674.

Whitney and Stormer (1977) *Am. Mineral* **62**, 687-691.

Andersen and Lindsley (1981) *GCA* **45**, 847-853.

Supplementary Reading

Griffen, Dana T. *Silicate Crystal Chemistry*, Chapter 2. New York, NY: Oxford University Press, 1992. ISBN: 0195044423

Ribbe, P. H. (ed.) *Feldspar Mineralogy (Reviews in Mineralogy, Vol. 2, 2nd ed.)*. Washington, D. C.: Mineralogical Society of America, 1983. ISBN: 0939950146

Two feldspar themometry-barometry

The use of coexisting plagioclase and alkali feldspars to predict temperature and pressure of equilibration was first proposed by Barth (1951) who used natural phase assemblages to calibrate a thermometer. Stormer and Whitney and Stormer subsequently develop a simple thermodynamic model for treating feldspar equilibria. The important assumptions for this model were:

1. Use one of the conditions of equilibrium.

$$\mu_{Ab}^{AF} = \mu_{Ab}^{PF}$$

where AF = alkali feldspar and PF = plagioclase feldspar.

2. Assume that Or content in PF has no effect on a_{Ab}^{PF} and that An content in AF has no effect on a_{Ab}^{AF} .

The equilibrium condition can be written

$$\mu_{Ab}^{AF} = \mu_{Ab}^{\circ AF} + RT \ln a_{Ab}^{AF}$$

$$\mu_{Ab}^{PF} = \mu_{Ab}^{\circ PF} + RT \ln a_{Ab}^{PF}$$

3. Assume that the standard state chemical potential for pure Ab in both phases was the same. Then

$$\begin{aligned} 0 &= RT \ln \frac{a_{Ab}^{AF}}{a_{Ab}^{PF}} \\ &= RT \ln \frac{\gamma_{Ab}^{AF} X_{Ab}^{AF}}{\gamma_{Ab}^{PF} X_{Ab}^{PF}} \end{aligned}$$

4. Assume $\gamma_{Ab}^{PF} = 1$, so

$$\ln \frac{X_{Ab}^{AF}}{X_{Ab}^{PF}} = \ln \gamma_{Ab}^{AF}$$

$$\ln \gamma_{Ab}^{AF} = \frac{1}{RT} (2W_{GO_r} - W_{GAb}) (X_{O_r}^{AF})^2 + 2(W_{GAb} - W_{GO_r}) (X_{O_r}^{AF})^3$$

Several of the simplifications of the Whitney and Stormer thermometer have been dealt with in subsequent models of feldspar equilibria, and some have not.

First Assumption

The μ^o 's are not equal and we require two sets of W 's! So

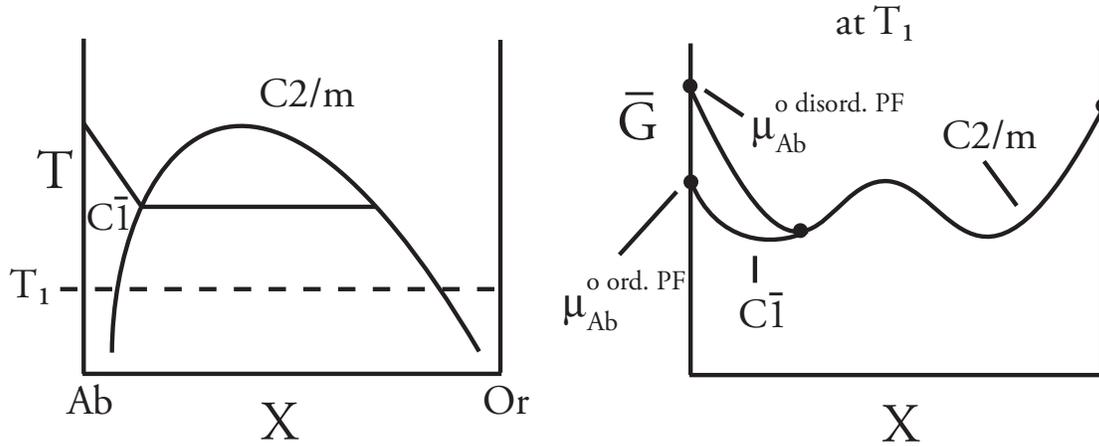
$$\mu_{Ab}^{AF} = \mu_{Ab}^{PF}$$

$$\mu_{Ab}^{AF} = \mu_{Ab}^{o\text{disord}AF} + RT \ln \gamma_{Ab} X_{Ab}$$

where W 's were appropriate to disordered 0.

$$\mu_{Ab}^{PF} = \mu_{AB}^{o\text{ordered plag}} + RT \ln \gamma_{Ab} X_{Ab}$$

potentially we also need W 's for a different solution.



Second Assumption

A ternary solution model would be more appropriate.

$$\begin{aligned} RT \ln \gamma_1 = & W_{G12} X_1^2 X_2 + W_{G21} X_2^2 X_1 \\ & + W_{G13} X_1^2 X_3 + W_{G31} X_3^2 X_1 \\ & + W_{G23} X_2^2 X_3 + W_{G32} X_3^2 X_2 \end{aligned}$$

Margules formulations for ternary and quaternary solutions were developed by Wohl (1946, 1953).

Lindsley and Anderson (1981) go through the exercise of deriving an expression for a ternary asymmetric model. These models, like the ones for binary systems, assume that a polynomial of degree 2 (symmetric) or degree 3 (asymmetric) in component 2 and 3 are adequate models of the excess free energy of mixing.