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2.500 Desalination and Water Purification
Spring 2009

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2.500 Desalination & Water Purification

Spring 2009
Tuesday/Thursday 1:00-2:30
Professor John H. Lienhard V

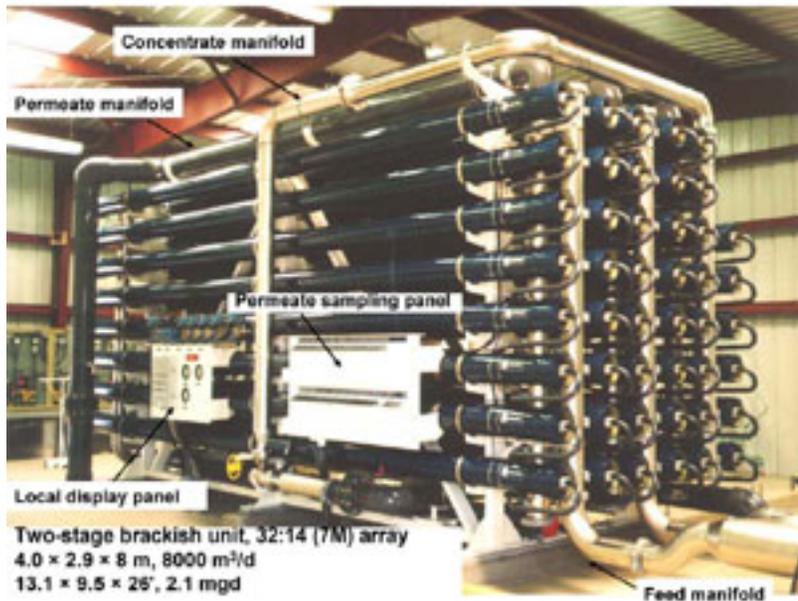


Figure from Wilf, M., and M. Balaban. *Membrane Desalination and Membrane Filtration*. L'Aquila, Italy: European Desalination Society, 2007. Used with permission.



2.500 Desalination and Water Purification



Image from [Wikimedia Commons](#),

Tanzania

- More than 1 billion people lack access to clean drinking water
- Half the hospital beds in the world are occupied by patients with easily prevented water-borne disease
- Half the people in the world do not have sanitation systems as good as those in Ancient Rome.

- In 2000, unsafe water mortality amounted to 80 million years of lost life (*Science*, 25 Jan 2008)

- This situation is expected to get WORSE.



Yangon, Myanmar

May 2008
...after cyclone

Images removed due to copyright restrictions.

Please see http://www.nytimes.com/slideshow/2008/05/05/world/0505-MYANMAR_index.html

<http://graphics8.nytimes.com/images/2008/05/05/nytfrontpage/23097528.JPG>

Photo source:

The New York Times, 7 May 2008

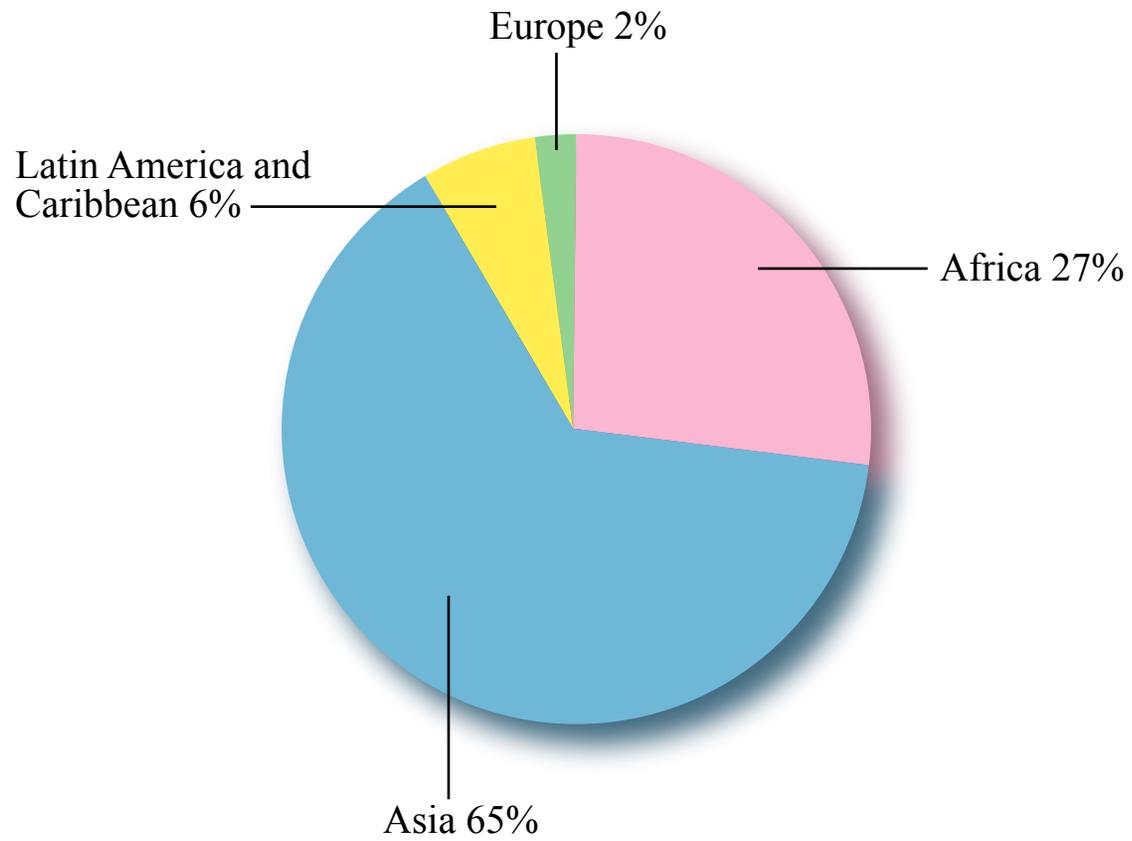


Department of Mechanical Engineering
Massachusetts Institute of Technology

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Please see: <http://www.flickr.com/photos/andrewheavens/100063338/>

http://jimbicentral.typepad.com/photos/uncategorized/2007/09/18/water_scarcity.jpg



Water supply, distribution of unserved populations

Figure by MIT OpenCourseWare.

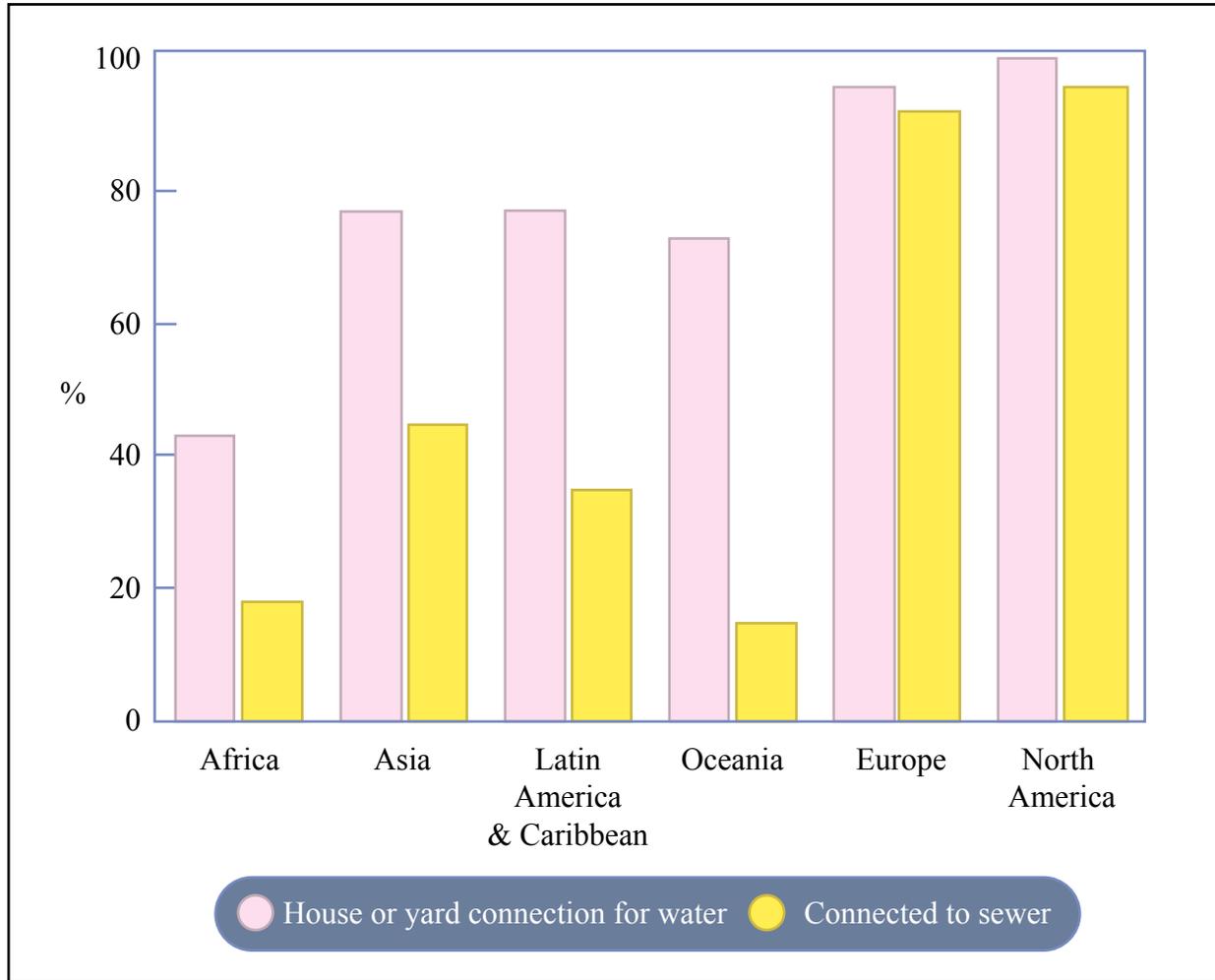
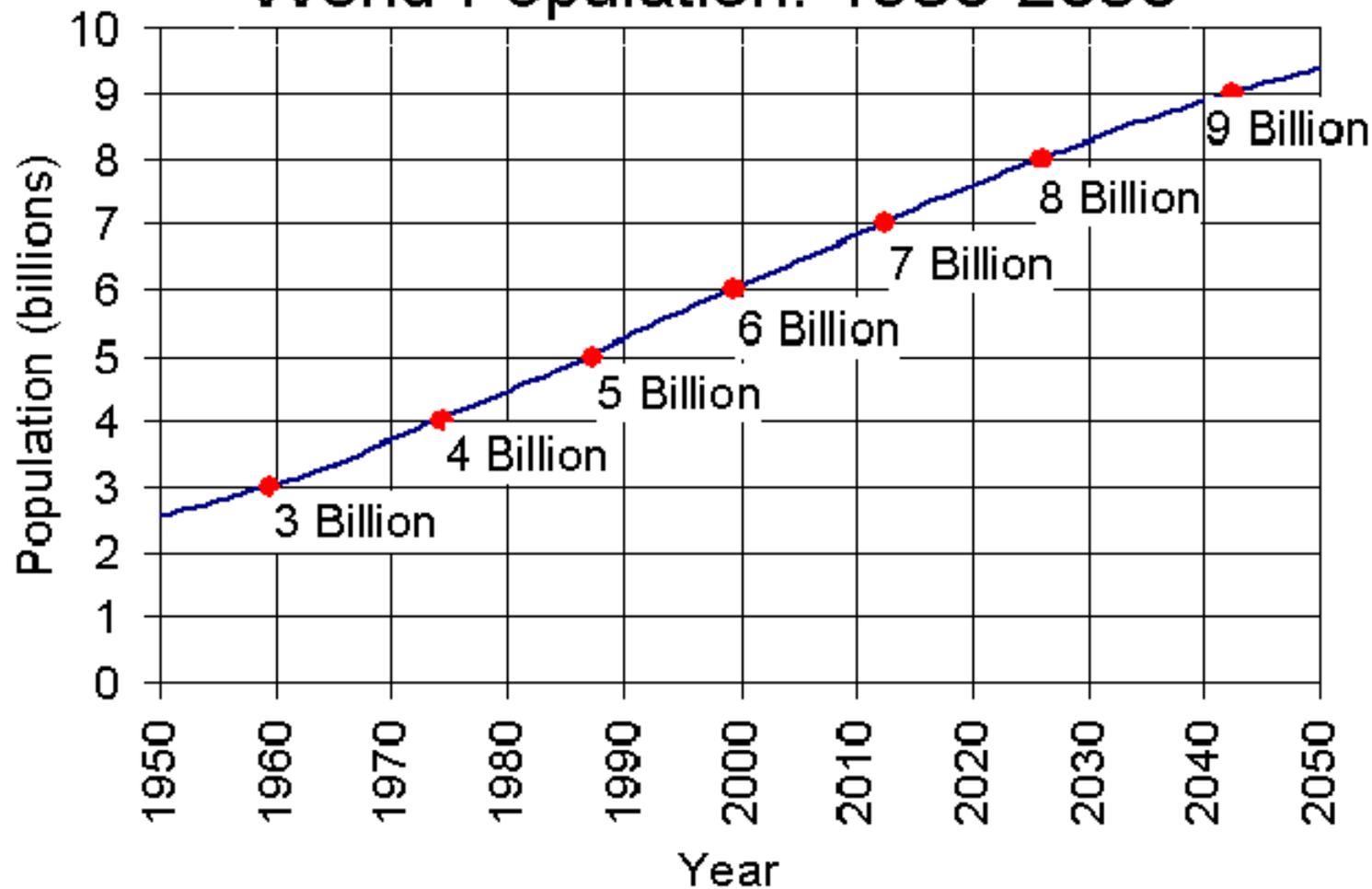


Figure by MIT OpenCourseWare.

World Population: 1950-2050

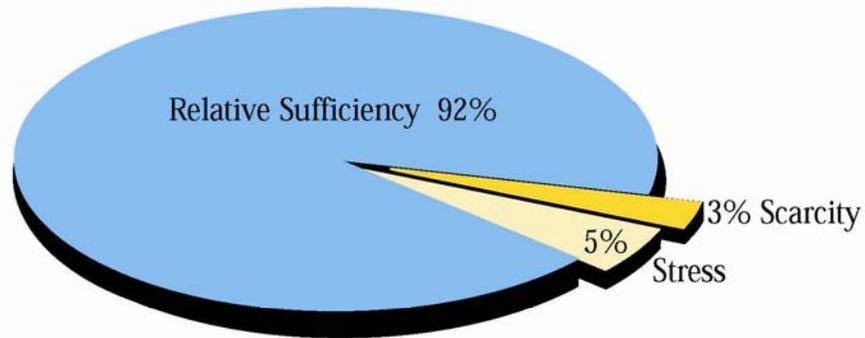


Source: U.S. Census Bureau, International Data Base, July 2007 version.

WORLD POPULATION IN FRESHWATER SCARCITY, STRESS AND RELATIVE SUFFICIENCY IN 1995 AND 2050

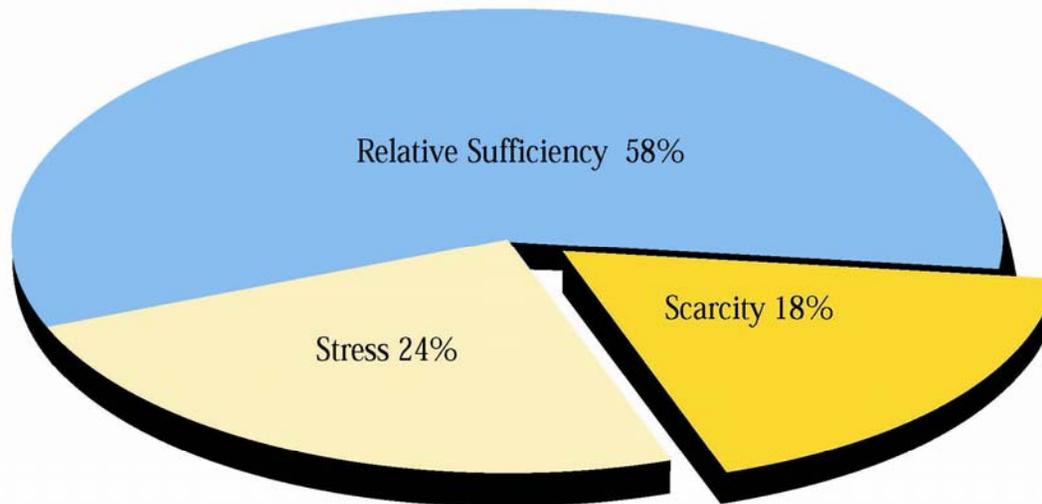
1995—Total Population: 5.7 billion

Medium Population Projection



2050—Total Population: 9.4 billion

Medium Population Projection



Water stress means that the annual water supply is below 1700 m³ per person.

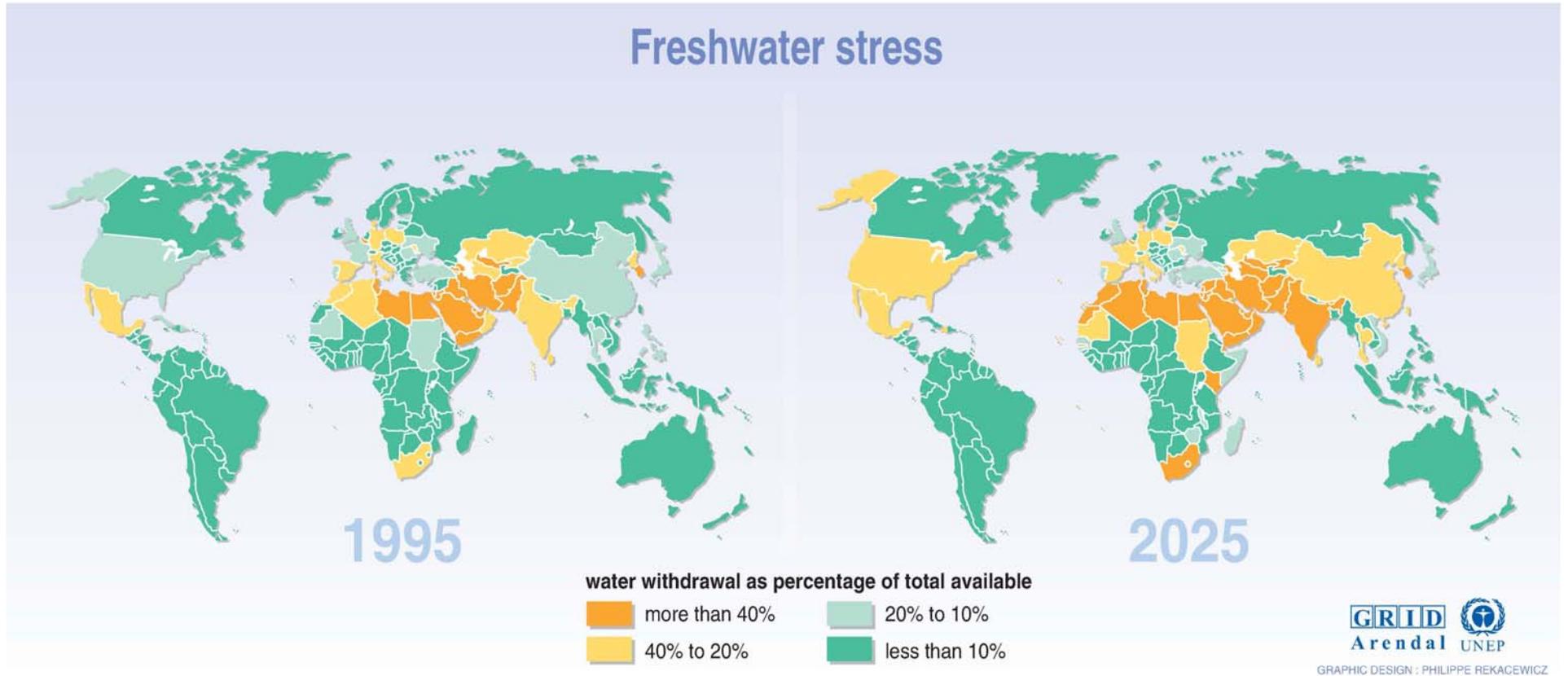
Water scarcity means that the annual water supply is below 1000 m³ per person.

*Note: The sizes of the pies are proportional to world population in both years.
Chart: Population Action International Data Source: UN Population Division*

Source: Gardener-Outlaw & Engelman

Source: Gardner-Outlaw, Tom, and Robert Engelman. "Sustaining Water, Easing Scarcity: A Second Update." *Population Action International*, May 1997. ([PDF](#))

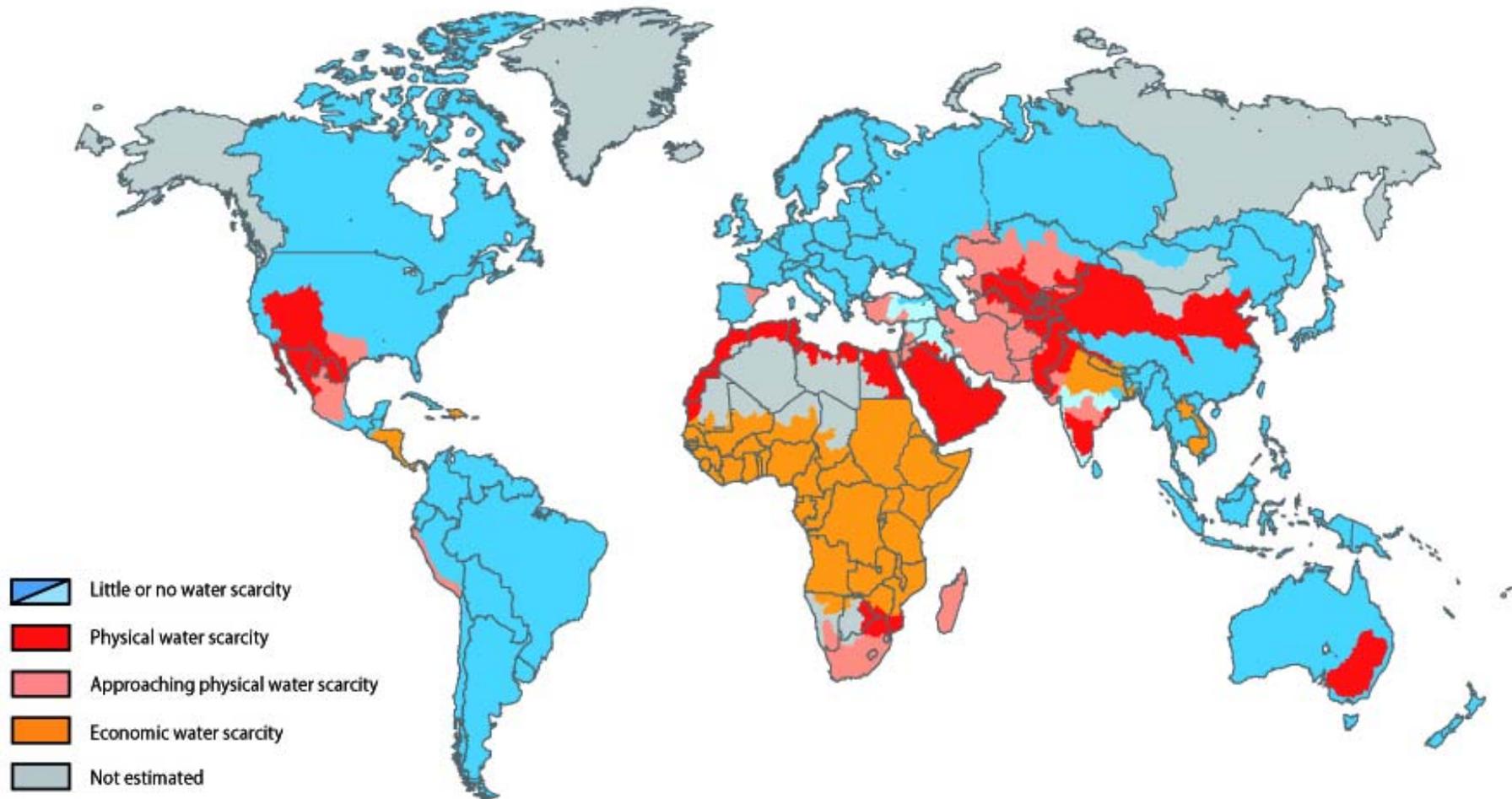
Freshwater stress



Source: Global environment outlook 2000 (GEO), UNEP, Earthscan, London, 1999.

Image by Philippe Rekacewicz for UNEP/GRID-Arendal. "[Freshwater Stress](#)." *UNEP/GRID-Arendal Maps and Graphics Library*. UNEP/GRID-Arendal, 2000. Accessed September 25, 2009.

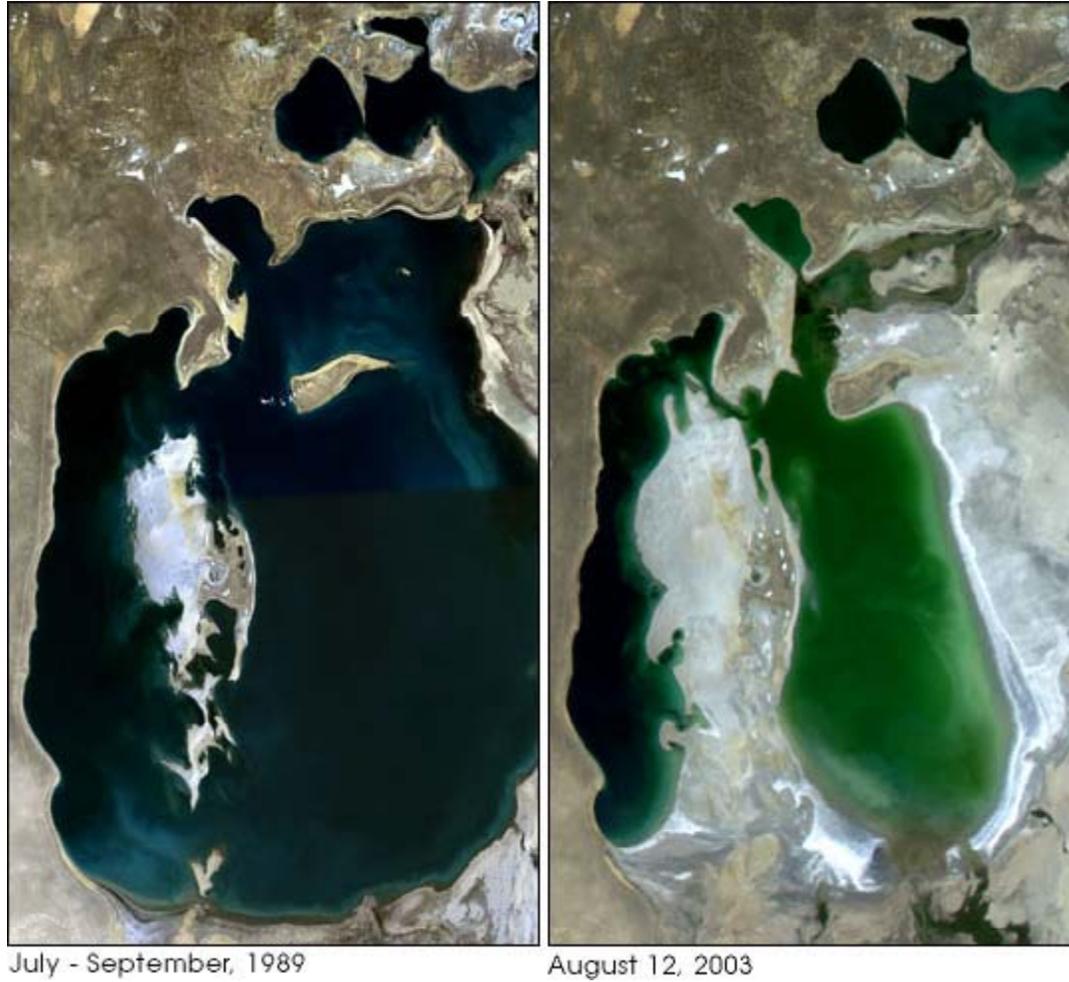
Areas of physical and economic water scarcity



Source: IMWI report, Insights from the Comprehensive Assessment of Water Management in Agriculture, 2006 / p8

UNEP/GRID-Arendal. "Areas of Physical and Economic Water Scarcity."

UNEP/GRID-Arendal Map and Graphics Library. UNEP/GRID-Arendal, 2008. Accessed September 25, 2009.



Aral Sea – water diverted for agriculture

Image from NASA Earth Observatory.

Source: infranetlab.org

World Insolation (kWh/m²-day)

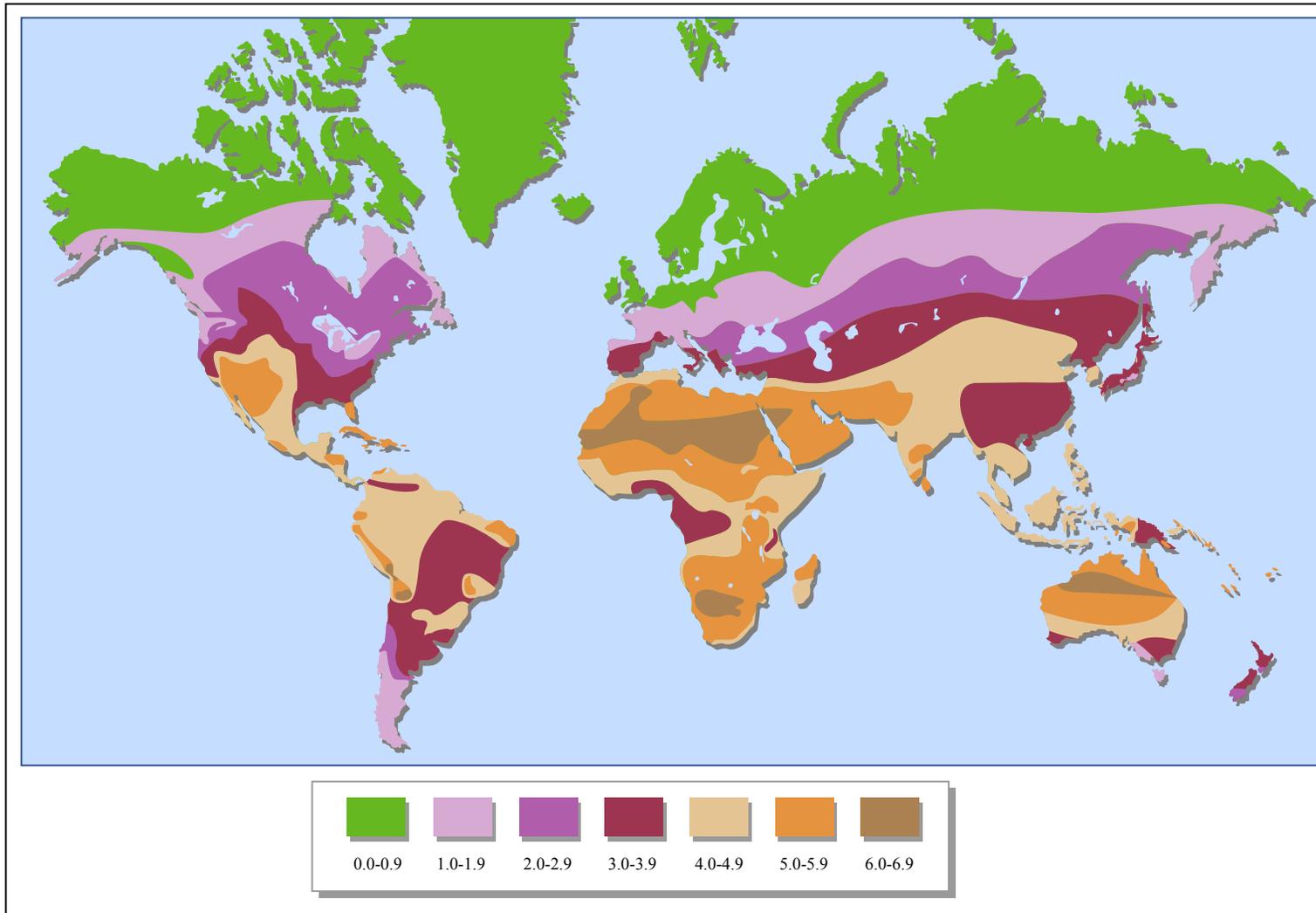


Figure by MIT OpenCourseWare.

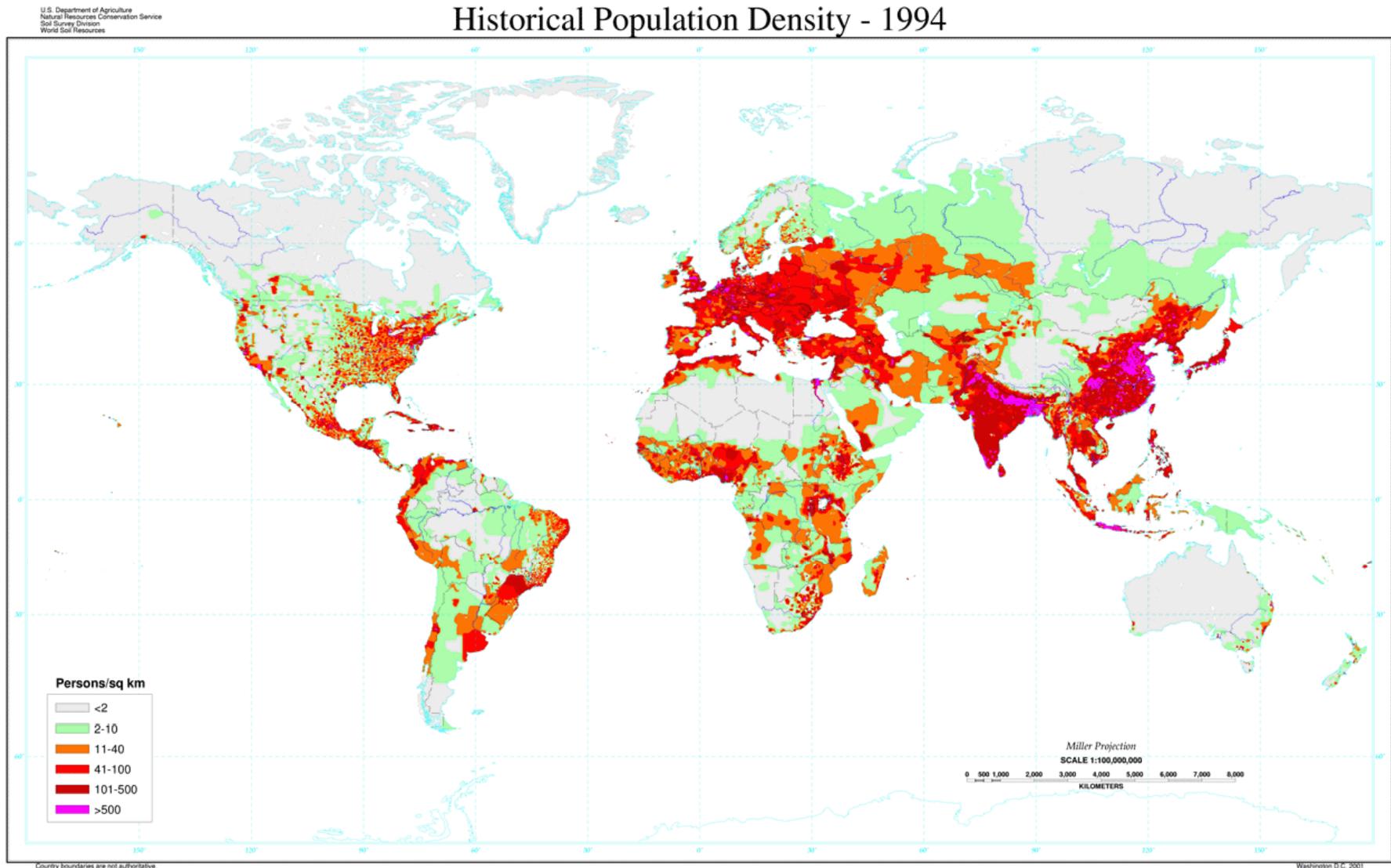
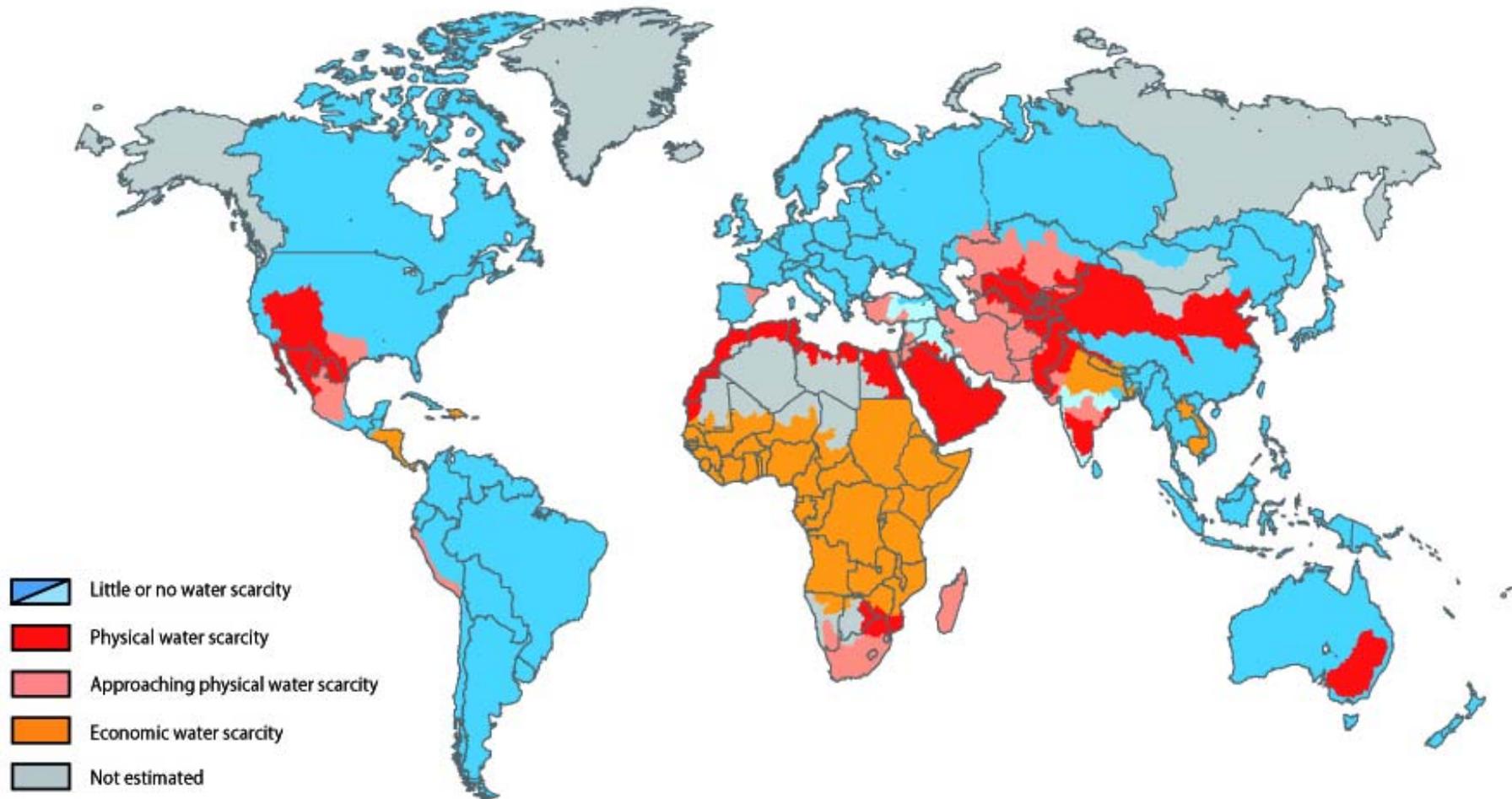


Image from Tobler, W., et al. "The Global Demography Project."
TR-95-6. Santa Barbara, CA: National Center for Geographic Information Analysis, 1995.
Image is in the public domain.

Areas of physical and economic water scarcity



Source: IMWI report, Insights from the Comprehensive Assessment of Water Management in Agriculture, 2006 / p8



Per capita water consumption (m³/y)

■ Worldwide average	800
■ Nigeria	50
■ China	300
■ Mexico	800
■ Italy	1000
■ USA	2000
■ World desalting capacity	2

laist.com

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Please see: <http://www.flickr.com/photos/peggyarcher/975676140/in/set-72157601398334771/>

Cleaning a sidewalk in Long Beach, CA

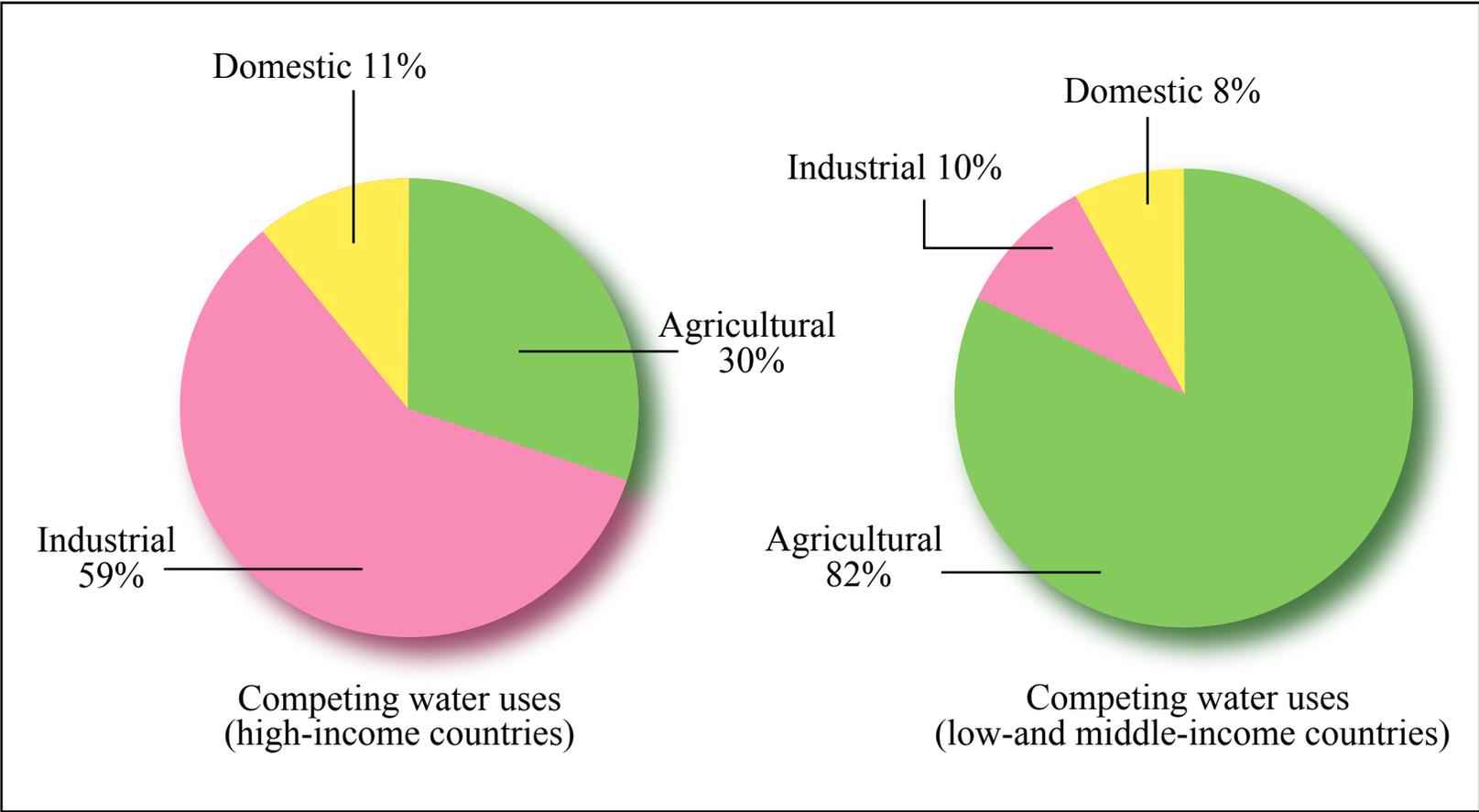
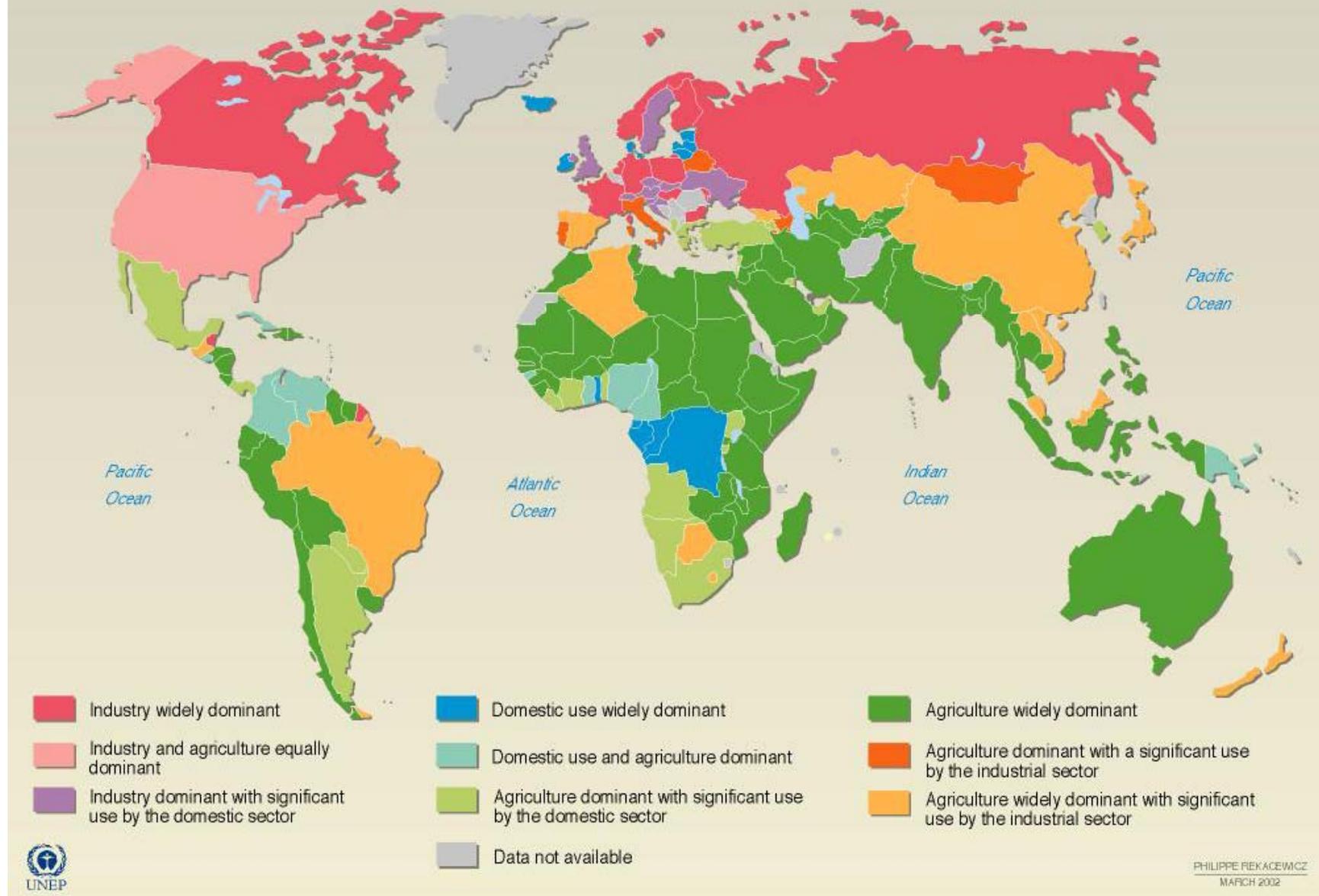


Figure by MIT OpenCourseWare.

Global Freshwater Withdrawal

Country Profiles Based on Agricultural, Industrial and Domestic Use

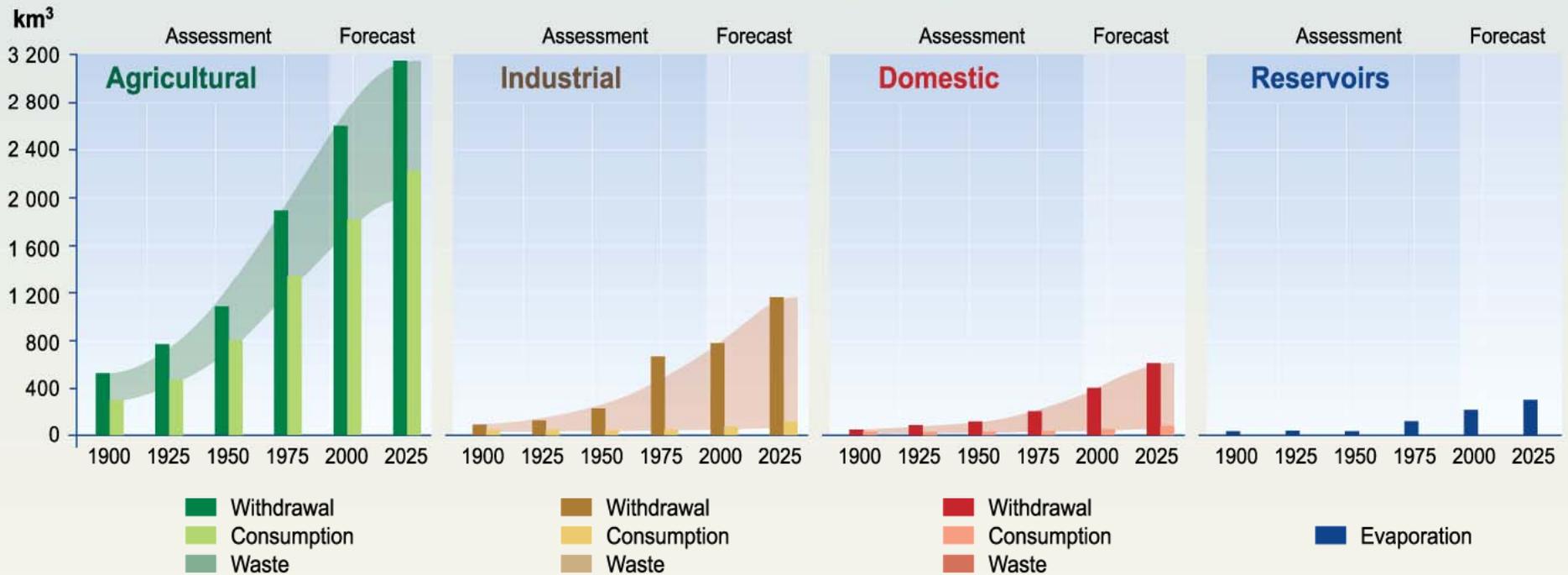


Source: Based on data from Table FW1 in *World Resources 2000-2001, People and Ecosystems: The Fraying Web of Life*, World Resources Institute (WRI), Washington DC, 2000.

Image by Philippe Rekacewicz for UNEP/GRID-Arendal. "[Global Freshwater Withdrawal](#)."

UNEP/GRID-Arendal Maps and Graphics Library. UNEP/GRID-Arendal, 2002. Accessed September 25, 2009.

Evolution of Global Water Use Withdrawal and Consumption by Sector



Note: Domestic water consumption in developed countries (500-800 litres per person per day) is about six times greater than in developing countries (60-150 litres per person per day).

PHILIPPE REKACEWICZ
FEBRUARY 2002

Source: Igor A. Shiklomanov, State Hydrological Institute (SHI, St. Petersburg) and United Nations Educational, Scientific and Cultural Organisation (UNESCO, Paris), 1999.

Author's definition of "waste" is not clear; however, it is common for municipal water distribution systems to lose 20 to 40% of water by leakage.

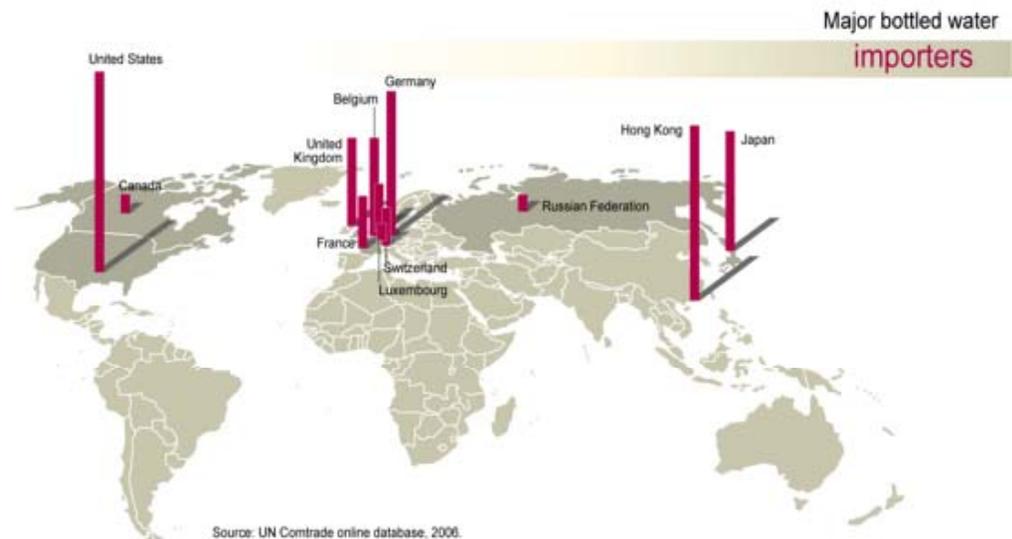
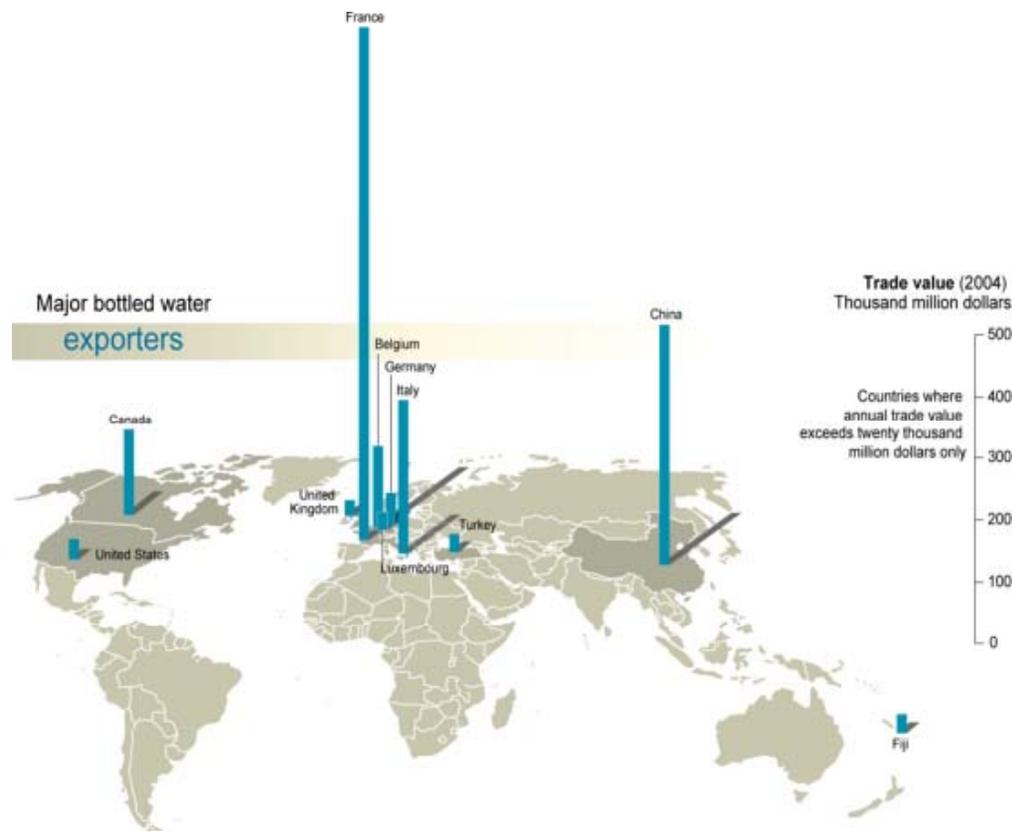
Image by Philippe Rekacewicz for UNEP/GRID-Arendal. "Trends and Forecasts in Water Use, by Sector." UNEP/GRID-Arendal Maps and Graphics Library. UNEP/GRID-Arendal, 2002. Accessed September 25, 2009.

Product	Unit	Equivalent water in cubic meters
Bovine (cattle)	Head	4,000
Sheep and goats	Head	500
Meat (bovine fresh)	Kilogram	15
Meat (sheep fresh)	Kilogram	10
Meat (poultry, fresh)	Kilogram	6
Cereals	Kilogram	1.5
Citrus fruit	Kilogram	1
Palm oil	Kilogram	2
Pulses, roots and tubers	Kilogram	1

Water requirement equivalent of main food production.

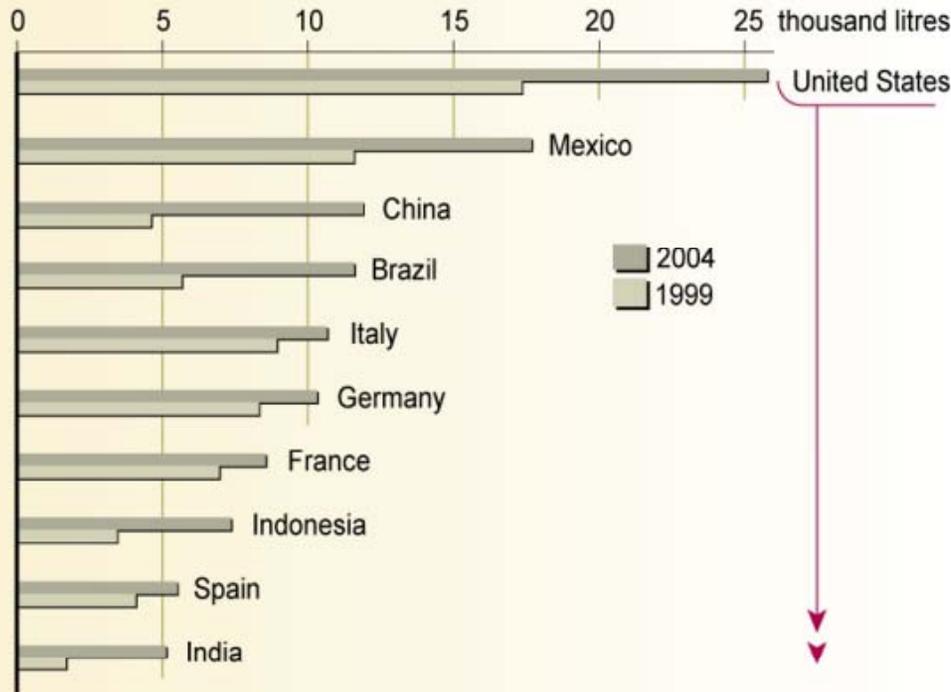
Figure by MIT OpenCourseWare.

Bottled water exportation...

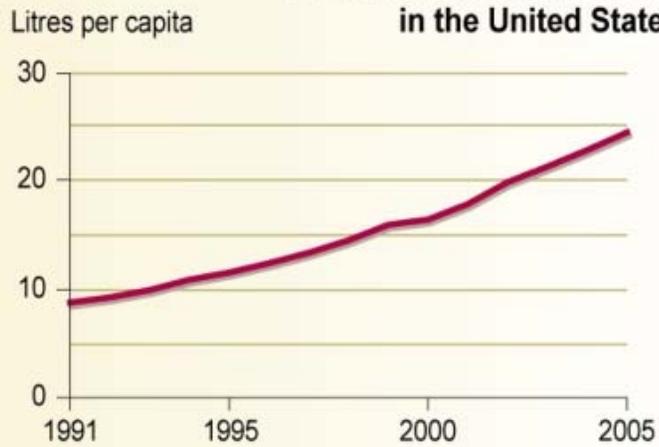


...and importation

Total bottled water consumption (leading consumers)



Bottled water consumption in the United States



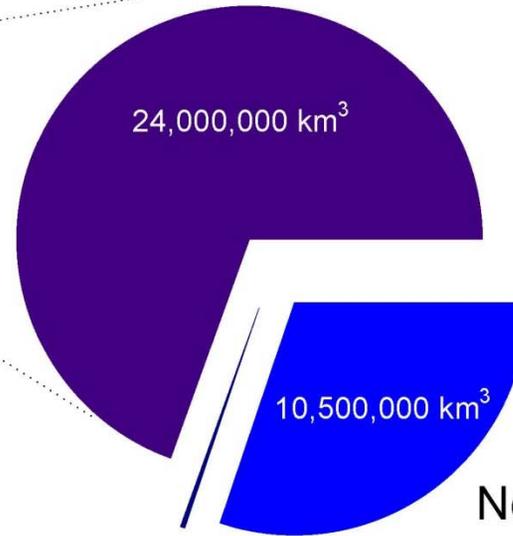
Many countries that
consume large
amounts of bottled
water have excellent
tap water...

Sources: International Bottled Water Association, 2005; Beverage Marketing Corporation, 2005.

Fresh Water - 2.5 %

Glaciers and Ice Caps - 69.5 %

Total = 1.4 billion km³



Nonrenewable - 30.2 %

(Ground water: lakes, etc.)

110,000 km³

Renewable - 0.3 %

(Precipitation over land)

Salt Water - 97.5 %

Approximately 23% of renewable water is appropriated for human uses, including agriculture. Accessible annual run-off is about 12,500 km³/y, of which about 54% is acquired for human use.

Courtesy of Sandia National Labs. Used with permission.

Source: Miller, 2003.

Water flows (km³/y)

Precipitation on land

120,000

- Evaporation on land
70,000
- River runoff and groundwater
recharge
50,000

Available river flow and recharge

12,000

- Withdrawal for human use
 - Agriculture 3,500
 - Industry 1,000
 - Domestic 500

WATER SUPPLY ALTERNATIVES

Water source	Energy requirement kWh/m ³ (kWh/kgallon)	Availability and constraints
Surface water or underground aquifer	0.1–1.5 (0.4–6)	<ul style="list-style-type: none"> • Quantity is limited
Desalination of brackish water	0.4–0.8 (1.5–3)	<ul style="list-style-type: none"> • Localized and limited availability
Desalination of seawater	3–4 (11–15)	<ul style="list-style-type: none"> • Available at coastal locations only • Environmental constraints
Advanced wastewater reclamation	0.8–1 (3–4)	<ul style="list-style-type: none"> • Available at centers of use • Public perception problem

Figures from Wilf, M., and M. Balaban. *Membrane Desalination and Membrane Filtration*. L'Aquila, Italy: European Desalination Society, 2007. Used with permission.

Source: Sommariva, 2007

Water Quality Characteristics

Ref: Reynolds &
Richards

Biological
Characteristics
microorganisms

Physical
Characteristics
*taste, odor,
color,...*

Chemical
Characteristics
*natural or
manmade*



Images from Wikimedia Commons,
<http://commons.wikimedia.org>

Biological

- Bacteria
- Viruses
- Protozoa
- Coliform bacteria (*indicate human waste*)
- Helminths
- Fungi, algae

Physical

- Total solids (*dissolved and suspended*)
- Turbidity
- Color (*apparent and true*)
- Taste & odor (*organic compounds in surface water; dissolved gases in ground water*)
- Temperature

Chemical

- pH
- Anions & cations (*dissolved solids*)
- Alkalinity (HCO_3^- , CO_3^{2-} , OH^- system)
- Hardness (Ca^{2+} , Mg^{2+})
- Dissolved gases (O_2 , CO_2 , H_2S , NH_3 , N_2 , CH_4 ...)
- Priority pollutants (*organic and inorganic*)

Microbial contamination is the #1 concern for water

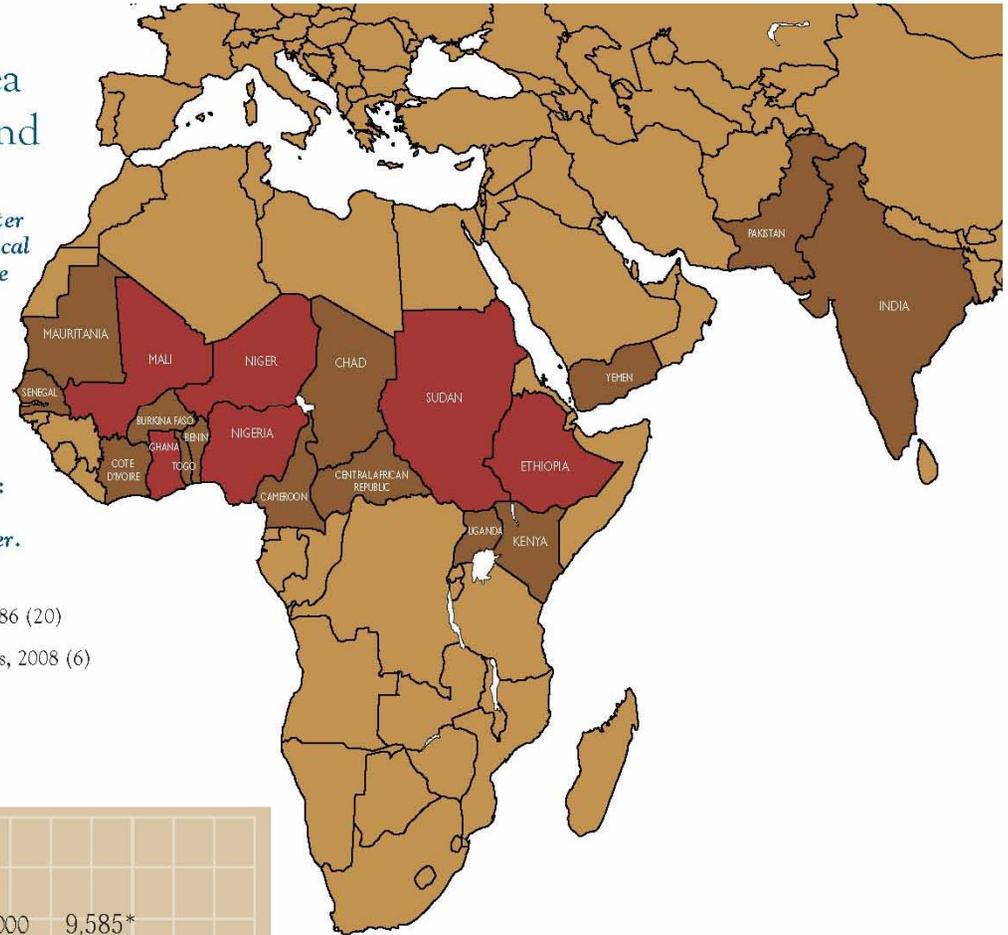
- **Protozoans**
 - Amoeba, cryptosporidium, giardia, algae,...
- **Bacteria**
 - Salmonella, typhus, cholera, shigella, ...
- **Viruses**
 - Polio, hepatitis A, meningitis, encephalitis,...
- **Helminths**
 - Guinea worm, hookworm, roundworm,...
- Principal transmission is by **human waste**
- Principal purification technique is **chlorination**, especially for bacteria.

Some water borne diseases can be eradicated...

...3,500,000 cases of Guinea worm in 1986....

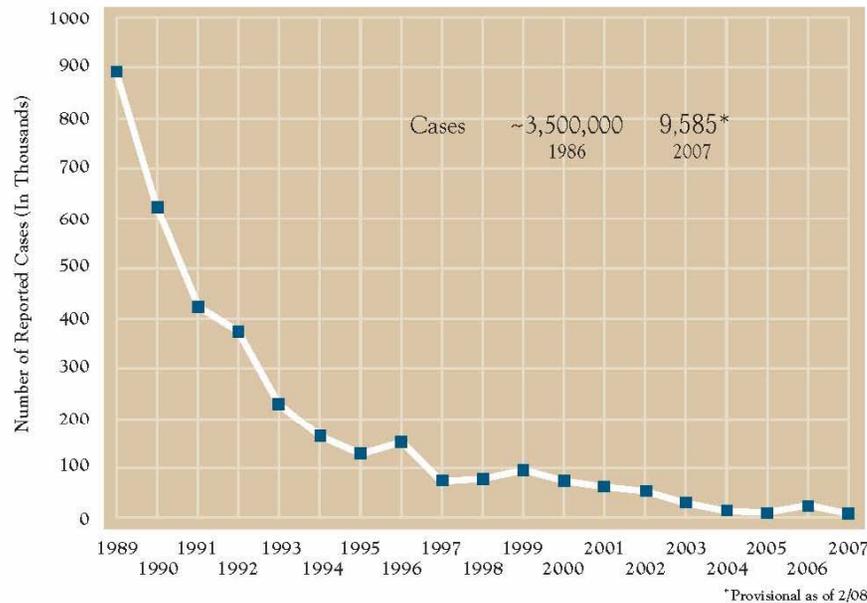
Where Guinea Worm Is Found

When The Carter Center began to provide technical and financial assistance to national eradication programs in 1986, Guinea worm disease was found in 20 countries in Africa and Asia. Today the disease remains in six countries, all in Africa: Sudan, Ghana, Mali, Ethiopia, Nigeria, Niger.



- Endemic countries, 1986 (20)
- Still endemic countries, 2008 (6)

Number of Reported Cases of Guinea Worm Disease by Year, 1989-2007



...<5000 cases in 2008...

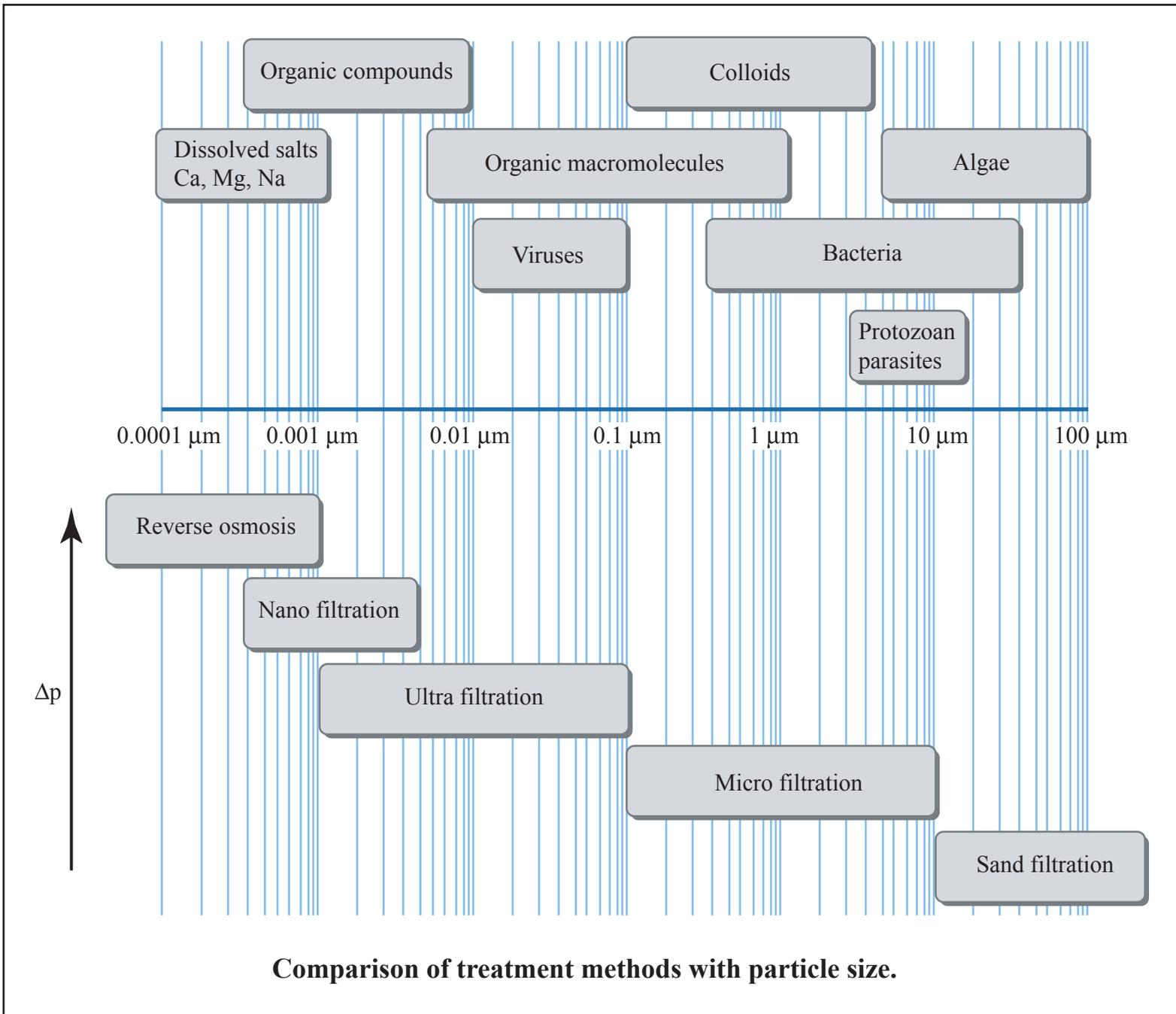
...80% in Sudan.

Physical characteristics

- **Suspended solids** include silt, clay, algae, colloids, bacteria...remove by settling, filtration, or flocculation
- **Turbidity** interferes with passage of light, usually as the result of colloidal material
- **Color** is due to dissolved (true color) or colloidal (apparent color) material...iron, manganese, clay,...
- **Taste/odor** ...typically treated by aeration (to release dissolved gas from ground water) or activated carbon (to remove organics from surface water)

EPA Primary Standards for ~130 chemicals

- **Toxic metals** – Arsenic, lead, mercury, cadmium, chromium,...
- **Organic compounds** – insecticides, herbicides, PCBs, petrochemicals, PAH, benzene, halogenated hydrocarbons,...very long list
- **Nitrate or nitrite** – fertilizer byproduct
- **Fluorine** – damages teeth and bones at high concentrations
- **Radionuclides** – mainly natural alpha emitters...
- **Secondary standards** for taste, odor, appearance: Cl^- , SO_4^{2-} , pH, color, odor, iron, manganese, copper, zinc, foaming agents...



Comparison of treatment methods with particle size.

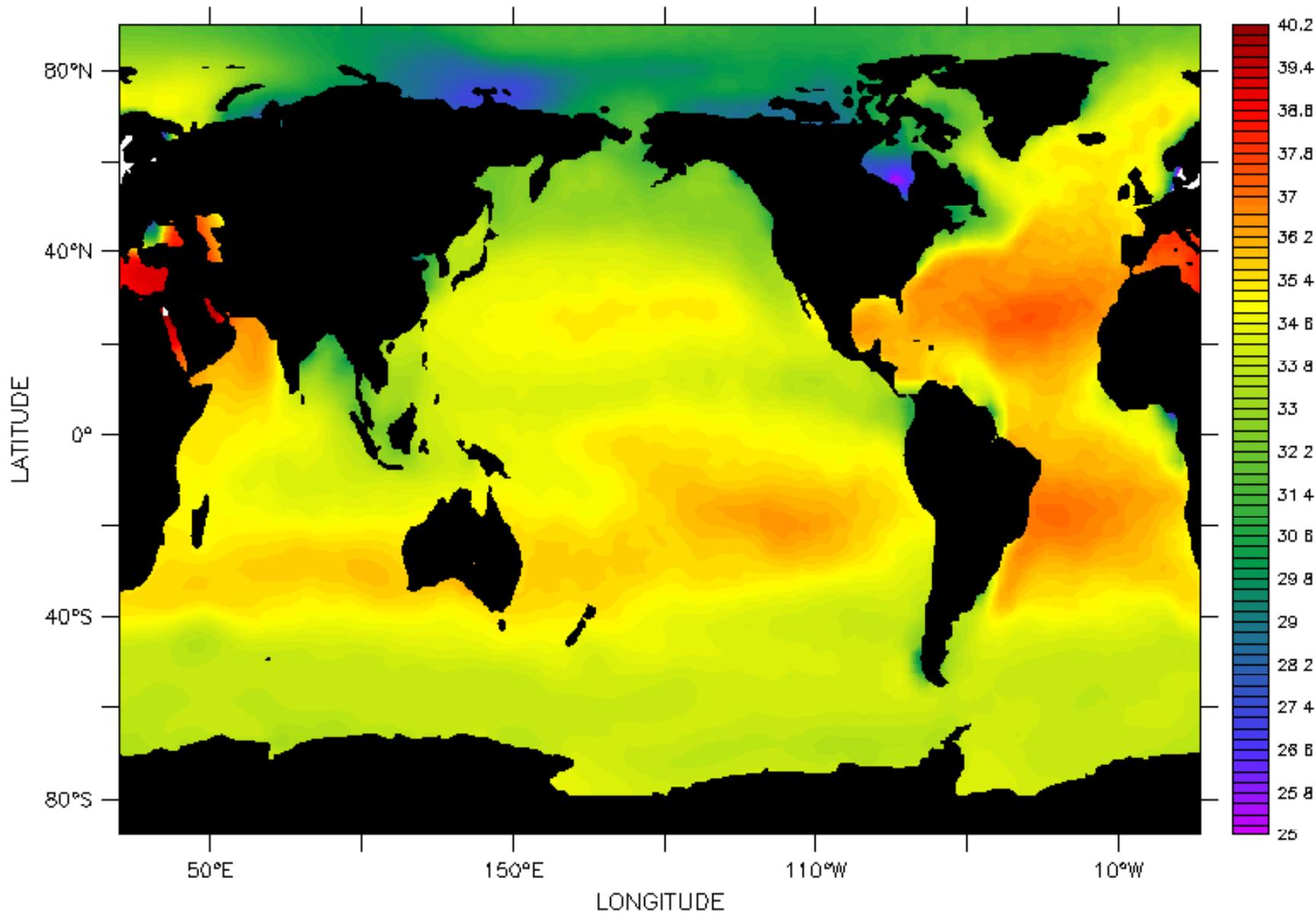
Why not drink seawater?

- Seawater is usually about 3.5% by weight dissolved salts (35000 ppm)
- Human blood has the [osmotic] equivalent of about 0.9% salinity (~9000 ppm)
- **Ingesting too much salt** leads to excretion of water and dehydration. Seawater contains about 3x more dissolved salts than human blood. You can't drink it. It also tastes bad.

Characterization of Water Salinity

	Minimum Salinity, TDS [ppm or mg/kg]	Maximum Salinity, TDS [ppm or mg/kg]
Seawater	15,000	50,000
Brackish water	1,500	15,000
River water (brackish)	500	1,500
Pure water	0	500

Annual Mean Sea Surface Salinity



Salinity Data from Levitus (1982)

Time: Sep. 4, 1997

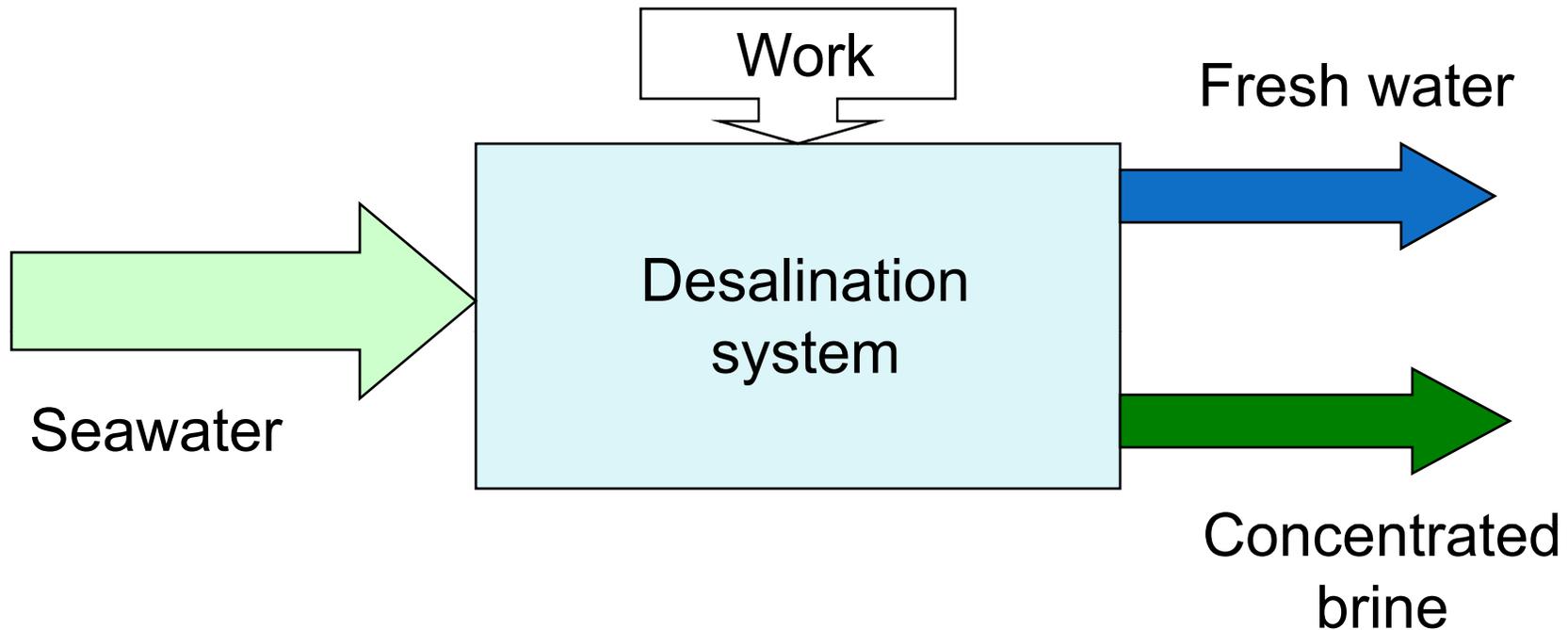
Substance (amounts in mg/kg)	Standard Seawater	Cambridge City Water	Massachusetts Water Resources Authority	Poland Springs Bottled Water	Maximum Allowable
Sodium, Na ⁺	10781	79	30	2.6-5.6	aesthetics: 200
Magnesium, Mg ²⁺	1284	5	0.8	0.7-1.9	-
Calcium, Ca ²⁺	412	25	4.5	3.5-9.5	-
Potassium, K ⁺	399	nr ^[1]	0.9	0.74-0.88	-
Strontium, Sr ⁺	13	nr	nr	nr	-
Chloride, Cl ⁻	19353	140	21	1.5-6.6	250
Sulfate, SO ₄ ²⁻	2712	27	8	0.87-5.9	250
Bicarbonate, HCO ₃ ⁻	126	nr	nr	13-28	-
Bromide, Br ⁻	67	nr	0.016	not detected	-
Boric Acid, B(OH) ₃	26	nr	nr	nr	-
Fluoride, F ⁻	1.3	1	1	0.0-0.27	2-4
Water	965000	-	-	-	-
Total dissolved solids	35200	320	110	33-57	500
Nitrate, NO ₃		0.46	0.11	0.12-0.42	10
Retail Cost, US\$/m ³	free?	1.05	1.18	~300 to 3000	-

^[1] nr = not reported.

Substance (amounts in mg/kg)	Standard Seawater	High Salinity Brackish Water	Low Salinity Brackish Water	Massachusetts Water Resources Authority
Sodium, Na ⁺	10781	1837	90	30
Magnesium, Mg ²⁺	1284	130	11.7	0.8
Calcium, Ca ²⁺	412	105	96	4.5
Potassium, K ⁺	399	85	6.5	0.9
Strontium, Sr ⁺	13	nr	nr	nr
Chloride, Cl ⁻	19353	2970	191	21
Sulfate, SO ₄ ²⁻	2712	479	159	8
Bicarbonate, HCO ₃ ⁻	126	250	72.6	nr
Bromide, Br ⁻	67	nr	nr	0.016
Boric Acid, B(OH) ₃	26	nr	nr	nr
Fluoride, F ⁻	1.3	1.4	0.2	1
SiO ₂	2	17	24	3.3
Nitrate, NO ₃	nr	5.0	nr	0.11
Total dissolved solids	35200	5881	647	110

Brackish compositions are representative;
from M. Wilf, 2007

Seawater purification



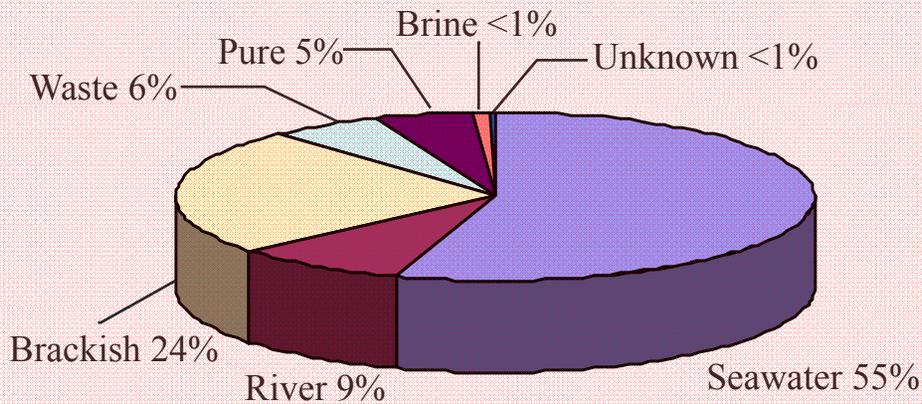
Ideally, this requires 2.5 to 7 kJ per kg fresh water produced.
Practically, it takes an order of magnitude more energy.

Principal desalination techniques

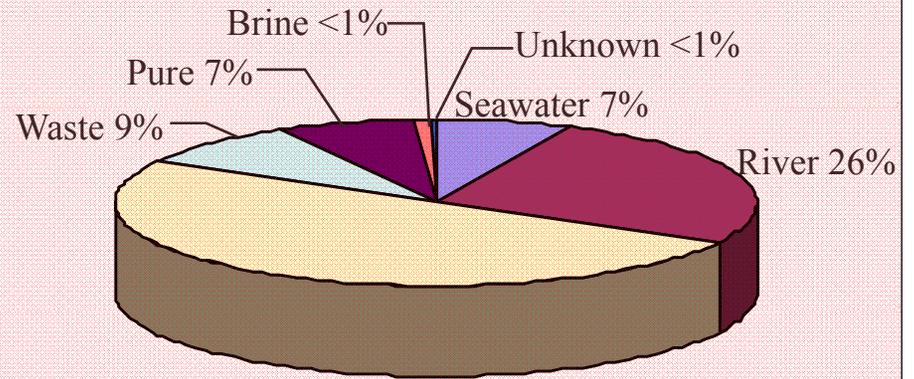
- **Membrane techniques**
 - Reverse osmosis (SWRO or BWRO)
 - Electrodialysis (ED)
 - Capacitative deionization (CDI)
 - Nanofiltration (NF)
- **Distillation techniques**
 - Multistage flash evaporation (MSF)
 - Multieffect distillation (MED or MEE)
 - Vapor compression distillation
 - Solar thermal distillation (concentrating or not)
- **Related methods**
 - Deionization
 - Water softening

Installed desalination capacity

S
O
U
R
C
E

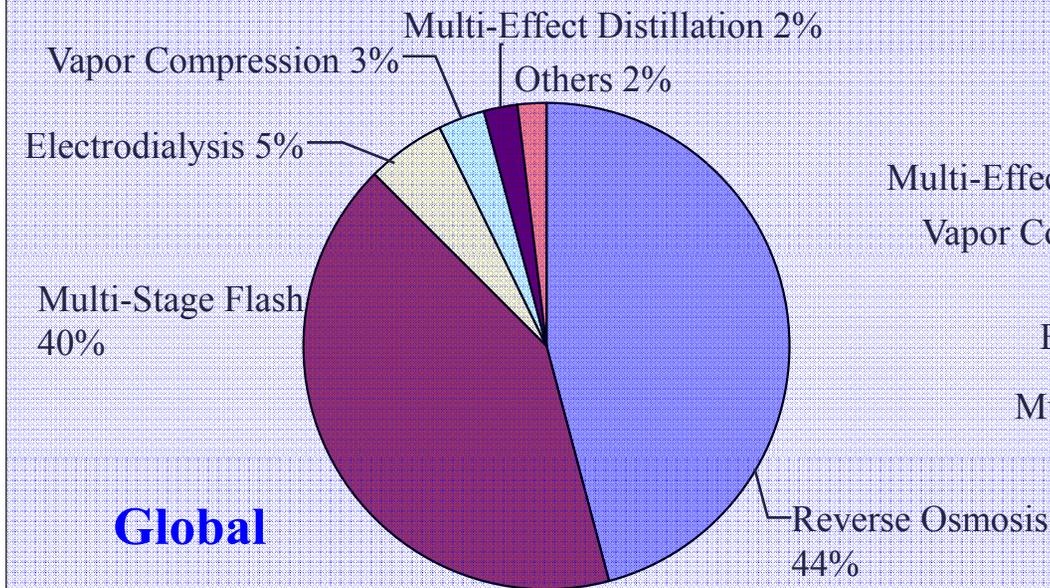


Global

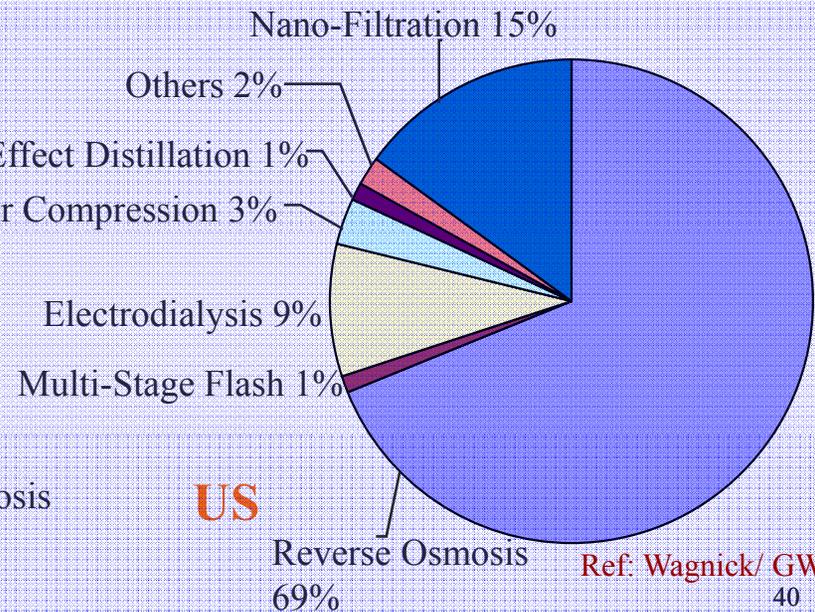


US

Brackish 51% Ref: Wagnick/ GWI 2005



Global



US

Ref: Wagnick/ GWI 2005

P
R
O
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S

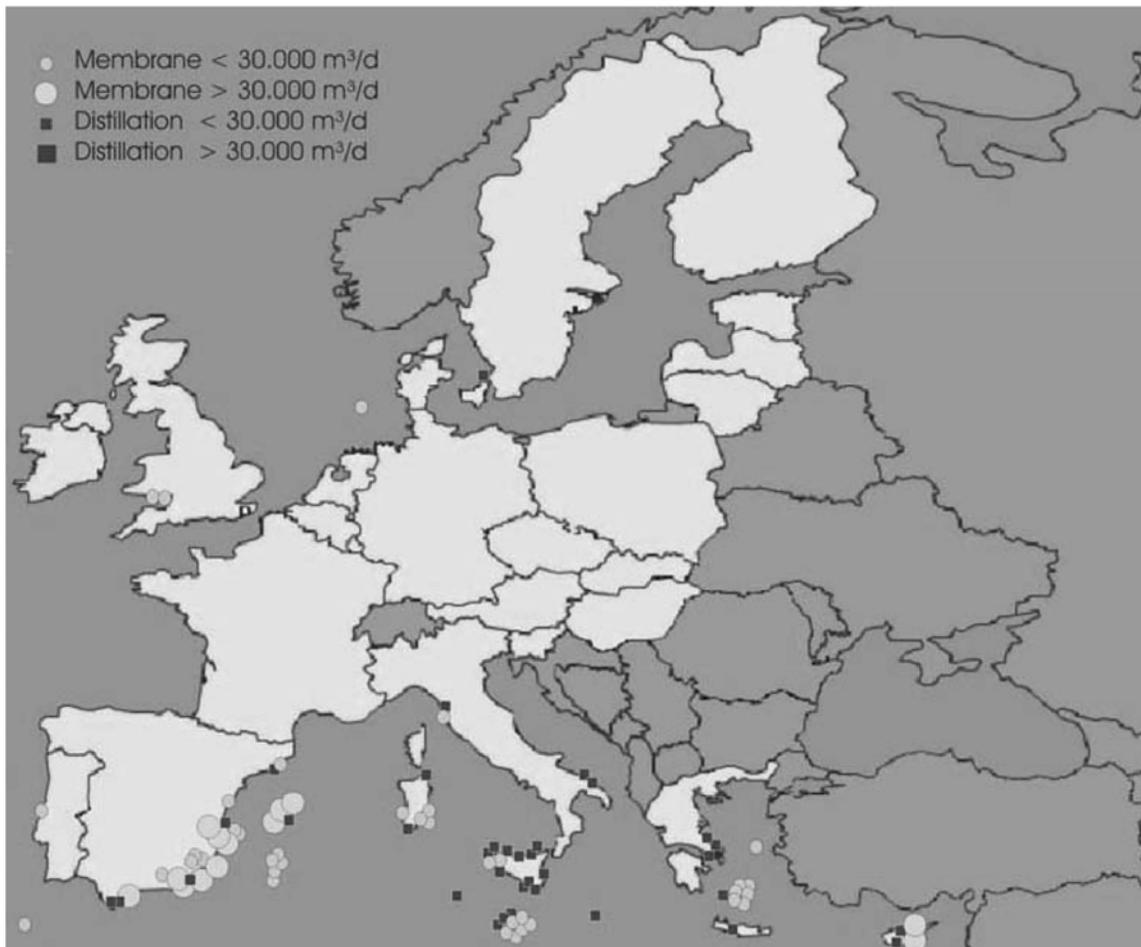


Fig. 5. Seawater desalination plants constructed from 1985 until today. (Only plants with a capacity of at least 700 m³/d were considered). Modified from [2].

Courtesy of Elsevier, Inc., <http://www.sciencedirect.com>. Used with permission.

Source: Fritzmann et al., *Desalination*, 2007

1) Multi Stage Flash - MSF

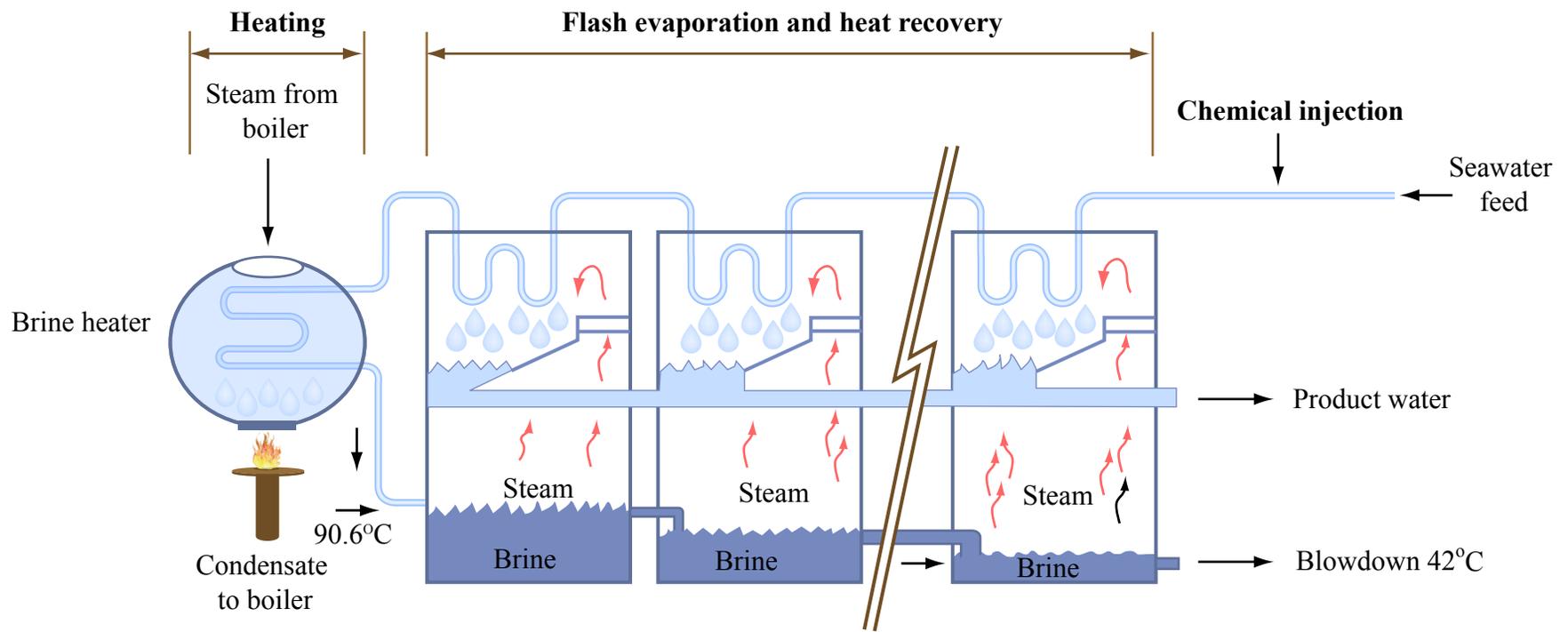
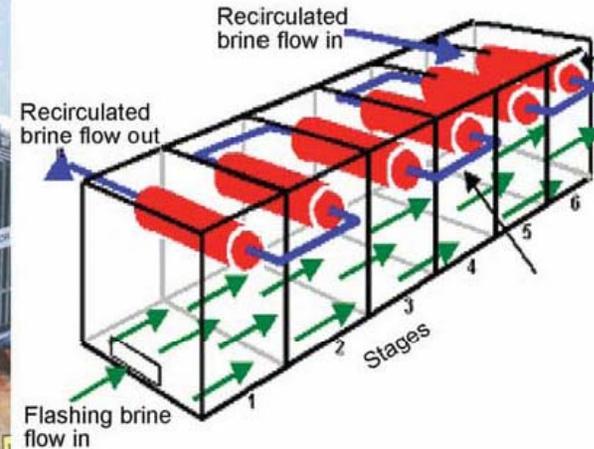


Figure by MIT OpenCourseWare.

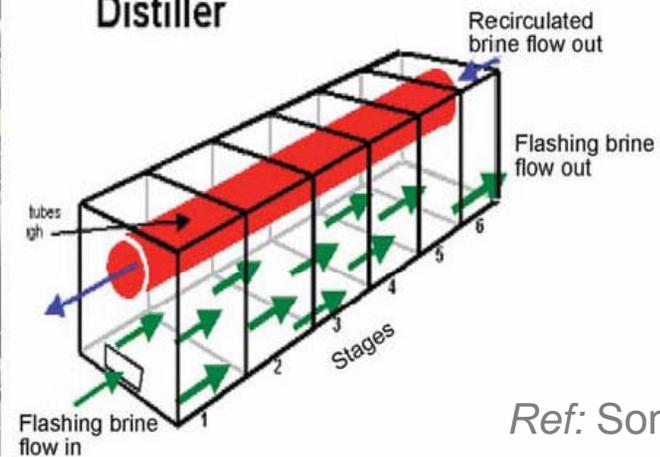
Multi stage flash



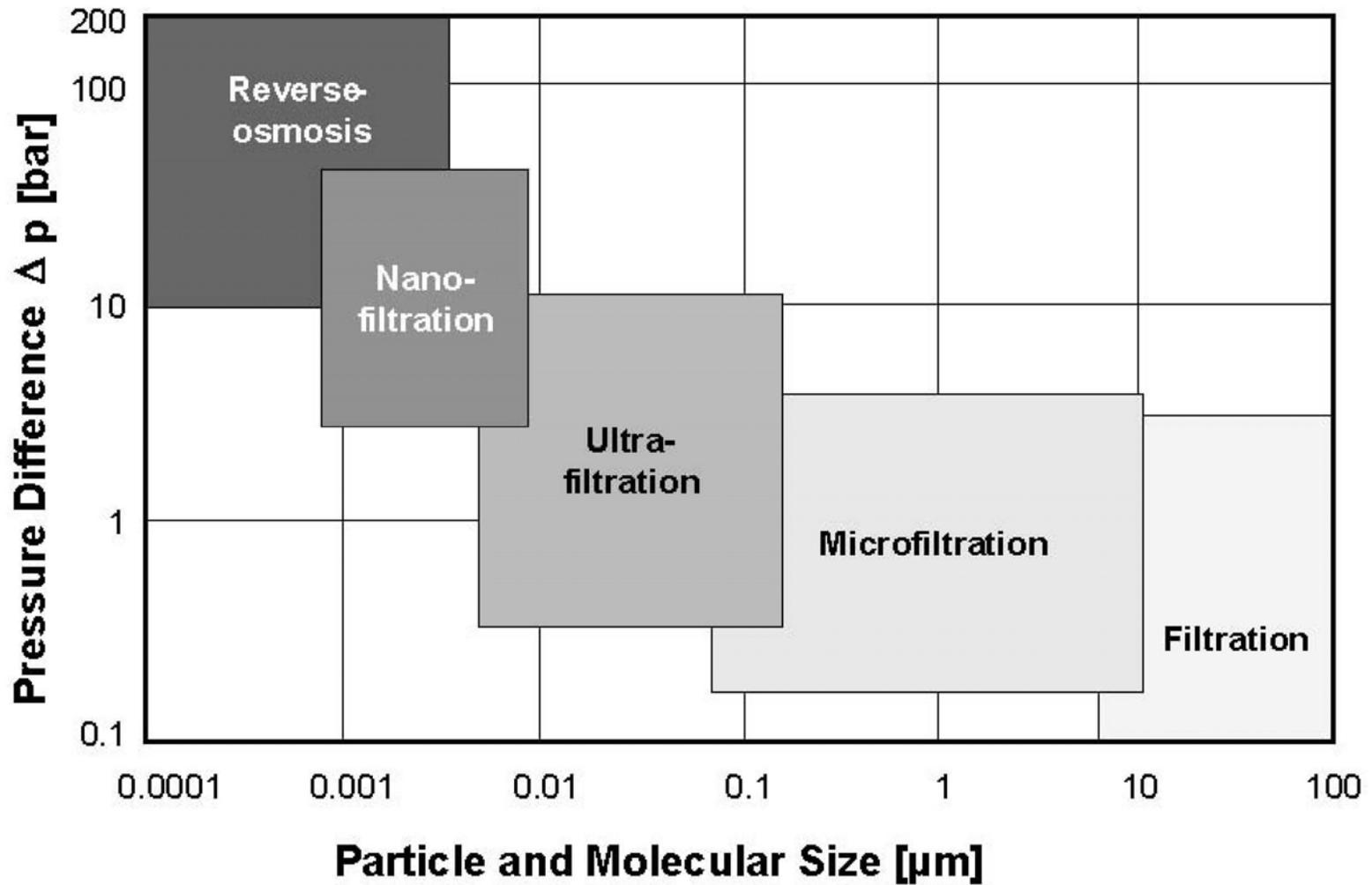
Cross Tube



Long Tube Distiller



Ref: Sommariva, 2007



Courtesy of Elsevier, Inc., <http://www.sciencedirect.com>. Used with permission.

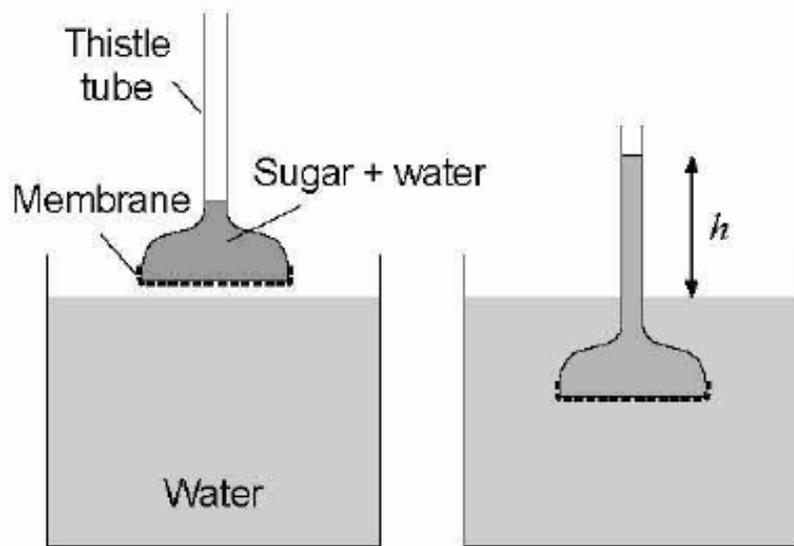


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Please see:

<http://library.tedankara.k12.tr/chemistry/vol1/balances/trans76.jpg>

Courtesy of Dennis Freeman, Martha Gray, and Alexander Aranyosi.

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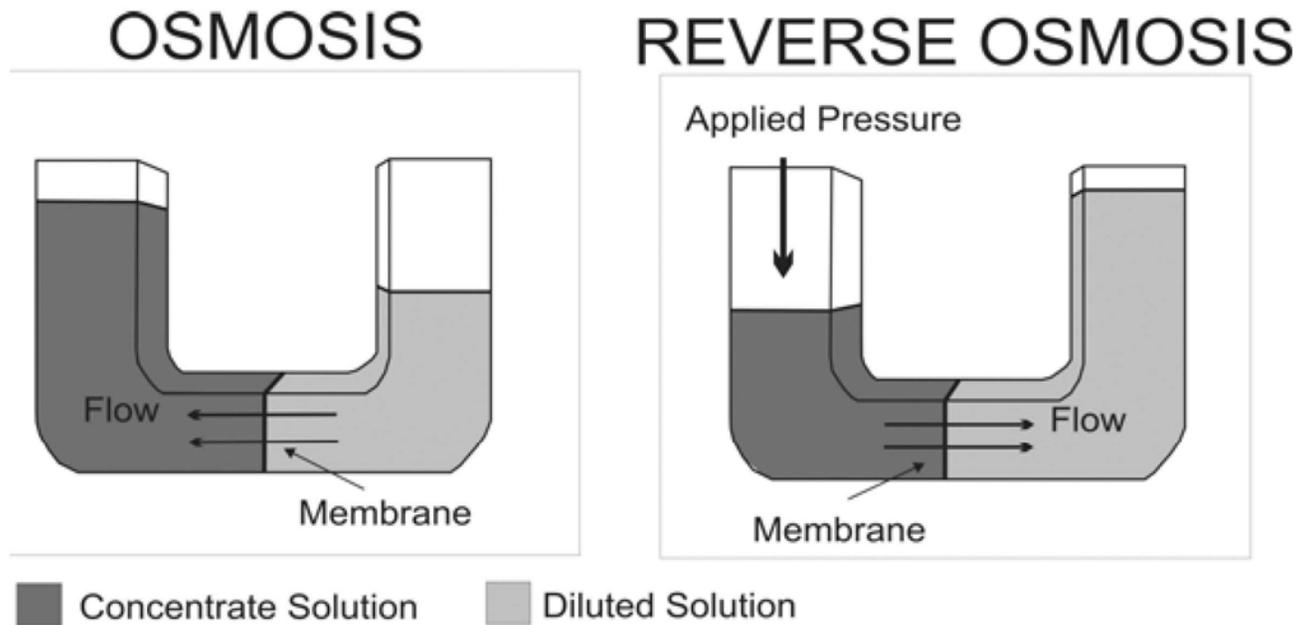
<http://images.encarta.msn.com/xrefmedia/aencmed/targets/illus/ilt/0007ff2f.gif>

Osmosis. to achieve equilibrium, water will diffuse through a semi-permeable membrane into a solution. This occurs until sufficient hydrostatic pressure develops to offset the *osmotic pressure*

Sources:

<http://encarta.msn.com> (left)

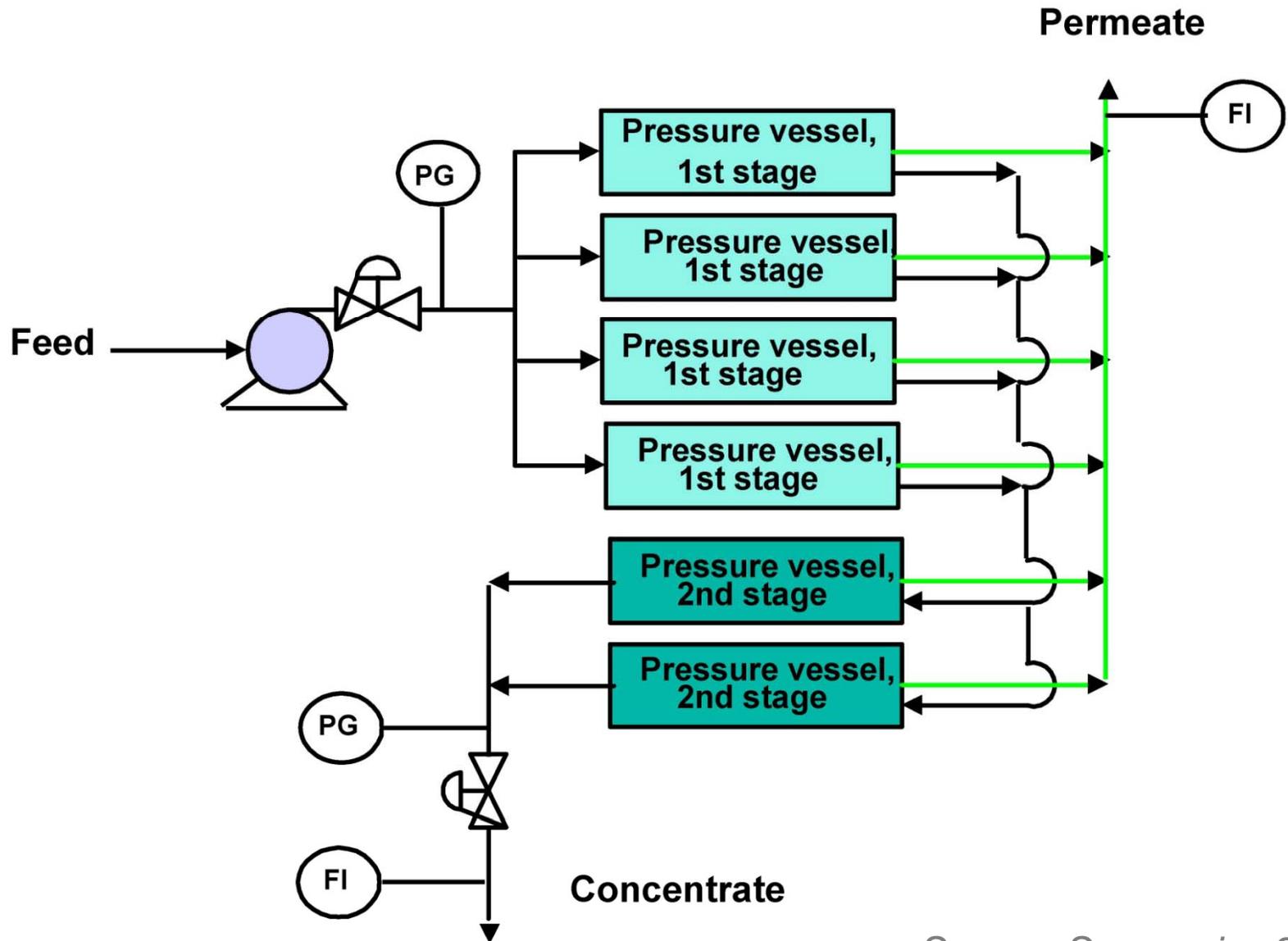
<http://library.tedankara.k12.tr/>



Courtesy of Elsevier, Inc., <http://www.sciencedirect.com>. Used with permission.

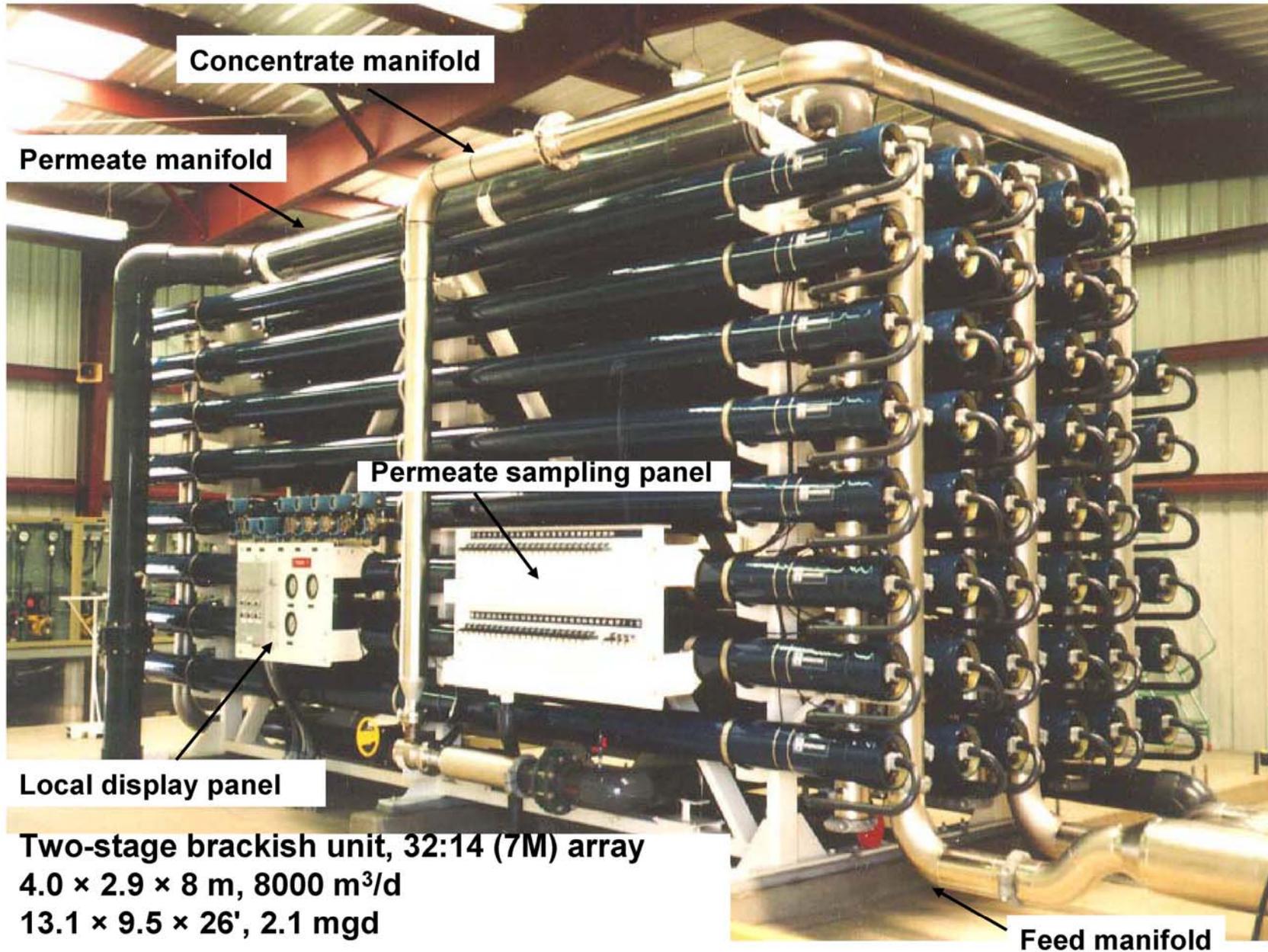
If pressure is applied to the solution, the direction of osmotic flow can be reversed. In this way solvent can be driven through the membrane, purifying it.

Two stage RO system



Source: Sommariva, 2007

Figures from Wilf, M., and M. Balaban. *Membrane Desalination and Membrane Filtration*. L'Aquila, Italy: European Desalination Society, 2007. Used with permission.



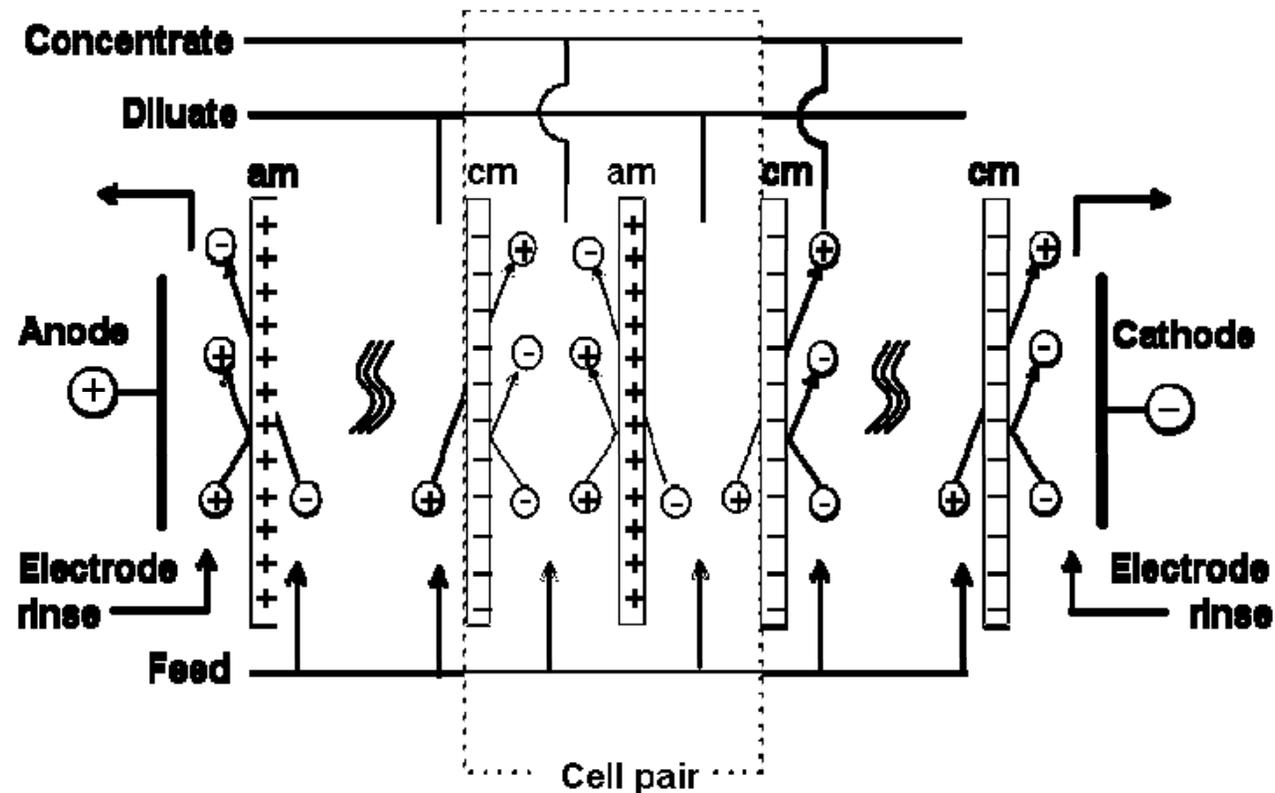
First stage has 32 pressure vessels; second stage has 14 (Wilf & Balaban ,2007)

Figures from Wilf, M., and M. Balaban. *Membrane Desalination and Membrane Filtration*. L'Aquila, Italy: European Desalination Society, 2007. Used with permission.

CONVENTIONAL ELECTRODIALYSIS



The process principle



Courtesy of Heiner Strathmann. Used with permission.

ions are removed from a feed solution and concentrated in alternating cells

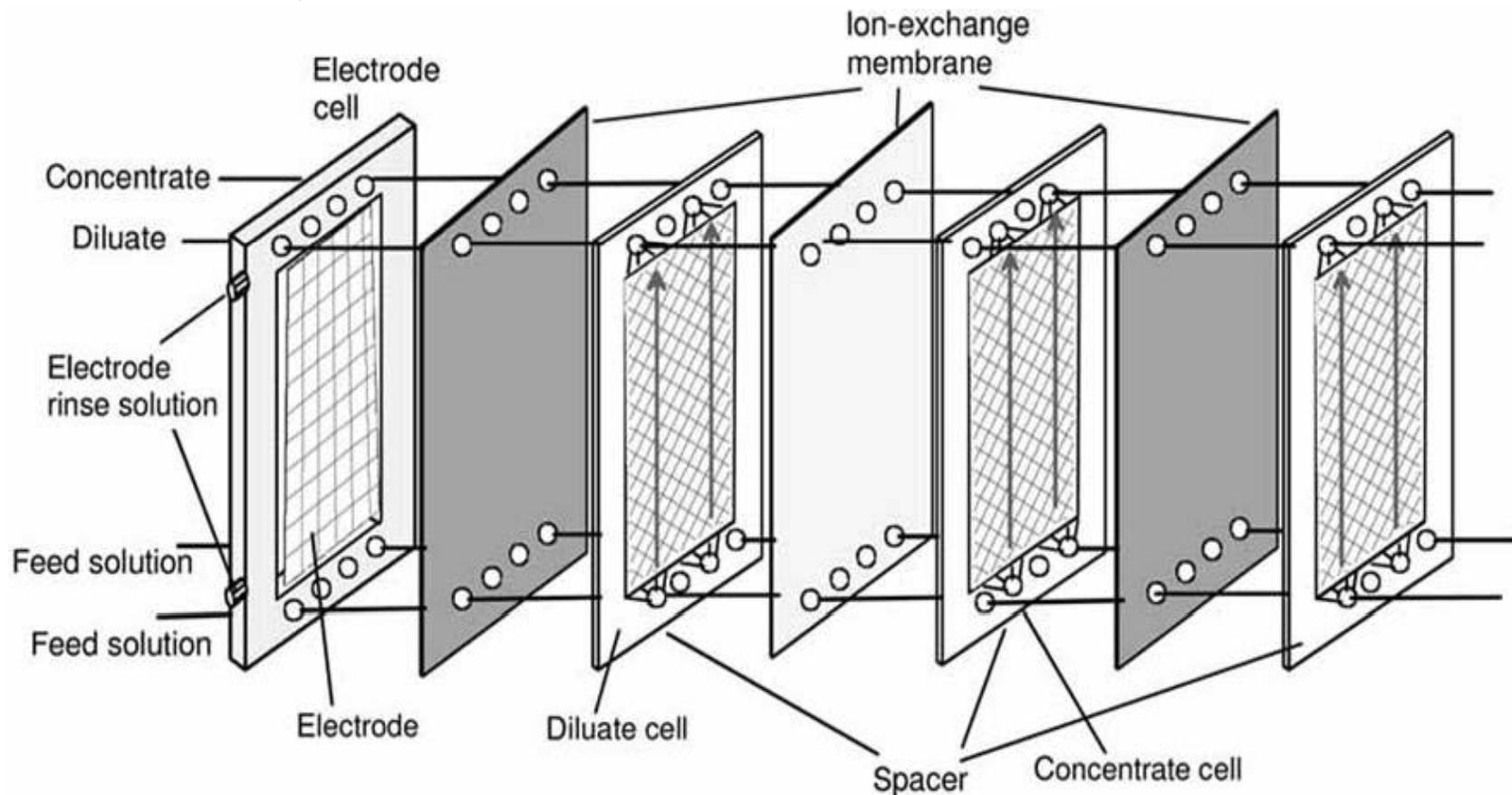
a cation and an anion-exchange membrane, and a diluate and concentrate cell form a cell pair

Strathmann, 2007

CONVENTIONAL ELECTRODIALYSIS



The electrodialysis stack



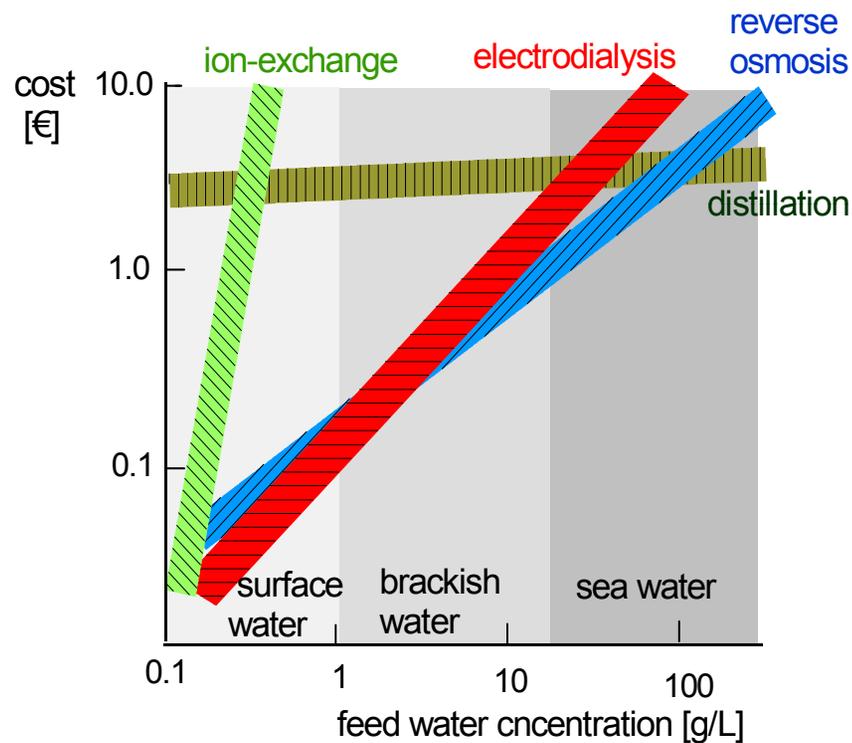
Courtesy of Heiner Strathmann. Used with permission.

an electrodialysis stack is composed of 100 to 400 cell pairs between electrode compartments

CONVENTIONAL ELECTRODIALYSIS

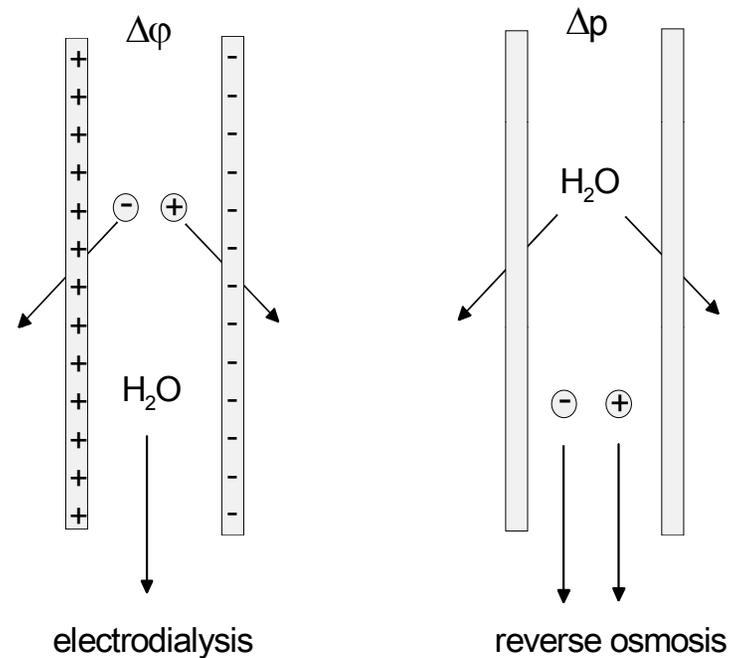


Water desalination costs



costs estimated for a required product concentration of < 0.2 g/L (200 ppm)

Process principles of electrodesalination and reverse osmosis



irreversible energy loss proportional to ion transport
 $(E_{irr} = z_i F \Delta C_i U V)$

irreversible energy loss proportional to water transport
 $(E_{irr} = \Delta p V_{water})$

Strathmann, 2007

Solar Distillation

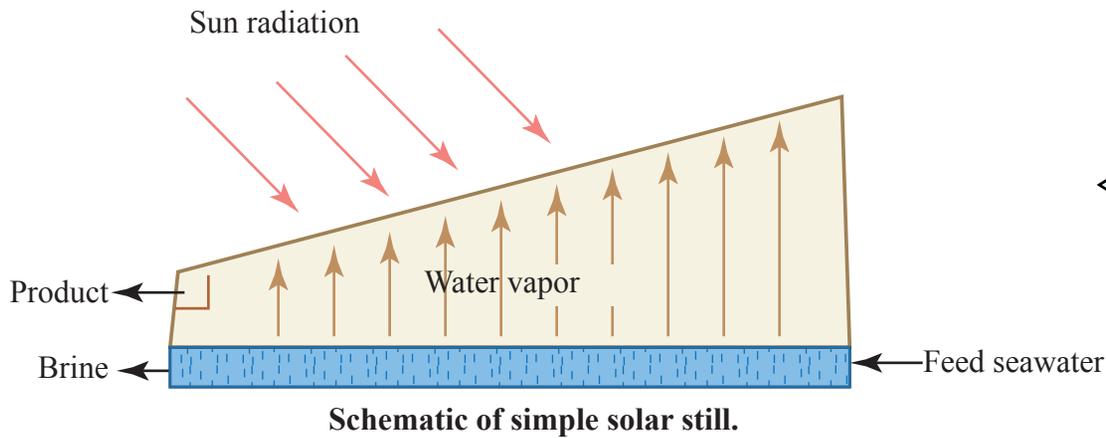
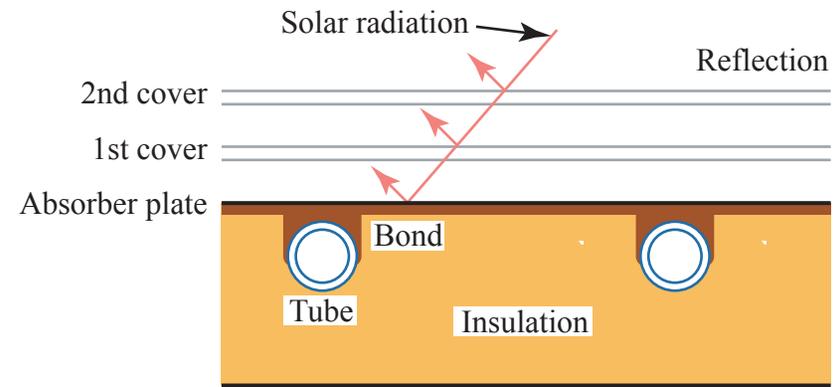
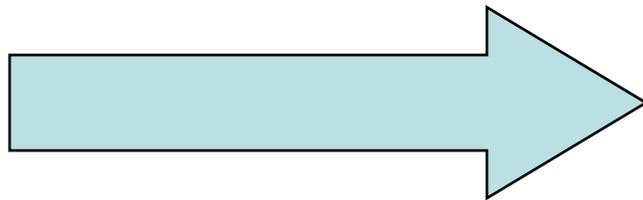


Figure by MIT OpenCourseWare.

Optimally bad?

~ 1 L/m²-day



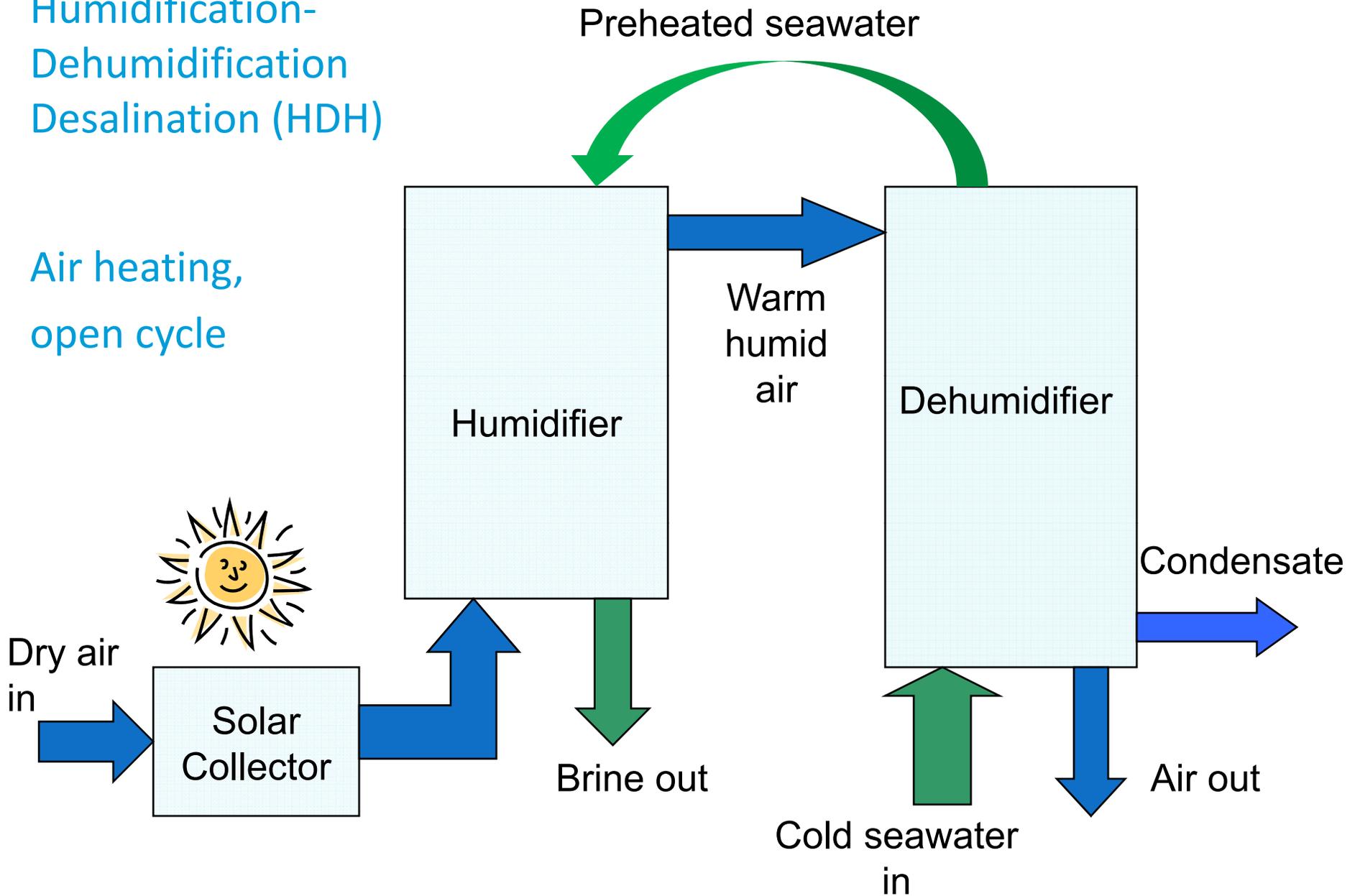
Sheet-and-tube solar collector.

Figure by MIT OpenCourseWare.

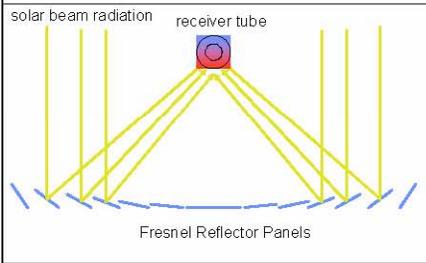
*..obvious improvements...
 ...multiple glazing to control
 IR and convective loss...*

Humidification- Dehumidification Desalination (HDH)

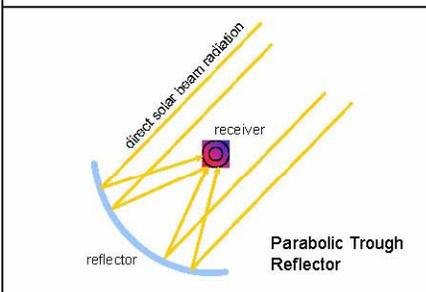
Air heating,
open cycle



Linear Fresnel Concentrating Solar Thermal Collector



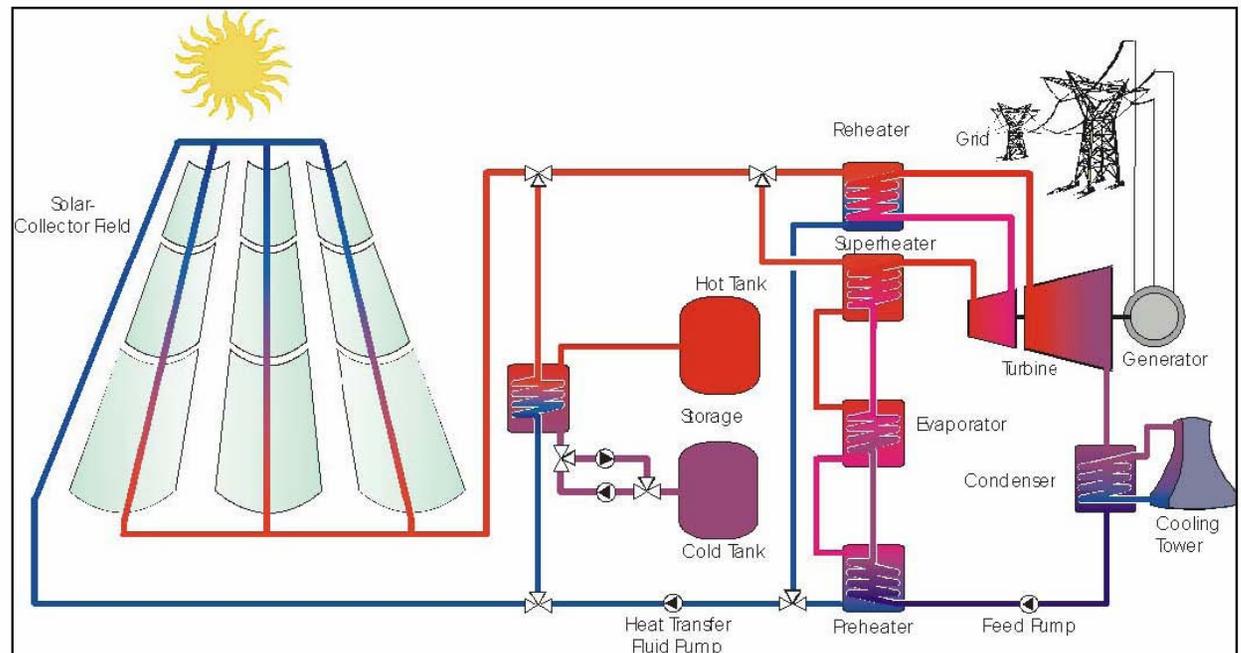
Parabolic Trough Concentrating Solar Thermal Collector



Concepts from concentrating solar power can be applied to solar distillation...generate electricity, then make water from waste heat or electricity

...theoretical performance can be ~100X better than solar still

Ref: Trieb et al., Nov. 2007



Major concerns in desalination systems

- **Cost:** hardware, site development
- **Cost:** energy consumption
- **Cost:** maintenance
 - Scaling, from precipitation of salts (has a *controlling* influence on design of thermal systems)
 - Fouling, from bacteria and other deposits
 - Degradation of membranes
 - Corrosion of hardware
- Disposal of **brine efflux**, environmental impact
- **Reliability, distribution,...**

Total water cost distribution in SWRO

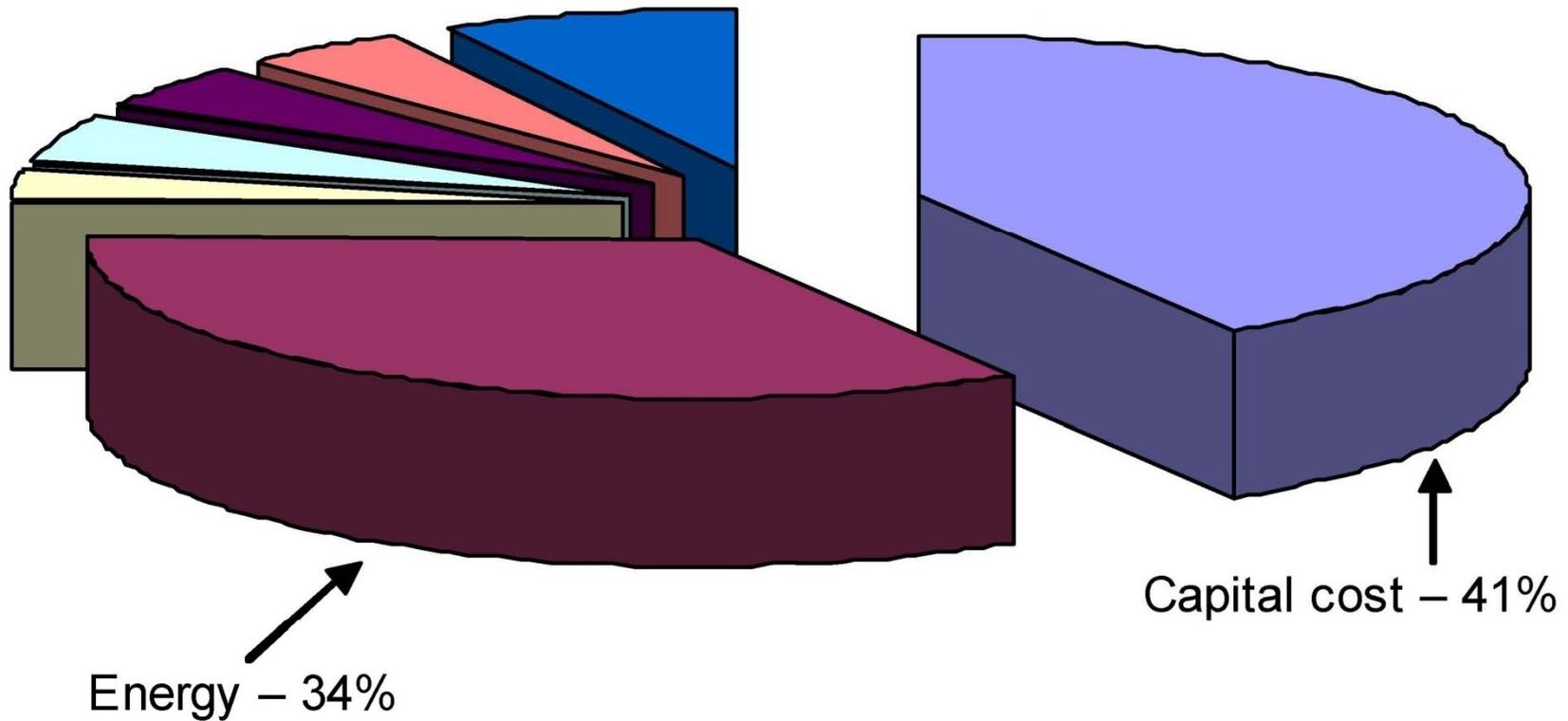
Membrane replacement – 3%

Labor – 5%

Maintenance and parts – 5%

Consumables – 5%

Others & contingency – 7%



Figures from Wilf, M., and M. Balaban. *Membrane Desalination and Membrane Filtration*. L'Aquila, Italy: European Desalination Society, 2007. Used with permission.

Source: Sommariva, 2007

Prices for consumers in office spaces occupying 4180 m² of city space and using 10,000 m³/y

Country	\$/M ³
Germany	\$1.91
Denmark	\$1.64
Belgium	\$1.54
Netherlands	\$1.25
France	\$1.23
United Kingdom	\$1.18
Italy	\$0.76
Finland	\$0.69
Ireland	\$0.63
Sweden	\$0.58
Spain	\$0.57
U.S.A.	\$0.51
Australia	\$0.50
South Africa	\$0.47
Canada	\$0.40

Figure by MIT OpenCourseWare.

Some notes on 2.500 this semester

- Term project on **purifying village well water** in Haiti (will be introduced in March)
- Visit **Koch Membrane Systems** in Wilmington on Tuesday, March 10
- Visit **GE Ionics** in Watertown on Friday, March 13
- Final project presentations on Tuesday May 12

- GRADING
 - Approximately: homework (55%), term project (35%), class participation (10%)

- *Listeners must register as listeners*

Readings to accompany this lecture, all on Course Website

- Michael Specter, *The Last Drop*, The New Yorker, 23 October 2006, pp. 60-71.
- *Water for People, Water for Life*, UNESCO, 2003 (skim)
- Section on Water Quality from Reynolds & Richards, *Unit Operations...*, 1995
- EPA Primary Drinking Water Standards
- Peruse the other general articles on water if you are interested...

References

1. *Water for People, Water for Life*, United Nations World Water Development Report. Paris: UNESCO Publishing, 2003.
2. UNESCO Photobank, <http://photobank.unesco.org/exec/index.htm>
3. J.P. Holdren, "Science and Technology for Sustainable Well-Being," *Science*, 319 (25 Jan 2008) 424-434.
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6. J.E. Miller, "Review of water resources and desalination technologies," Sandi report SAND-2003-0800, 2003. Albuquerque: Sandia National Laboratory.
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8. T.D. Reynolds and P.A. Richards, *Unit Operations and Processes in Environmental Engineering*, 2nd ed. Boston: PWS Publishing Co., 1995.
9. A.C. Twort, D.D. Rathnayaka, and M.J. Brandt, *Water Supply*, 5th ed., IWA Publishing.
10. M. Wilf, *The Guidebook to Membrane Desalination Technology*. L'Aquila Italy: Balaban Desalination Publications, 2007.
11. C. Fritzmann, J. Lowenberg, T. Wintgens, T. Melin, "State-of-the-art reverse osmosis desalination," *Desalination*, 216 (2007) 1-76.
12. M. Wilf and M. Balaban, *Membrane Desalination and Membrane Water Filtration*, European Desalination Society intensive course notes, L'Aquila, Italy, February 2007.
13. S.D. Faust and O.M. Aly, *Chemistry of Water Treatment*, 2nd ed. Boca Raton: Lewis Publishers/CRC, 1998.
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15. Seawater Salinity Graphic from Texas A&M University Physical Oceanographic Course Web site. http://oceanworld.tamu.edu/resources/ocng_textbook/chapter06/chapter06_03.htm
16. M. Al-Ghamdi, *Saline Water Conversion Corporation: Challenge, Achievement, and Future Prospective*. Lecture notes, April 2006.
17. C. Sommariva, Short Course Notes on Reverse Osmosis Desalination, 2007.
18. H. Strathmann, [*Electromembrane processes: State-of-the-art processes and recent developments, lecture notes*](#), 2007.
19. Dr. Franz Trieb et al., *Concentrating Solar Power for Seawater Desalination*, German Aerospace Center (DLR), Institute of Technical Thermodynamics, Nov. 2007.