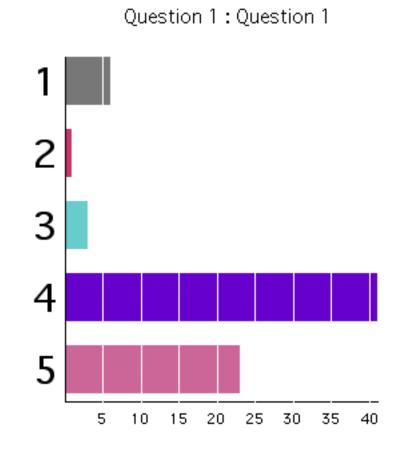
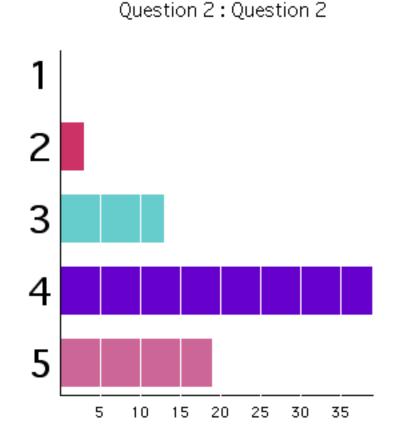
I am able to state the First Law and to define heat, work, thermal efficiency and the difference between various forms of energy.

- 1) Strongly disagree
- 2) Disagree
- 3) Somewhat agree
- 4) Agree
- 5) Strongly agree



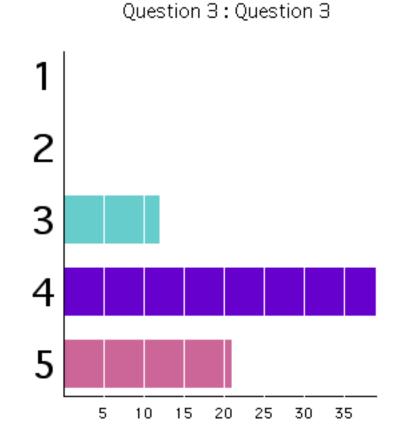
I am able to identify and describe energy exchange processes (in terms of various forms of energy, heat and work) in aerospace systems.

- 1) Strongly disagree
- 2) Disagree
- 3) Somewhat agree
- 4) Agree
- 5) Strongly agree



I am able to explain at a level understandable by a high school senior or non-technical person how various heat engines work (e.g. a refrigerator, an IC engine, a jet engine).

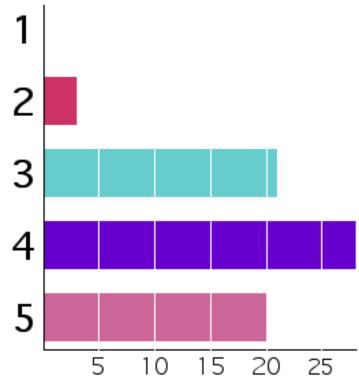
- 1) Strongly disagree
- 2) Disagree
- 3) Somewhat agree
- 4) Agree
- 5) Strongly agree



I am able to apply the steady-flow energy equation or the First Law of Thermodynamics to a system of thermodynamic components (heaters, coolers, pumps, turbines, pistons, etc.) to estimate required balances of heat, work and approx flow

1) Strongly disagree 1

- 2) Disagree
- 3) Somewhat agree
- 4) Agree
- 5) Strongly agree



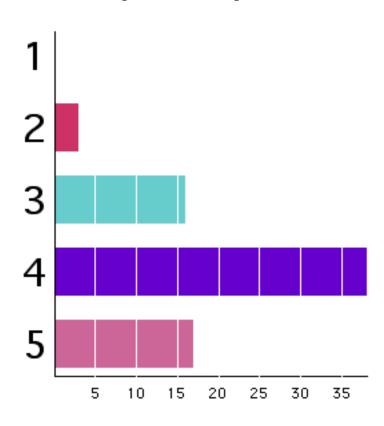
Question 4: Question 4

I am able to explain at a level understandable by a high school senior or non-technical person the concepts of path dependence/independence and reversibility/irreversibility of various thermodynamic processes, to represent these in terms of changes in thermodynamic state, and to cite examples of how these would impact the performance

Question 5: Question 5

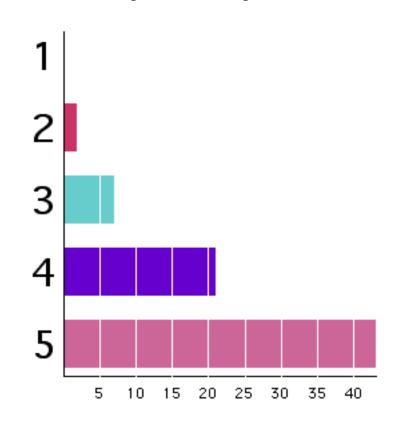
propulsion systems.

- 1) Strongly disagree
- 2) Disagree
- 3) Somewhat agree
- 4) Agree
- 5) Strongly agree



I am able to apply ideal cycle analysis to simple heat engine cycles to estimate thermal efficiency and work as a function of pressures and temperatures at various points in the cycle.

- 1) Strongly disagree
- 2) Disagree
- 3) Somewhat agree
- 4) Agree
- 5) Strongly agree



Ouestion 6: Ouestion 6