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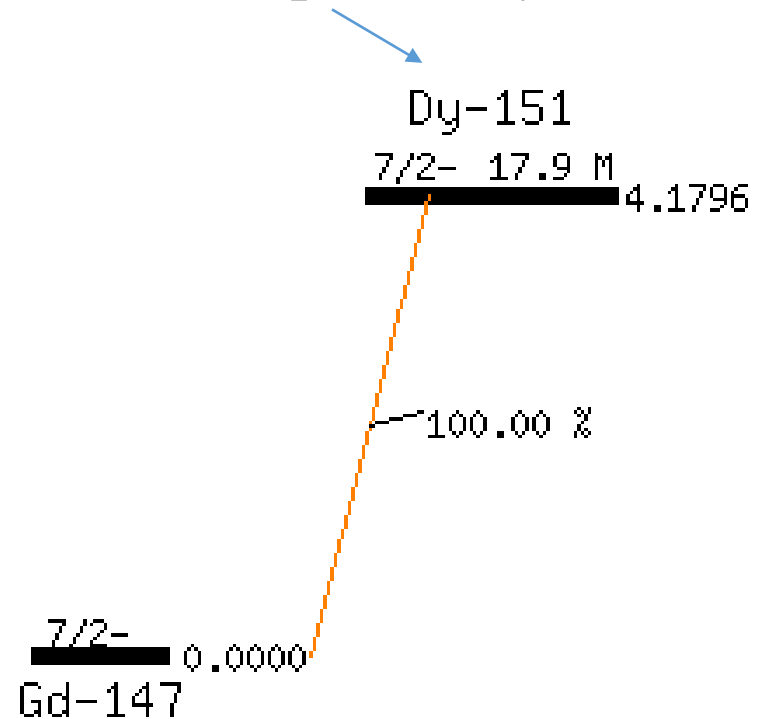
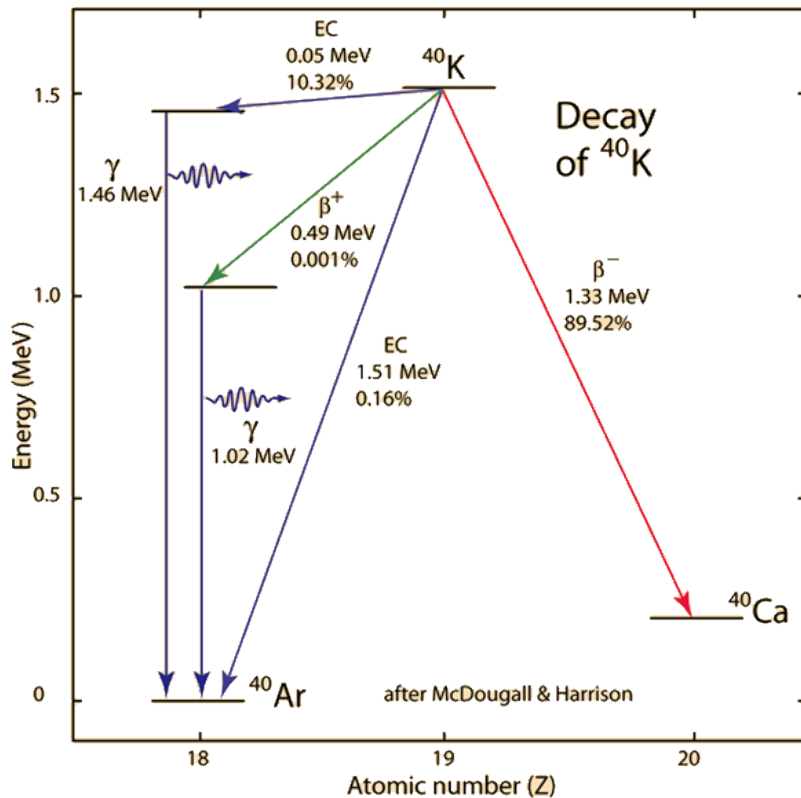
# Slides for Radioactive Decay

2024

# But First... Decay Diagrams

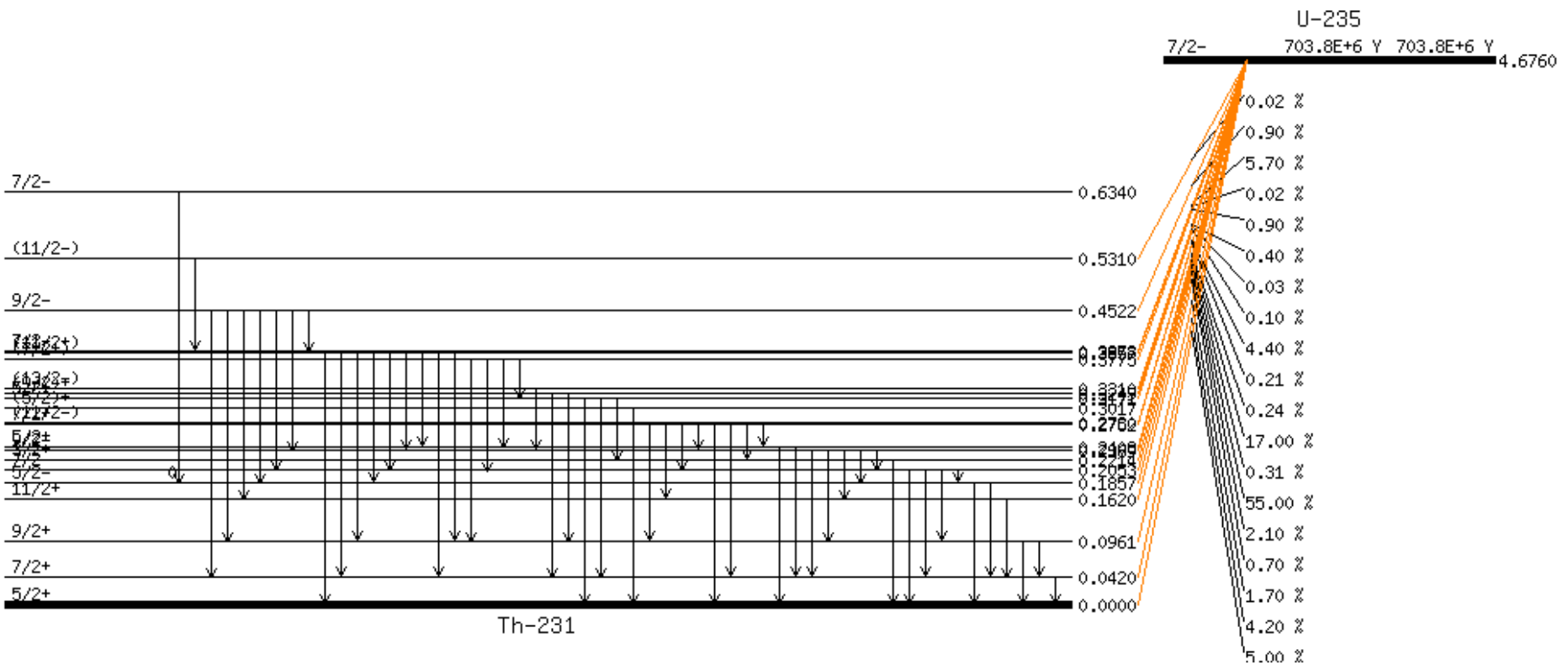
<http://hyperphysics.phy-astr.gsu.edu/hbase/nuclear/kar.html>

$^{40}\text{K}$  gives the most generalized example, minus alpha decay



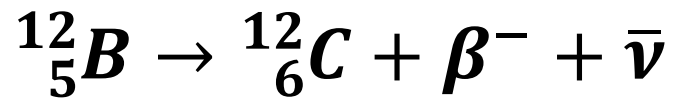
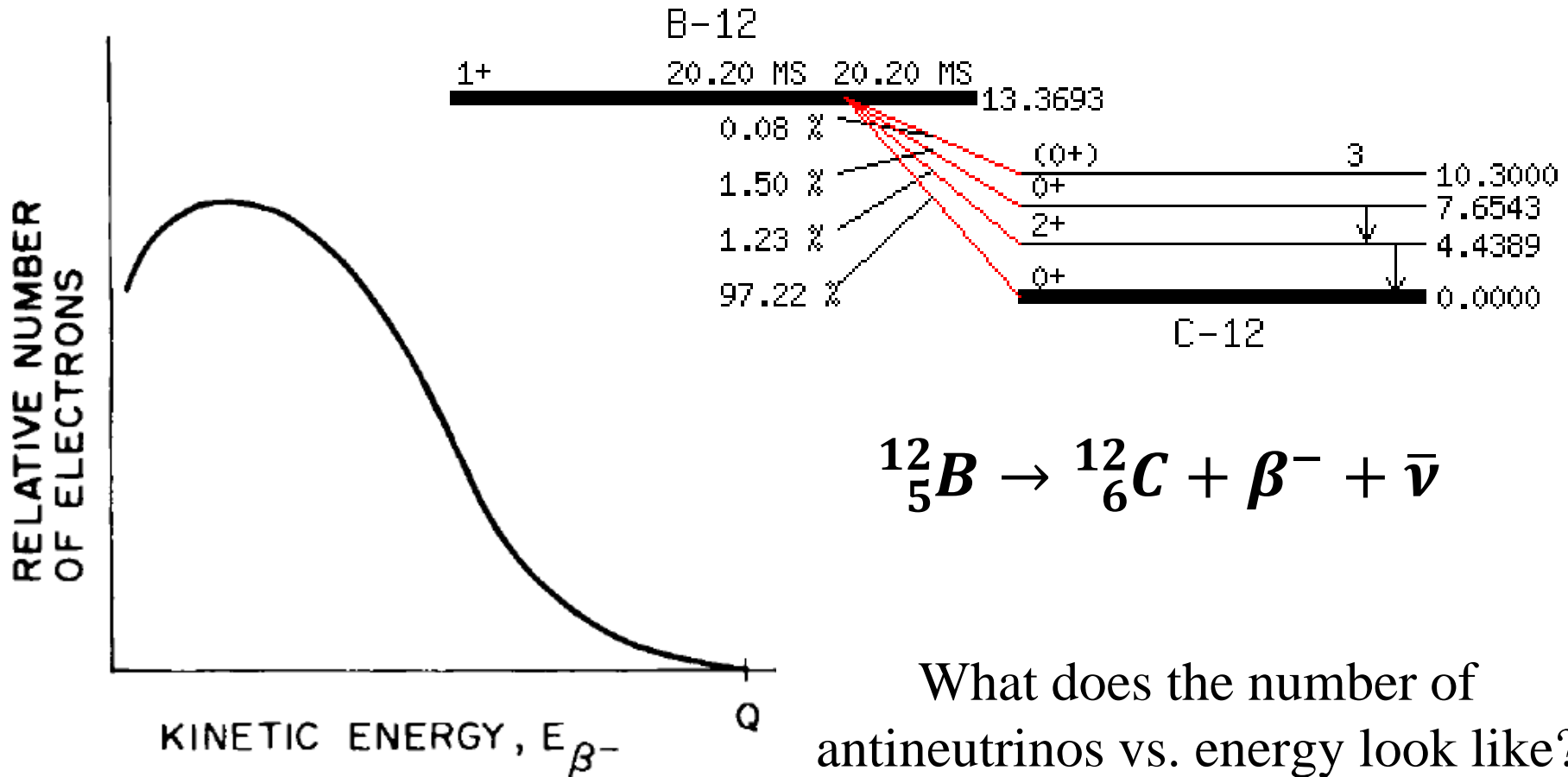
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# Alpha Decay Diagrams



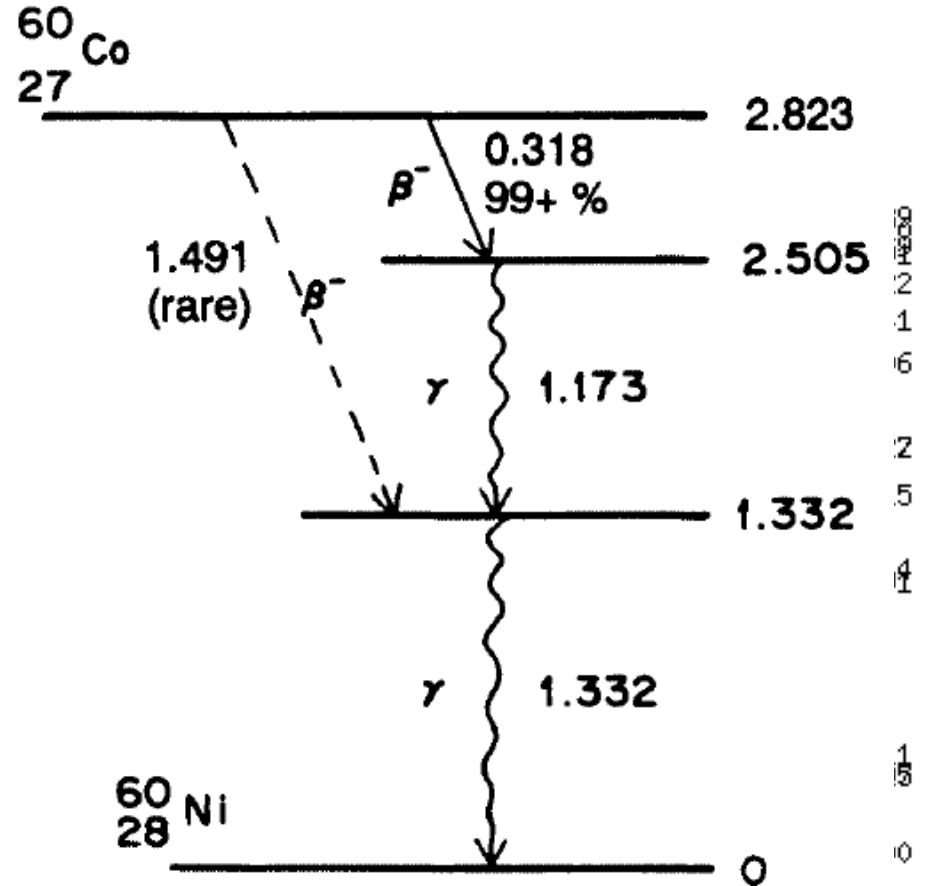
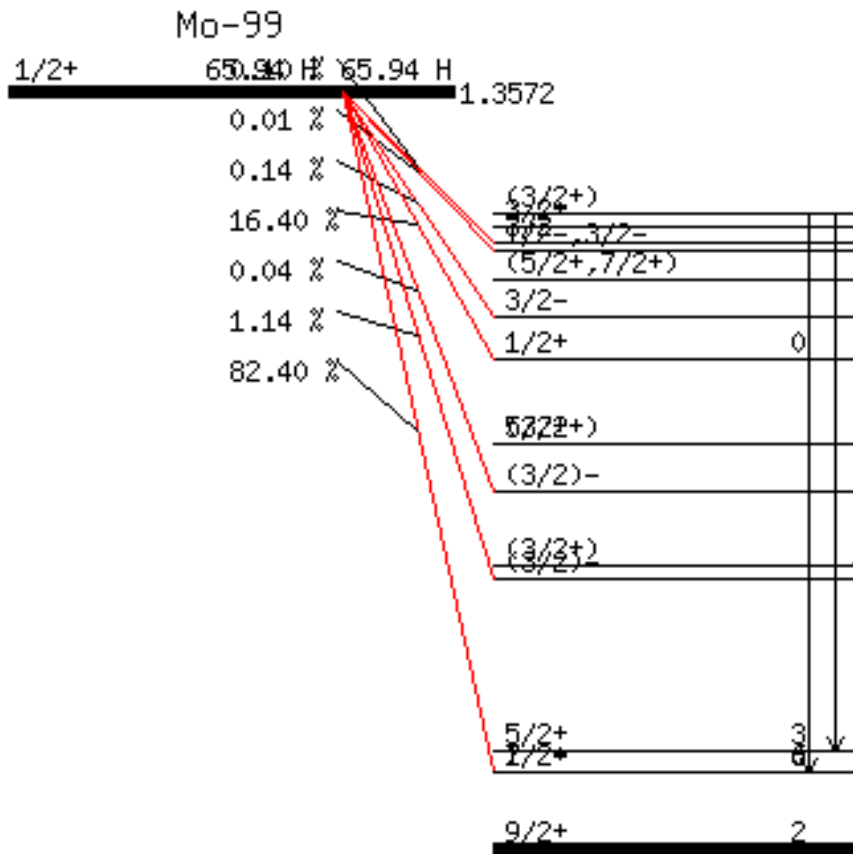
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# Beta Decay Diagrams and Energetics



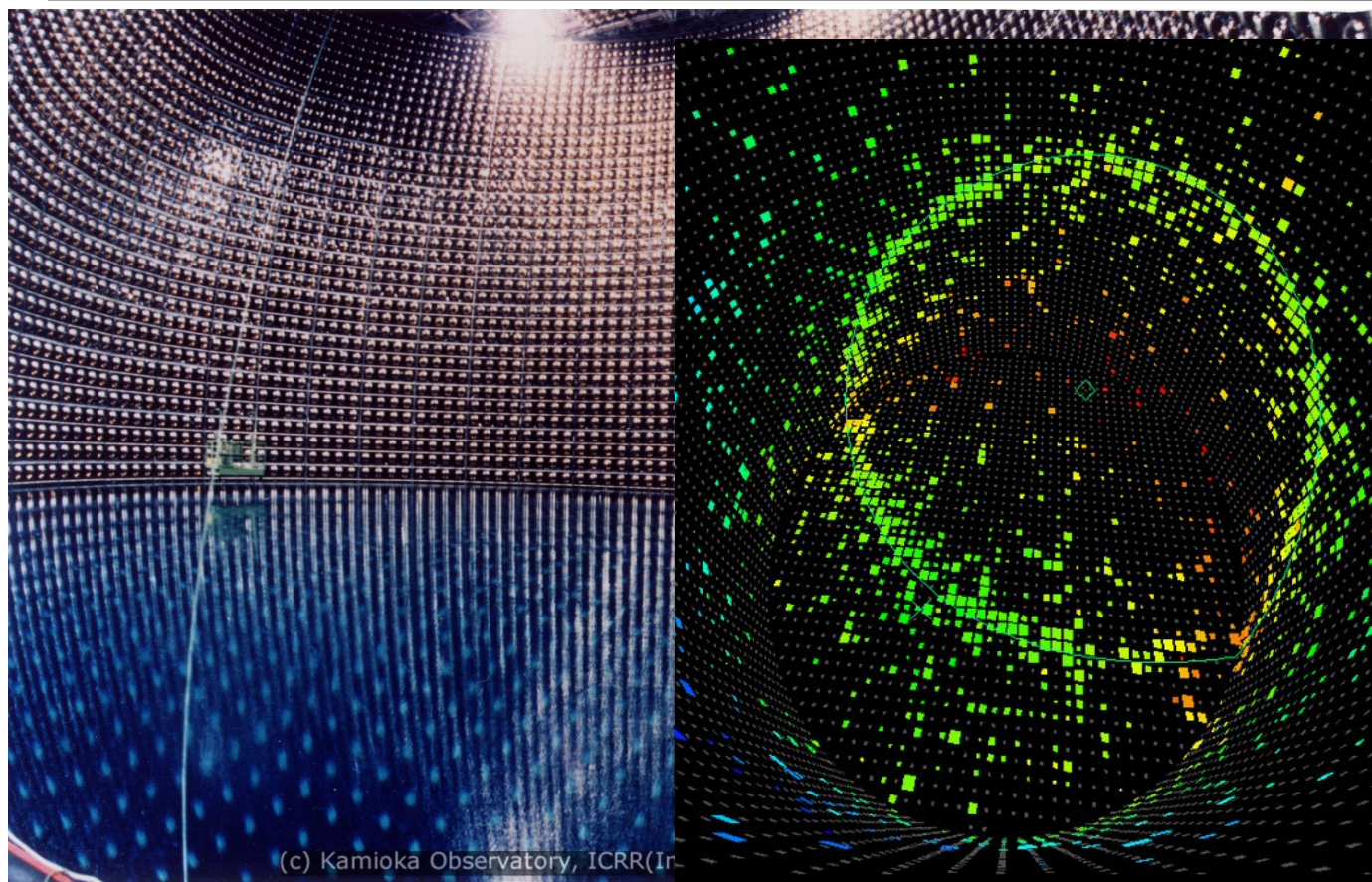
What does the number of antineutrinos vs. energy look like?

# Notable Beta Decay Reactions



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# Detecting Neutrinos – Super Kamiokande, Japan



news.bbc.co.uk/2/hi/science/nature/1664447.stm

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**BBC NEWS**

You are in: **Sci/Tech**  
Monday, 19 November, 2001, 12:49 GMT

## Particle physics telescope explodes



**COMMONWEALTH GAMES**

**BBC SPORT** A defective photomultiplier tube exploded, setting off a chain reaction

**BBC Weather**

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**By BBC News Online science editor Dr David Whitehouse**

One of the world's leading particle physics instruments has been severely damaged in an accident.

The underground Super-Kamiokande Observatory in Japan detects elusive neutrino particles from space by using photomultiplier tubes to register the flashes of light they produce when they pass through a huge tank of water.

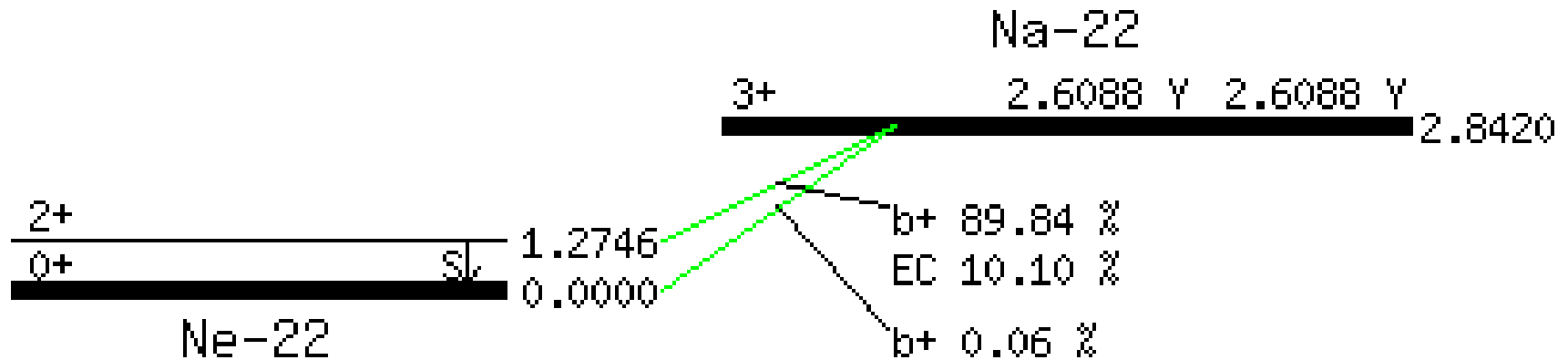
**“ We will rebuild the detector. There is no question ”**

**Yoji Totsuka, Kamioka Observatory**

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# Positron Decay Diagrams and Required Energetics



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$$Q_{\beta^+} = \Delta_P - \Delta_D - 2mc^2. \quad Q > 1.022 \text{ MeV}$$

Why  $2m_e c^2$ ? We emit one positron ( $E_{\text{rest}} = 0.511 \text{ MeV}$ ) *and* the daughter nucleus must shed one orbital electron to conserve charge.

# *Introduction to Positron annihilation spectroscopy measurements: Application to Irradiated Fe and some Y-Ti-O clusters in 14YWT*

---

---

X. Hu, D. Xu, and Brian D. Wirth<sup>1,#</sup>,  
with significant contributions from  
M. Alinger (GE), P. Asoka-Kumar<sup>2</sup>, G.R. Odette<sup>3</sup>, R.H. Howell<sup>2</sup>, P.A. Sterne<sup>2</sup>,  
Y. Nagai<sup>4</sup>, and M. Hasegawa<sup>4</sup>

20 March 2014

Presented at the  
U. Michigan Workshop

*This work was partially supported by the the U.S. Department of Energy, Office of Fusion Energy  
Sciences, Office of Basic Energy Sciences and Office of Nuclear Energy, Science and Technology.*

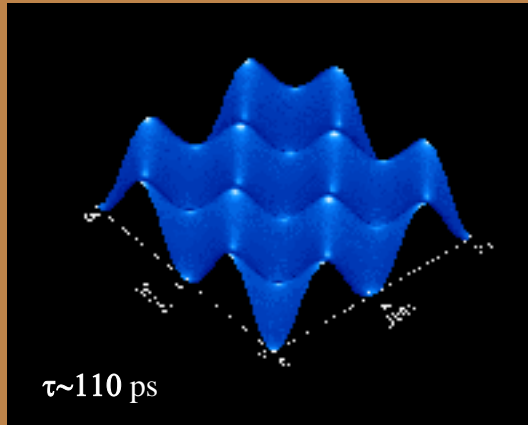


# Positron Annihilation Spectroscopy

- Positrons are a tremendously powerful and self-seeking probe of the chemical, electronic and & magnetic properties of vacancies/vacancy-clusters and locally enriched regions (precipitates) of stronger positron affinity in metallic alloys.

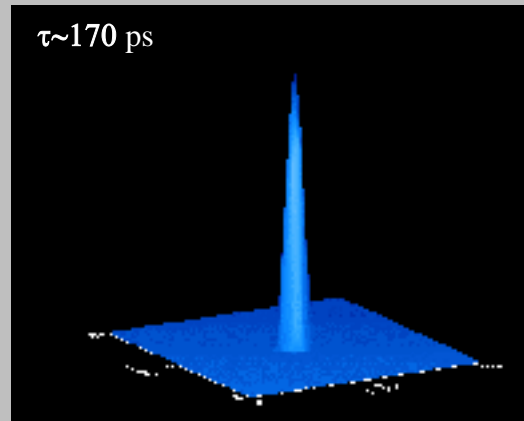
- localize in open-volume regions (vacancies, voids, other defects) due to lack of positively charged atomic nuclei
- localize in regions of higher positron affinity (elemental specific, eV)  
Cr: -2.62 Mn -3.72 Fe: -3.84 Cr: -2.62 Ni: -4.46 Cu: -4.81 Zr: -3.98

Bulk metal without defects



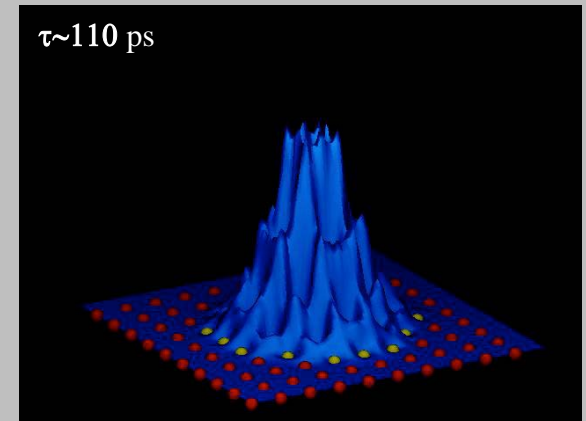
Delocalized positron density

Vacancy defects



Strongly localized positron density

Embedded particles of stronger positron affinity

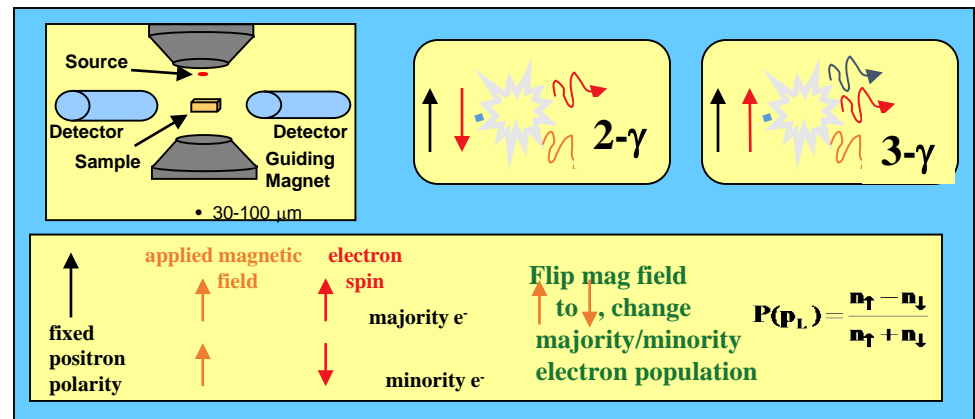
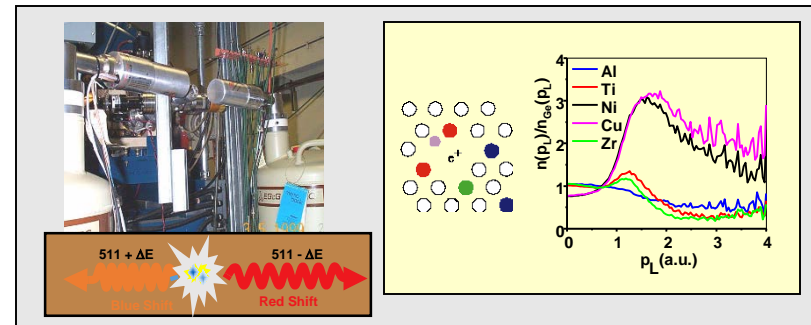
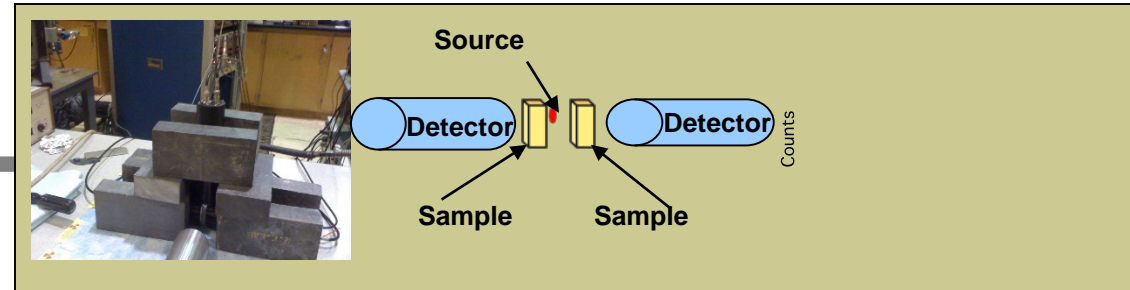


- Positron annihilation experiments must be carefully analyzed, but strong theoretical foundation exists; especially when combined with complementary techniques (3DAP, SANS, TEM, PIA, mechanical properties, ...)

# Positron Annihilation Spectroscopy Methods

## Methods:

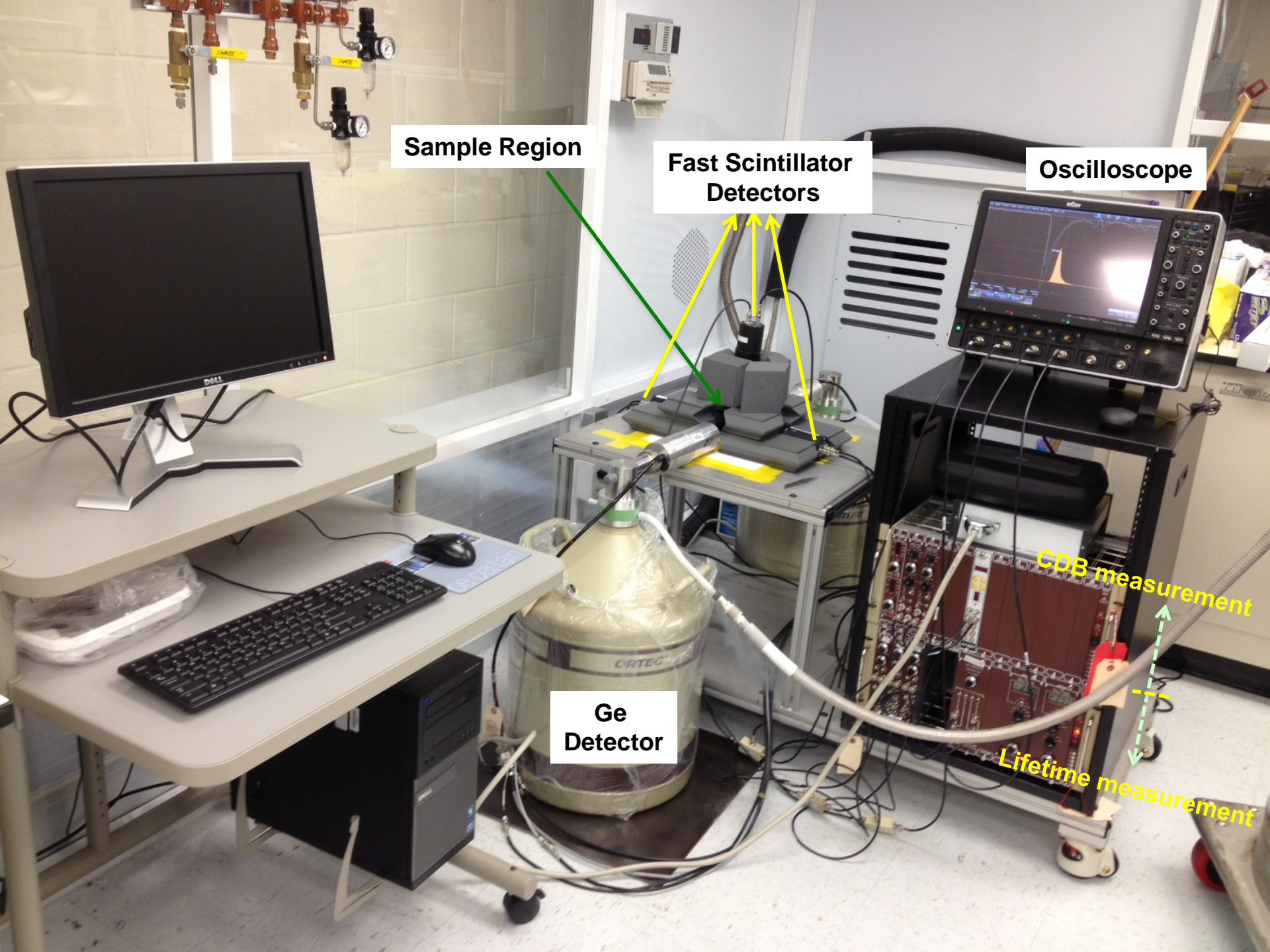
- Positron lifetime (*correlates with electron density; vacancies & vacancy cluster size*)
- Coincidence Doppler Broadening, CDB/OEMS\* (*e<sup>-</sup> momentum; composition - Vacancies influence low momentum; chemical variations generally observed at high momentum*)
- Magnetic, polarized CDB/OEMS (*majority & minority e<sup>-</sup>; magnetism*)\*\*
- 2D ACAR (*Fermi surface*)
- Age Momentum Correlation, AMOC (*distinguish copper vs vacancy trapping*)



\* P. Asoka-Kumar, M. Alatalo, V.J. Ghosh, A.C. Kruseman, B. Nielson, K.G. Lynn, *Phys Rev Let* **77** (1996) 2097.

\*\* P. Asoka-Kumar, B.D. Wirth, et al., *Phil. Mag. Lett.* **82** (2002) 609.





Sample Region

Fast Scintillator  
Detectors

Oscilloscope

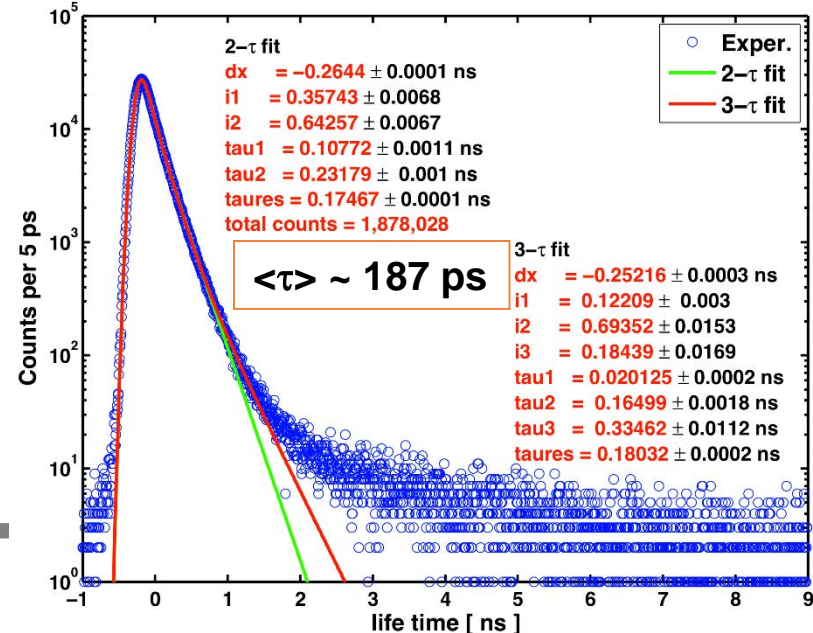
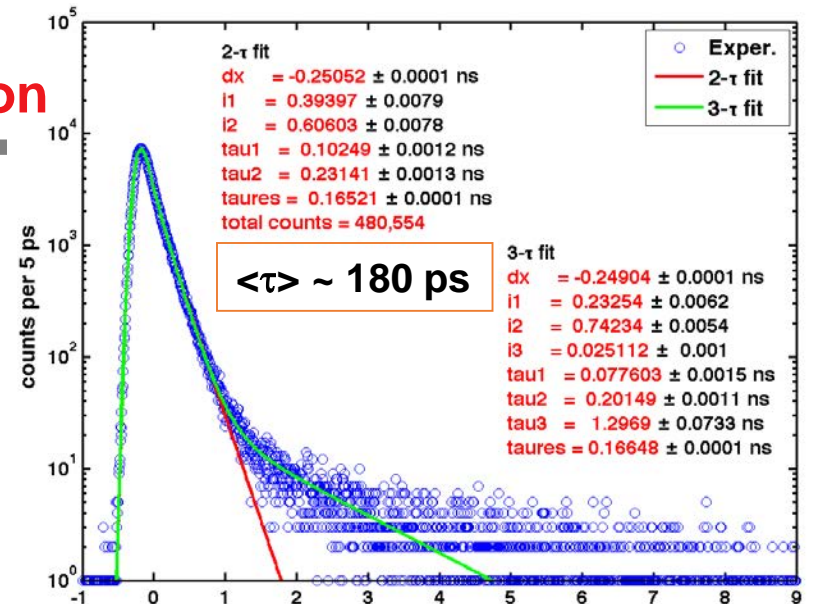
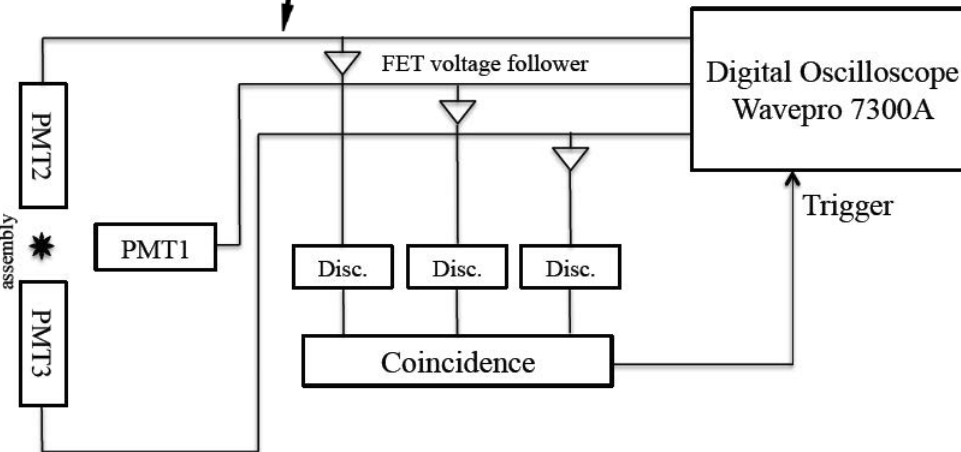
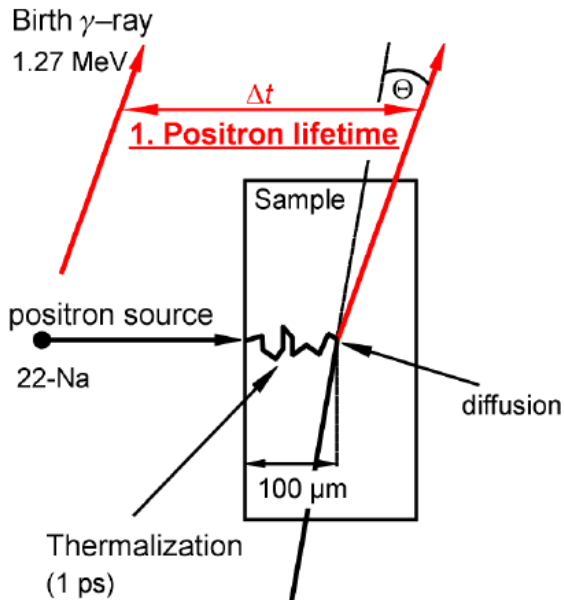
Ge  
Detector

CDB measurement

Lifetime measurement

# Positron Lifetime Measurement

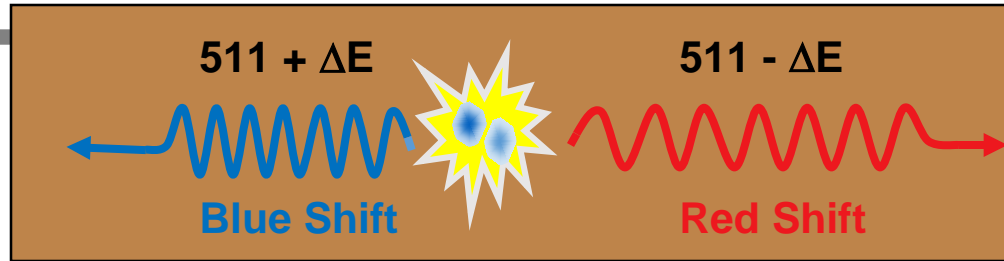
Lifetimes determined by measuring the time between **implantation** and **annihilation**



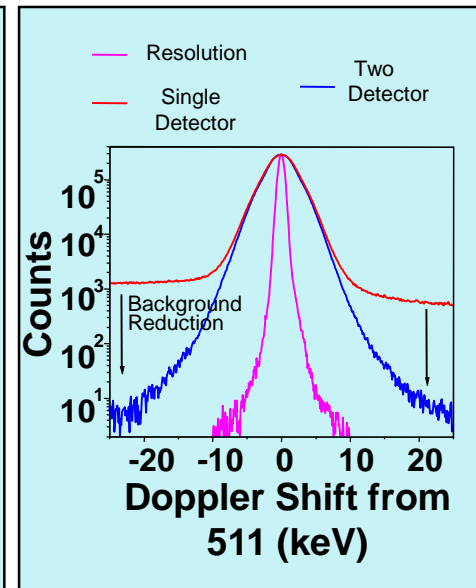
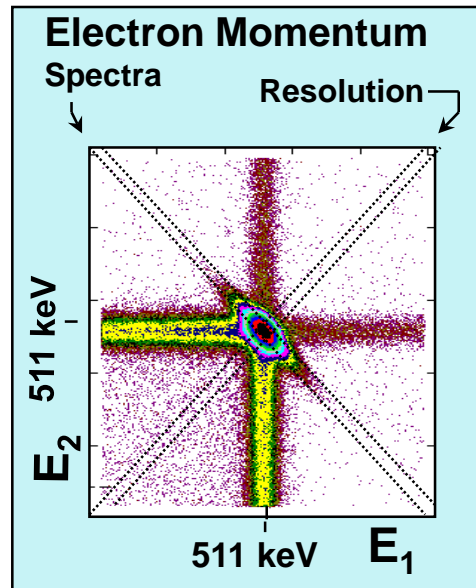
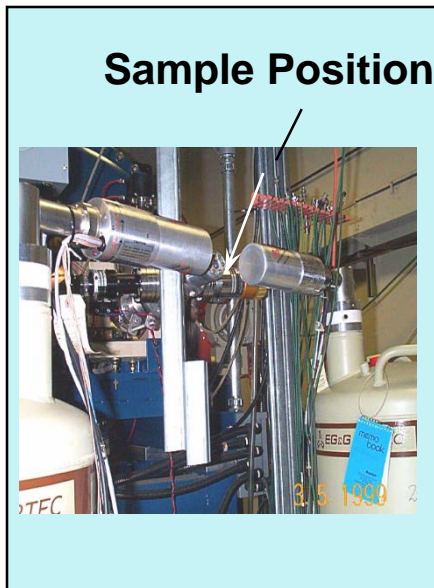


# Coincidence Doppler Broadening

Doppler shift probes the local electron momentum



Doppler shift,  $\Delta E$   
Is proportional to  
electron momentum,  $p_L$

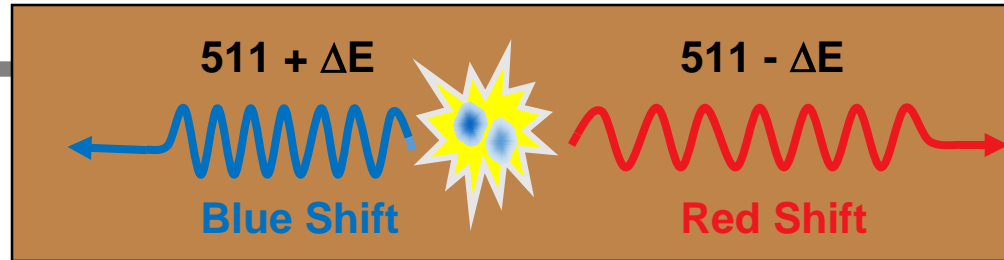


Coincident,  
two - detector  
spectroscopy  
improves  
signal-to-noise  
ratio

**Kinematic sections provide momentum spectra of orbital electrons, whose momenta are element-specific**

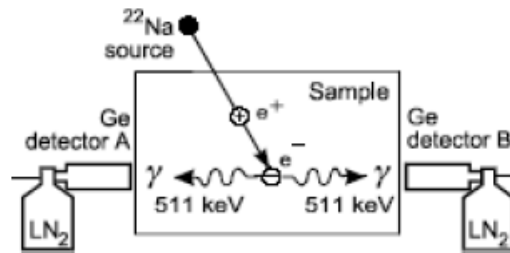
# Coincidence Doppler Broadening

Doppler shift probes the local electron momentum



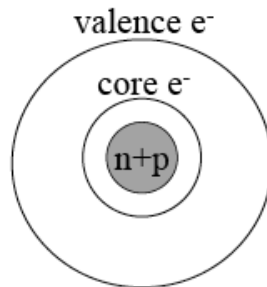
Doppler shift,  $\Delta E$   
Is proportional to  
electron momentum,  $p_L$

coincidence Doppler broadening (CDB) → Orbital Electron Momentum Spectra (OEMS)

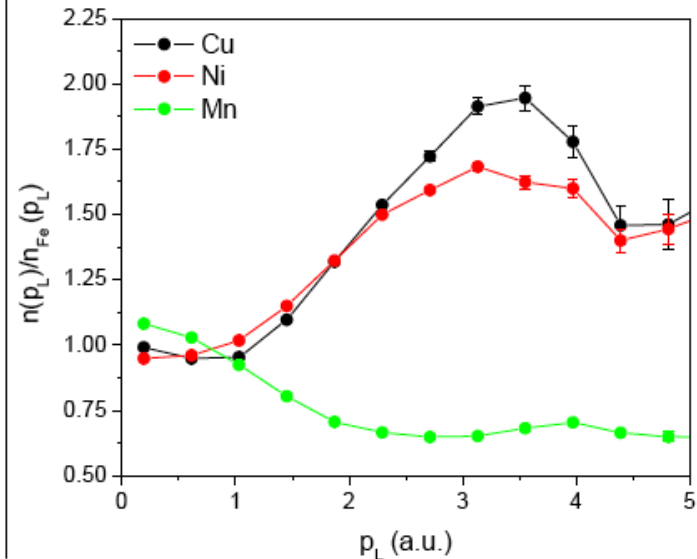


$$\Delta E = \frac{p_L c}{2}$$

1 a.u. =  $7.28 \text{ mrad} \times m_0 c$   
 $m_0$  = electron rest mass  
 $c$  = speed of light



valence  $e^-$  - low momentum, open-volume character  
core  $e^-$  - high momentum, elemental character

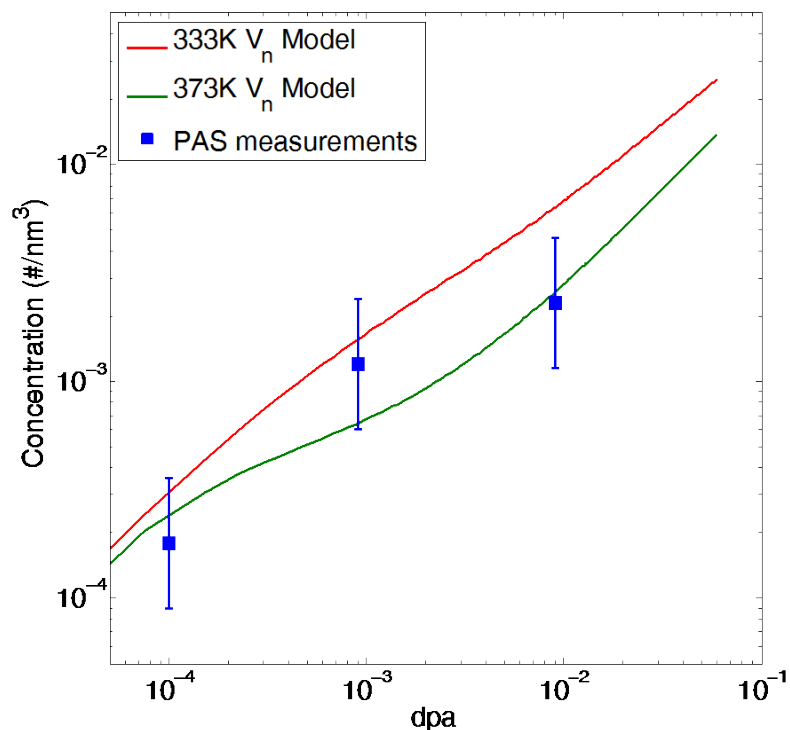
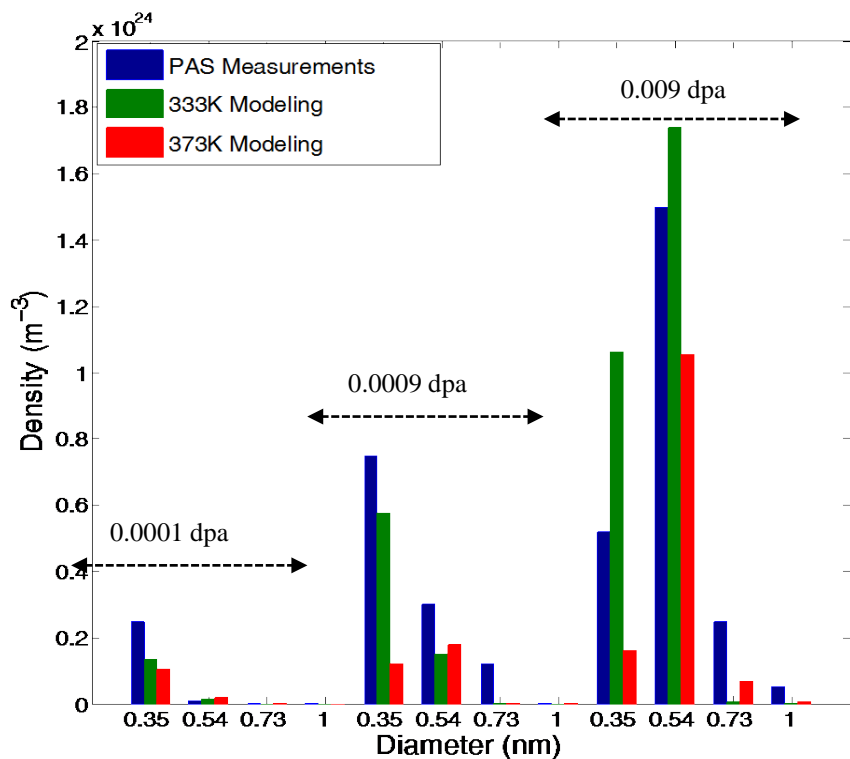


# Cluster Dynamics modeling of radiation damage in neutron irradiated Fe: Vacancy cluster comparisons with positrons\*\*

Eldrup etc. applied 'trapping model' to get the rough information of vacancy clusters' distribution at different irradiation levels\*

Five-component analysis is used, four of which have fixed lifetimes: 200, 300, 400, and 500 ps, equivalent to three-dimensional vacancy clusters of sizes of about 0.35 (2V), 0.54 (7V), 0.73 (18V) and >1.0 (45V) nm in diameter, respectively.

*\*M. Eldrup, etc. J. Nucl. Mater. 307-311 (2002) 912-917*



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Size distribution of vacancy clusters

Total density of vacancy clusters



# Interested in PAS? Read More Here!

---

**X. Hu, D. Xu, T.S. Byun, and B.D. Wirth, “Modeling of Irradiation Hardening of Iron after Low Dose and Low Temperature Neutron Irradiation”, *Modeling and Simulation in Materials Science & Engineering* 22 (2014) 0655002**

**M.J. Alinger, S.C. Glade, B.D. Wirth, G.R. Odette, T. Toyama, Y. Nagai, and M. Hasegawa, “Positron annihilation characterization of nanostructured ferritic alloys”, *Materials Science and Engineering A* 518 (2009) 150-157.**

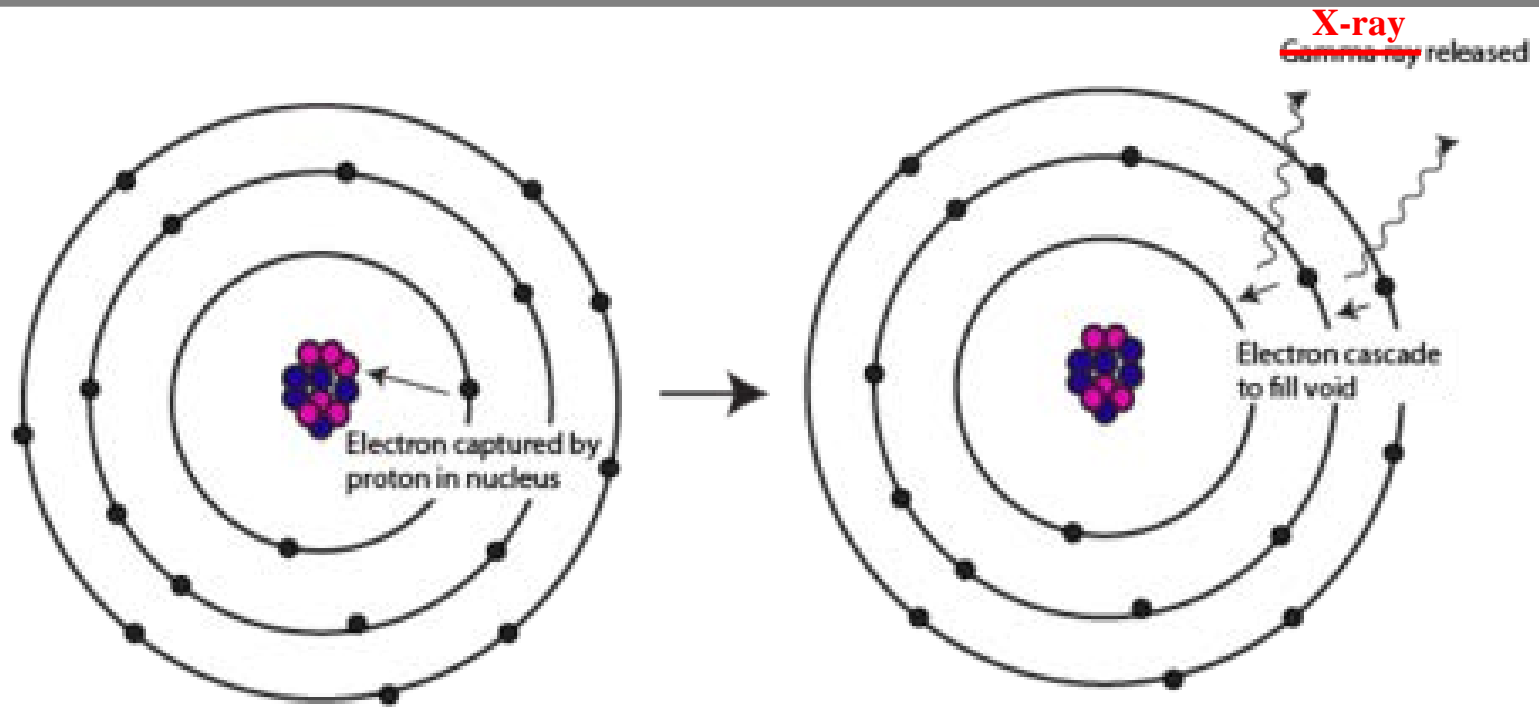
**S.C. Glade, B.D. Wirth, G.R. Odette and P. Asoka-Kumar, “Positron Annihilation Spectroscopy and Small Angle Neutron Scattering Characterization of Nanostructural Features in High-Nickel Model Reactor Pressure Vessel Steels”, *J. Nucl. Mater* 351 (2006) 197.**

**S.C. Glade, B.D. Wirth, G.R. Odette, P. Asoka-Kumar, P.A. Sterne, and R.H. Howell, “Positron annihilation spectroscopy and small angle neutron scattering characterizations of the effect of Mn on the nanostructural features formed in irradiated Fe-Cu-Mn alloys”, *Philosophical Magazine* 85 (2005) 629.**

**P. Asoka-Kumar, R. Howell, T.G. Nieh, P.A. Sterne, B.D. Wirth, R.H. Dauskardt, K.M. Flores, D. Suh, G.R. Odette, “Opportunities for materials characterization using high-energy positron beams”, *Applied Surface Science* 194 (2002) 160.**

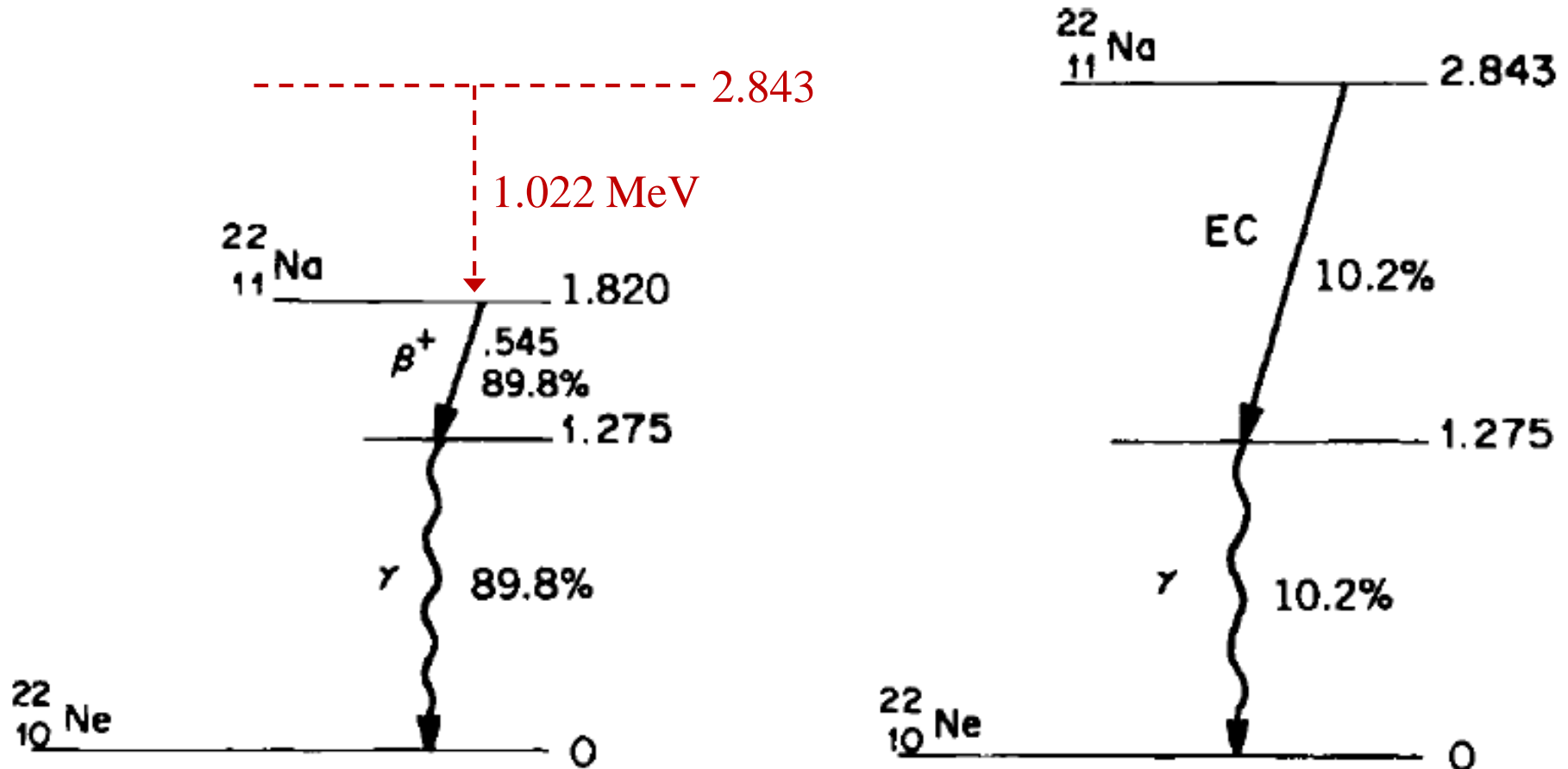
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# Electron Capture – Competes with Positron Decay

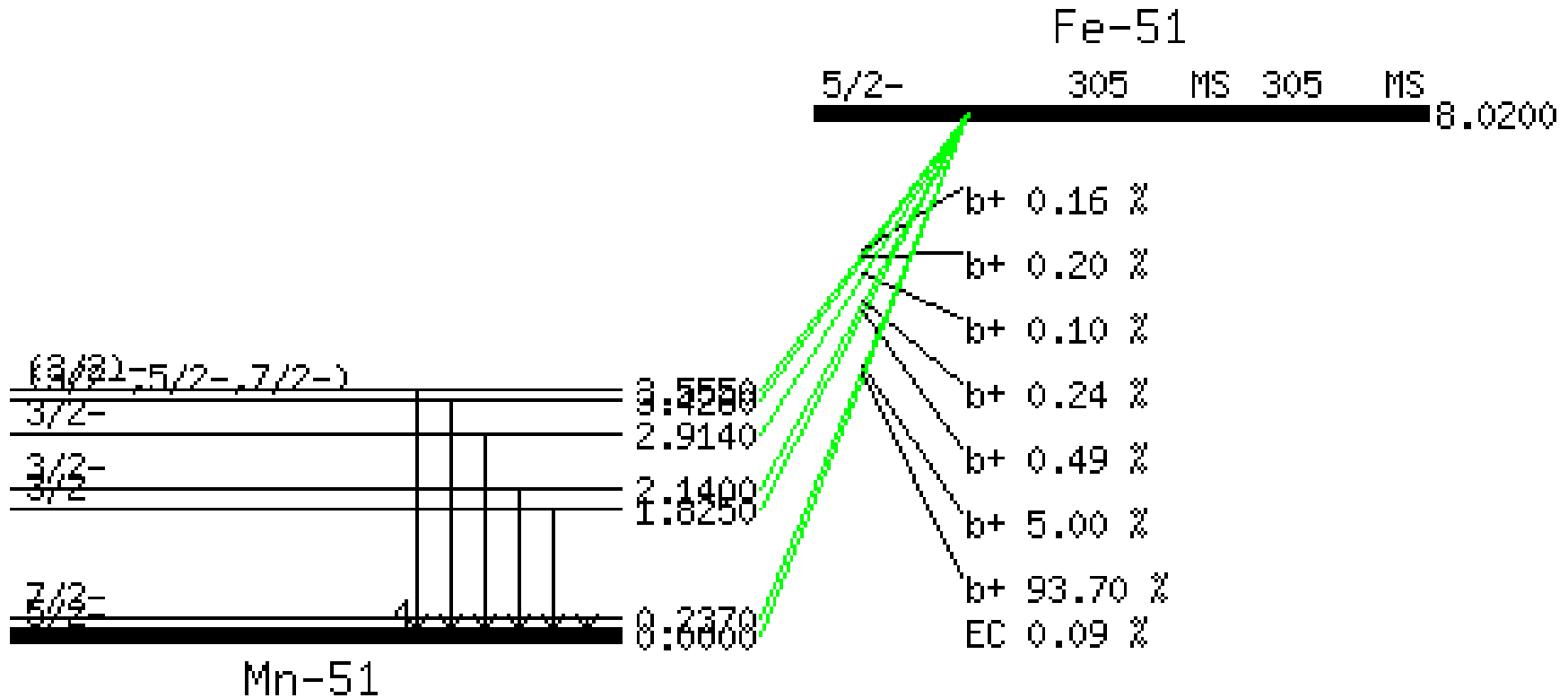


$$Q_{\text{EC}} = \Delta_{\text{P}} - \Delta_{\text{D}} - E_{\text{B}}.$$

# Electron Capture – Competes with Positron Decay

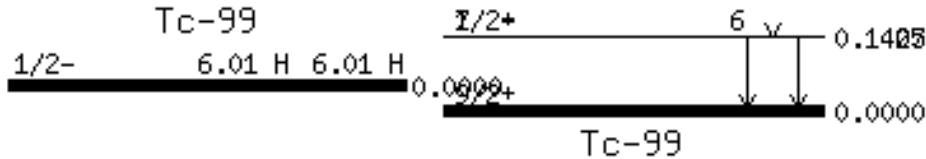


# Positron or Electron Capture?

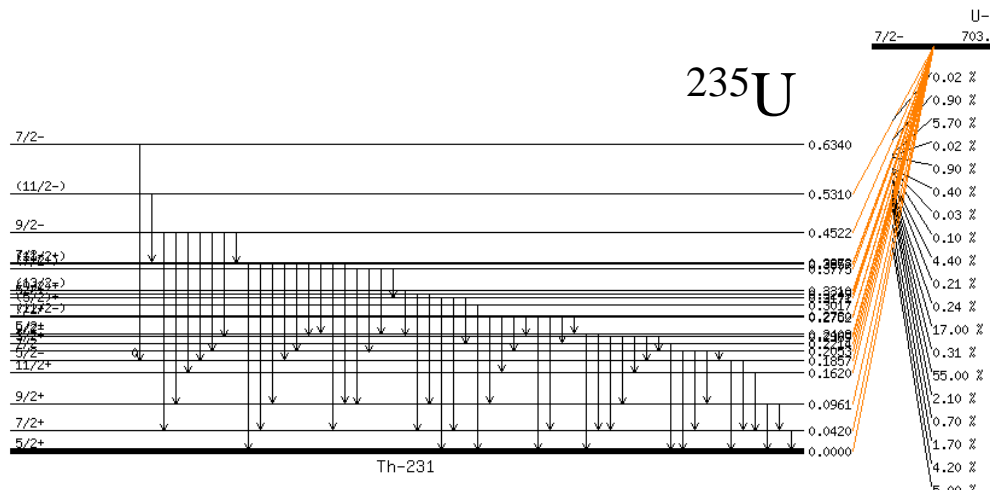
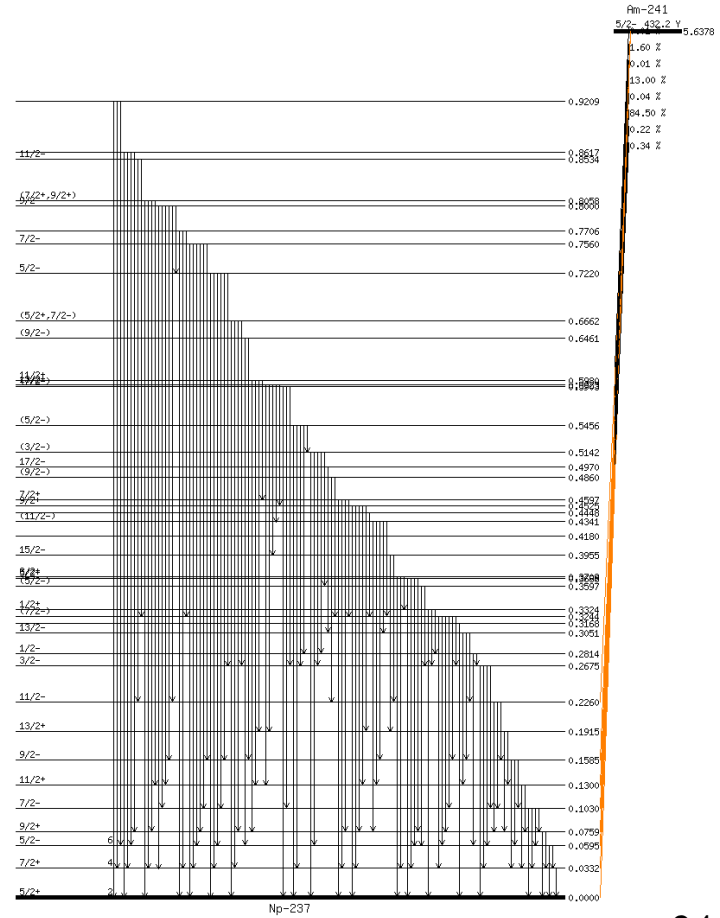


Higher Q-value is more likely to proceed via positron decay

# Gamma Decay (Isomeric Transition, or IT)



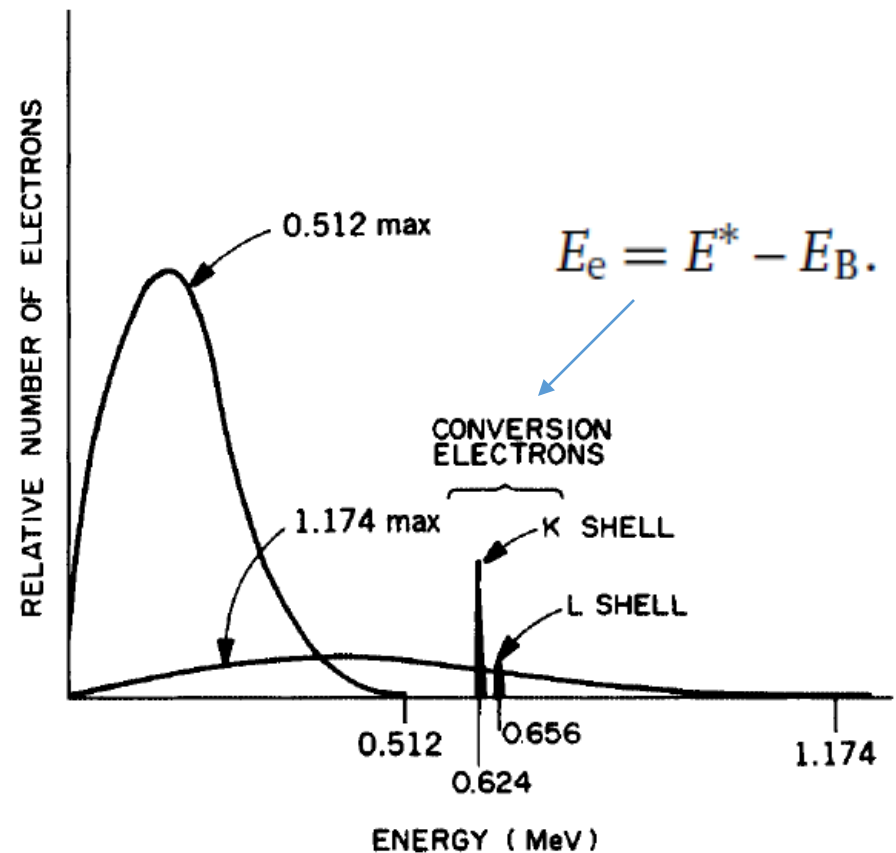
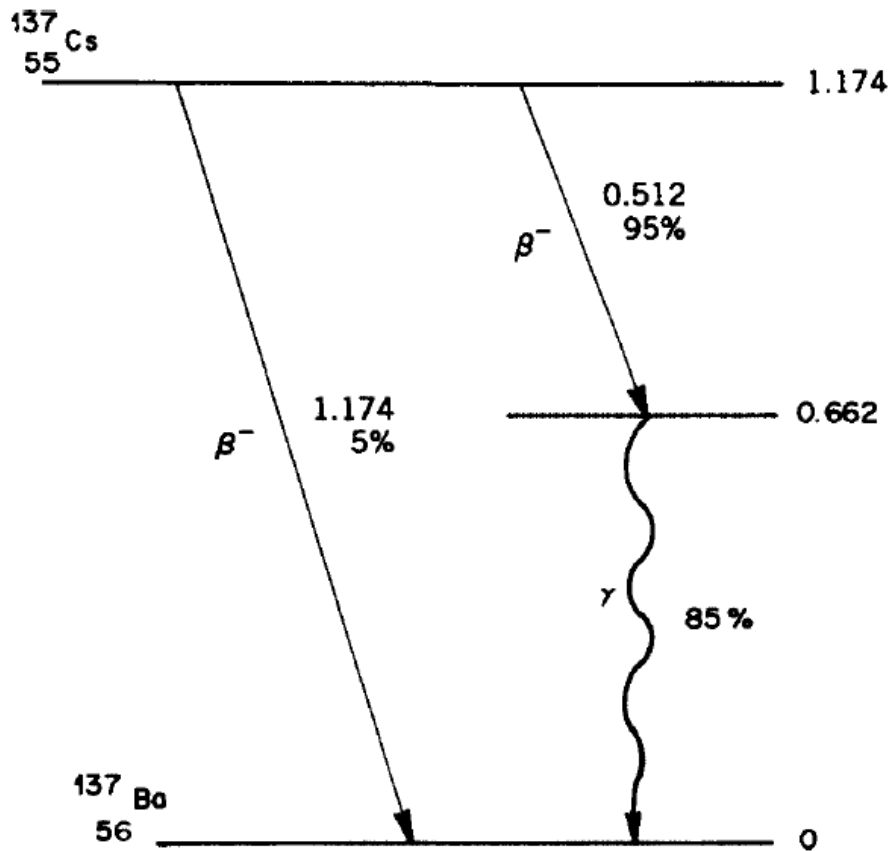
$$Q = BE_D - BE_P = E_\gamma$$



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<sup>241</sup>Am

# Internal Conversion (IC) Competes with Isomeric Transition (IT)

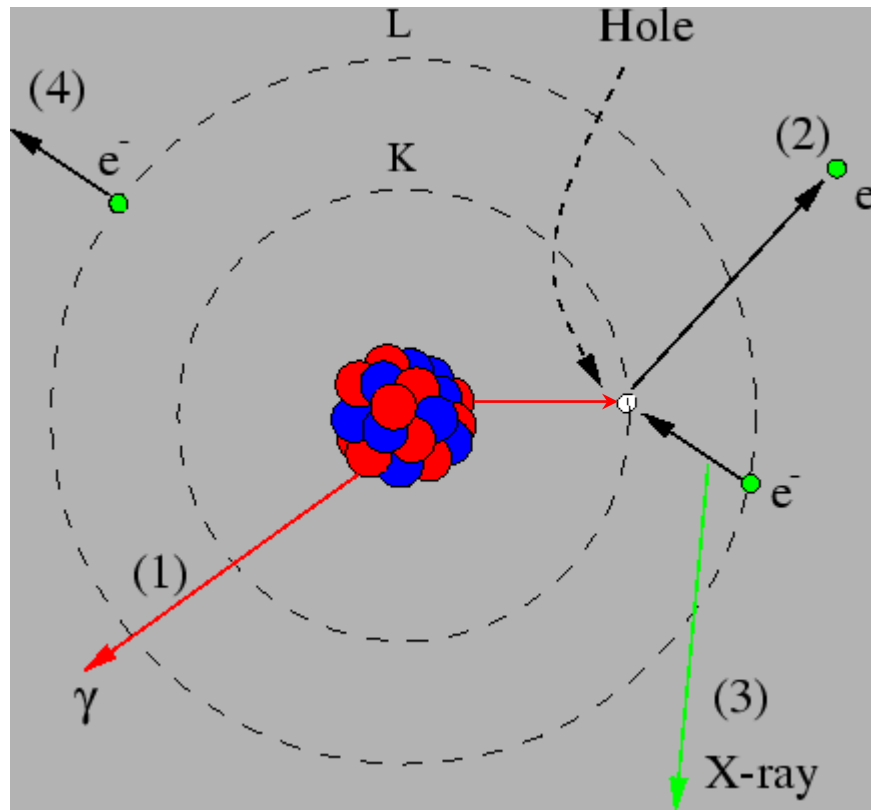


# IT-Like Decay Possibilities

Auger electron emission

$$KE = E_f - E_i - E_{\text{Auger}}$$

Normal *gamma ray* emission  
 $E_{\gamma} = Q$



Internal conversion

$$E_{e^-} = E_{\gamma} - E_b$$

Subsequent *x-ray* emission  
 $(K_{\alpha}, K_{\beta}, L_{\alpha} \dots)$

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# Auger Electron Emission

[http://www.lpdlabservices.co.uk/analytical\\_techniques/surface\\_analysis/aes.php](http://www.lpdlabservices.co.uk/analytical_techniques/surface_analysis/aes.php)

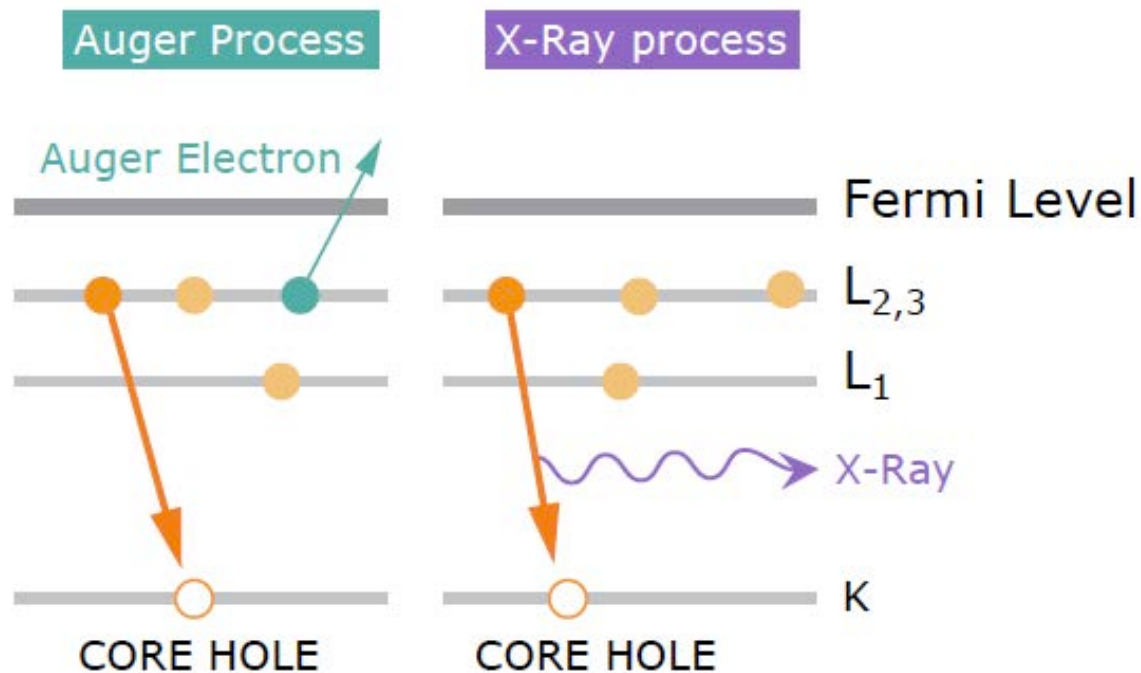
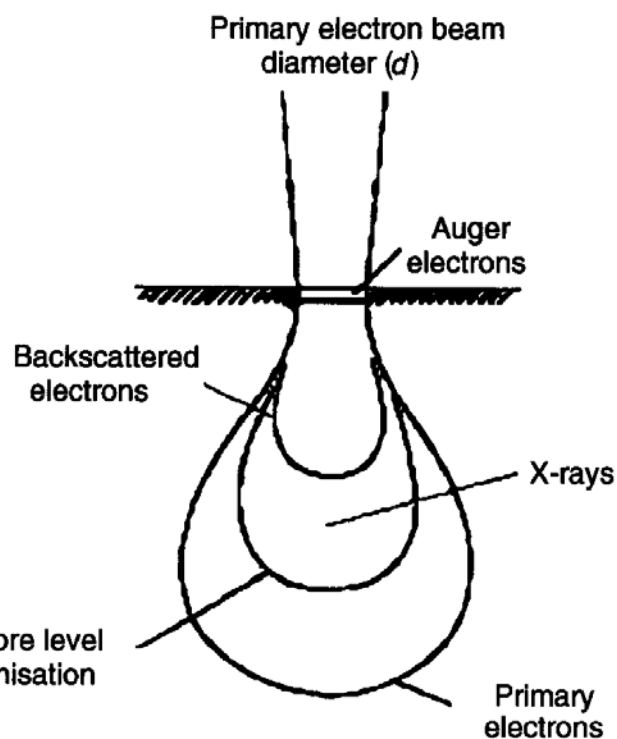
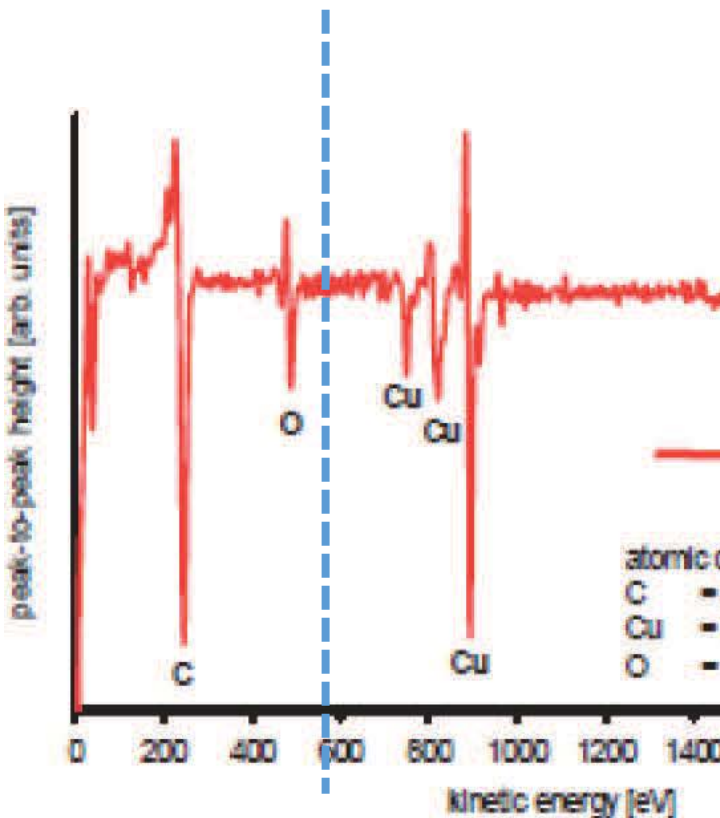


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# Auger Electron Spectroscopy

<https://www.knmf.kit.edu/AES.php>

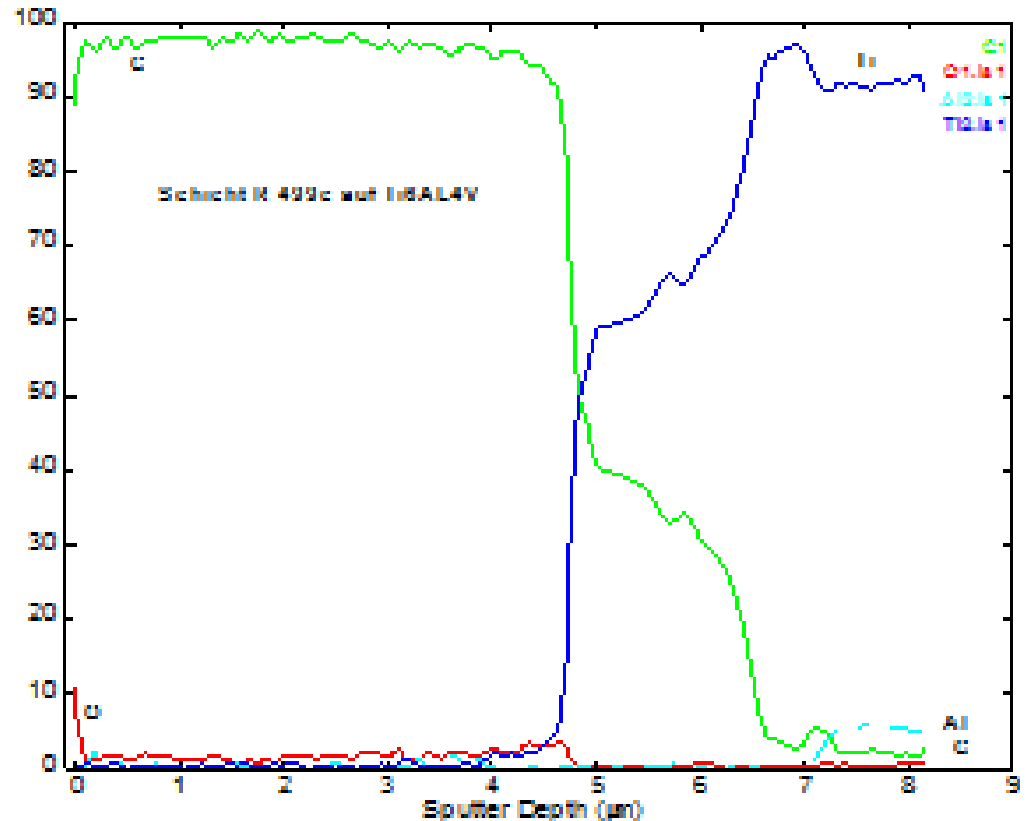
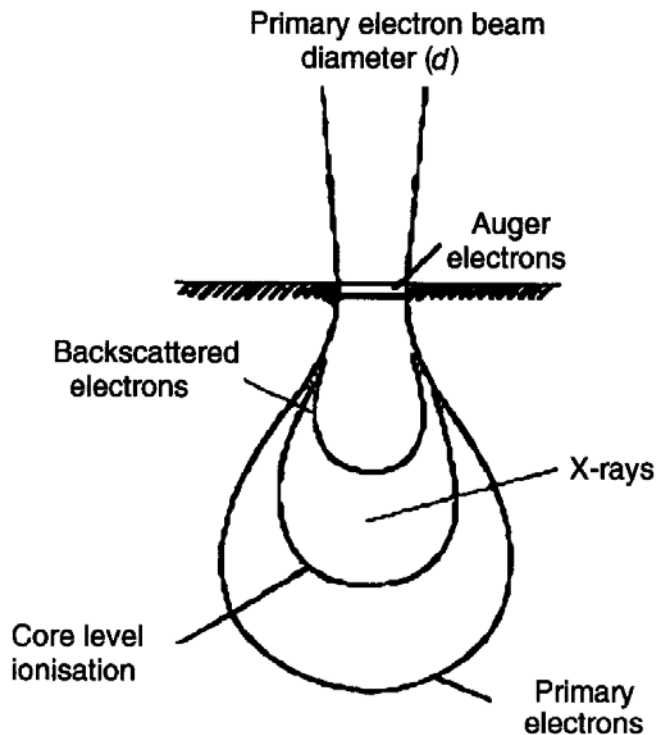


Z	El	1s <sub>1/2</sub> K	2s <sub>1/2</sub> L <sub>1</sub>	2p <sub>1/2</sub> L <sub>2</sub>	2p <sub>3/2</sub> L <sub>3</sub>
1	H	14			
2	He	25			
3	Li	55			
4	Be	111			
5	B	188			5
6	C	284			6
7	N	399			9
8	O	532	24		7
9	F	686	31		9
10	Ne	867	45		18

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# Auger Depth Profiling

<https://www.knmf.kit.edu/AES.php>



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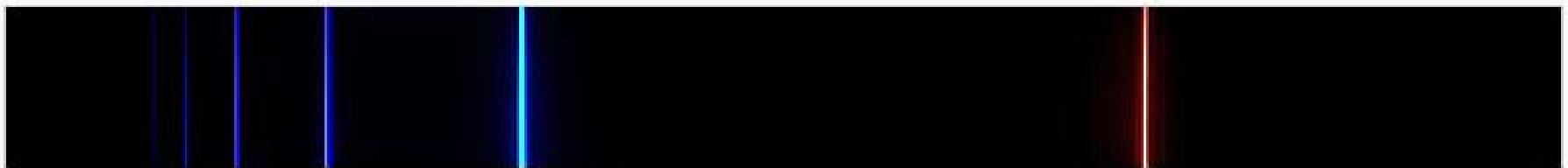
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# Radioactive Decay Summary

Table 3.1 Formulas for Energy Release,  $Q$ , in Terms of Mass Differences,  $\Delta_P$  and  $\Delta_D$ , of Parent and Daughter Atoms

Type of decay	Formula	Reference
$\alpha$	$Q_\alpha = \Delta_P - \Delta_D - \Delta_{\text{He}}$	Eq. (3.13)
$\beta^-$	$Q_{\beta^-} = \Delta_P - \Delta_D$	Eq. (3.25)
$\gamma$	$Q_{\text{IT}} = \Delta_P - \Delta_D$	Eq. (3.30)
EC	$Q_{\text{EC}} = \Delta_P - \Delta_D - E_B$	Eq. (3.35)
$\beta^+$	$Q_{\beta^+} = \Delta_P - \Delta_D - 2mc^2$	Eq. (3.41)

# Photon Emission Lines of Hydrogen

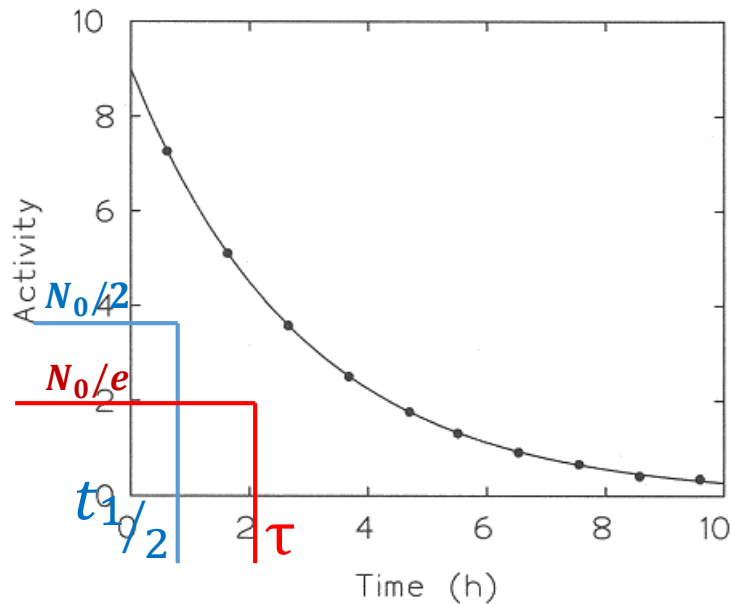


The visible hydrogen emission spectrum lines in the Balmer series. H-alpha is the red line at the right. The two leftmost lines are considered to be ultraviolet as they have wavelengths less than 400 nm.

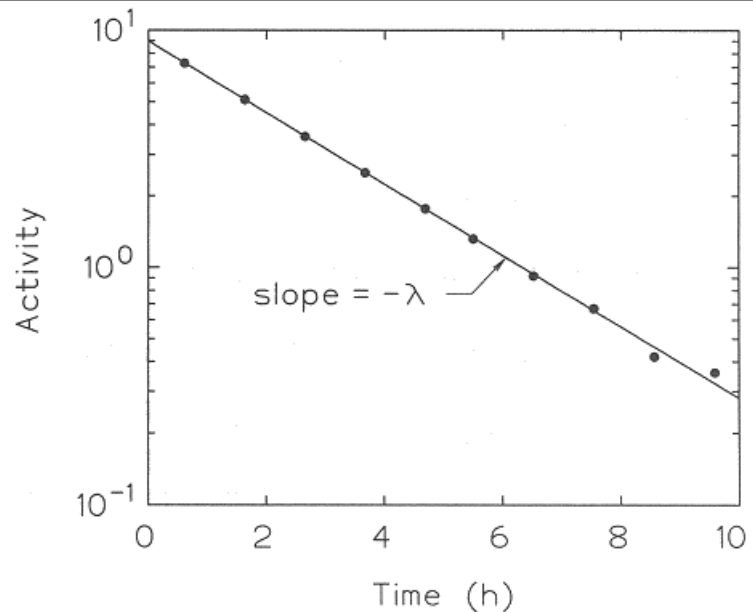
<b>Transition of <math>n</math></b>	3→2	4→2	5→2	6→2	7→2	8→2	9→2	$\infty$ →2
<b>Name</b>	H- $\alpha$	H- $\beta$	H- $\gamma$	H- $\delta$	H- $\epsilon$	H- $\zeta$	H- $\eta$	
<b>Wavelength (nm) <sup>[2]</sup></b>	656.3	486.1	434.1	410.2	397.0	388.9	383.5	364.6
<b>Color</b>	Red	Blue-green	Violet	Violet	(Ultraviolet)	(Ultraviolet)	(Ultraviolet)	(Ultraviolet)

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# Half Life vs. Decay Constant



**Figure 5.15.** The activity of a radioactive sample with a half-life of two hours. At any time on the exponential curve, the activity is one-half of the activity two hours earlier.

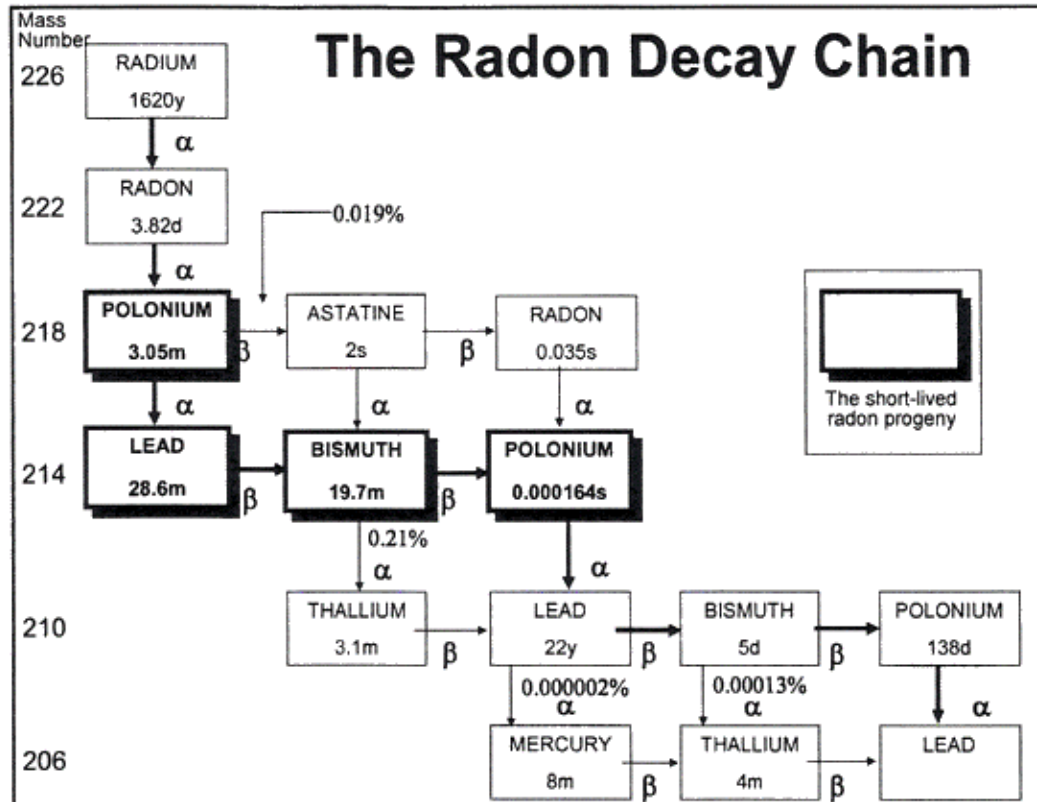


**Figure 5.16.** Semilog plot of the decay of the sample's activity. The decay curve is a straight line with a slope of  $-\lambda$ , from which the half-life  $T_{1/2} = \ln 2/\lambda$  can be calculated.

Source: Yip, Sidney. Nuclear Radiation Interactions, 2014. © World Scientific Publishing Co. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <https://ocw.mit.edu/help/faq-fair-use>.

# The Radon Decay Chain

[http://www.omfi.hu/cejoem/Volume13/Vol13No1/CE07\\_1-01.html](http://www.omfi.hu/cejoem/Volume13/Vol13No1/CE07_1-01.html)



Courtesy of National Academies Press. Used with permission.

Source: National Research Council. Health Effects of Exposure to Radon: BEIR VI. The National Academies Press, 1999. doi:10.17226/5499.



# The Primordial Nuclides

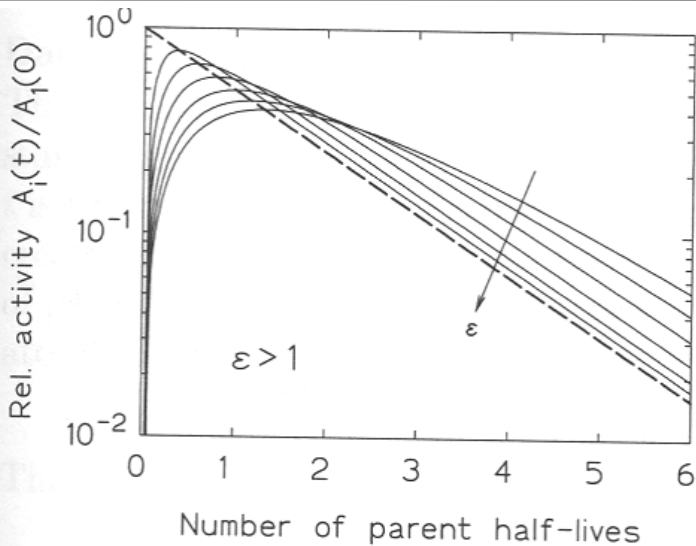
Table 5.2. The 17 isolated primordial radionuclides. Data taken from GE-NE [1996].

Radionuclide & the Decay Modes	Half-life (years)	% El. Abund.	Radionuclide & the Decay Modes	Half-life (years)	% El. Abund.
$^{40}_{19}\text{K}$ $\beta^-$ EC $\beta^+$	$1.27 \times 10^9$	0.0117	$^{50}_{23}\text{V}$ $\beta^-$ EC	$1.4 \times 10^{17}$	0.250
$^{87}_{37}\text{Rb}$ $\beta^-$	$4.88 \times 10^{10}$	27.84	$^{113}_{48}\text{Cd}$ $\beta^-$	$9 \times 10^{15}$	12.22
$^{115}_{49}\text{In}$ $\beta^-$	$4.4 \times 10^{14}$	95.71	$^{123}_{52}\text{Te}$ EC	$> 1.3 \times 10^{13}$	0.908
$^{138}_{57}\text{La}$ EC $\beta^-$	$1.05 \times 10^{11}$	0.090	$^{144}_{60}\text{Nd}$ $\alpha$	$2.38 \times 10^{15}$	23.80
$^{147}_{62}\text{Sm}$ $\alpha$	$1.06 \times 10^{11}$	15.0	$^{148}_{62}\text{Sm}$ $\alpha$	$7 \times 10^{15}$	11.3
$^{152}_{64}\text{Gd}$ $\alpha$	$1.1 \times 10^{14}$	0.20	$^{176}_{71}\text{Lu}$ $\beta^-$	$3.78 \times 10^{10}$	2.59
$^{174}_{72}\text{Hf}$ $\alpha$	$2.0 \times 10^{15}$	0.162	$^{180}_{73}\text{Ta}$ EC $\beta^-$	$> 1.2 \times 10^{15}$	0.012
$^{187}_{75}\text{Re}$ $\beta^-$	$4.3 \times 10^{10}$	62.60	$^{186}_{76}\text{Os}$ $\alpha$	$2 \times 10^{15}$	1.58
$^{190}_{78}\text{Pt}$ $\alpha$	$6.5 \times 10^{11}$	0.01			

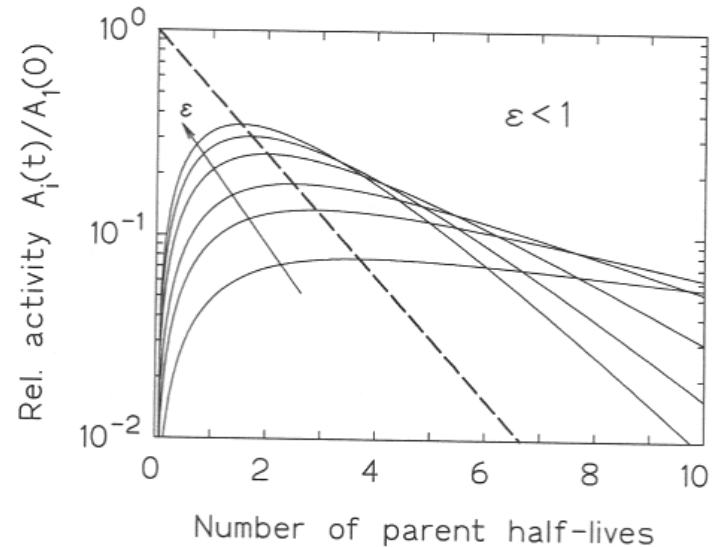
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Source: Shultis, J. K., and R. E. Faw. Fundamentals of Nuclear Science and Engineering, 2nd Edition. CRC Press, 2007.

# Series Decay Chains



**Figure 5.18.** Activity of the first daughter with a half-life less than that of the parent, i.e., the daughter's decay constant  $\lambda_2 = \epsilon\lambda_1$ ,  $\epsilon > 1$ . The six displayed daughter transients are for  $\epsilon = 1.2, 1.5, 2, 3, 5,$  and  $10$ . The heavy-dashed line is the parent's activity.



**Figure 5.19.** Activity of the first daughter with a half-life greater than that of the parent, i.e., the daughter's decay constant  $\lambda_2 = \epsilon\lambda_1$ ,  $\epsilon < 1$ . The six displayed daughter transients are for  $\epsilon = 0.9, 0.7, 0.5, 0.3, 0.2,$  and  $0.1$ . The heavy-dashed line is the parent's activity.

Source: Yip, Sidney. Nuclear Radiation Interactions, 2014. © World Scientific Publishing Co. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <https://ocw.mit.edu/help/faq-fair-use>.

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## 22.01 Introduction to Nuclear Engineering and Ionizing Radiation

Spring 2024

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