

22.01 - Recitation #6

- Please grab a snack, get up off the sofa, look at something that isn't a screen for 5 mins!
- Please turn on your video (if possible) and mute yourself.
- These slides are at:
bit.ly/2201Rec6

Outline + Intended Learning Outcomes (ILOs)

- Everything...
- Start the process of remembering what you learnt 5 weeks ago!

Explaining Terms

- Atomic mass

1 amu
neutron
proton
electron

- Excess mass

What does “excess mass” really mean?

- Binding energy

$$B(A, Z) \equiv$$

Q =

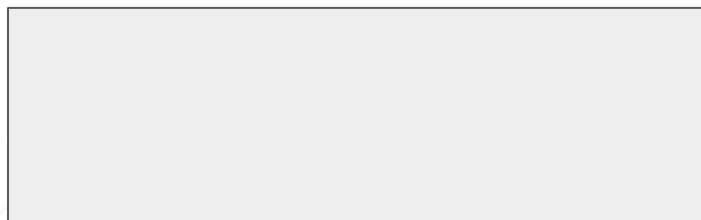
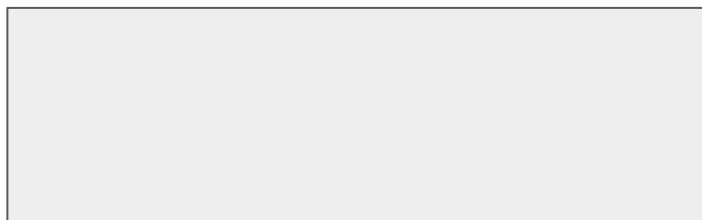
Decay	Individual Particle Reaction	Full Nuclear Reaction
Alpha	α	
Beta -		${}^A_Z X_N \rightarrow {}^A_{Z+1} X'_{N-1} + e^- + \bar{\nu}$
Beta +		${}^A_Z X_N \rightarrow {}^A_{Z-1} X'_{N+1} + e^+ + \nu$
Electron Capture		${}^A_Z X_N + e^- \rightarrow {}^A_{Z-1} X'_{N+1} + \nu$
Gamma (IT)	γ	
Internal Conversion	e^-	
...		N/A

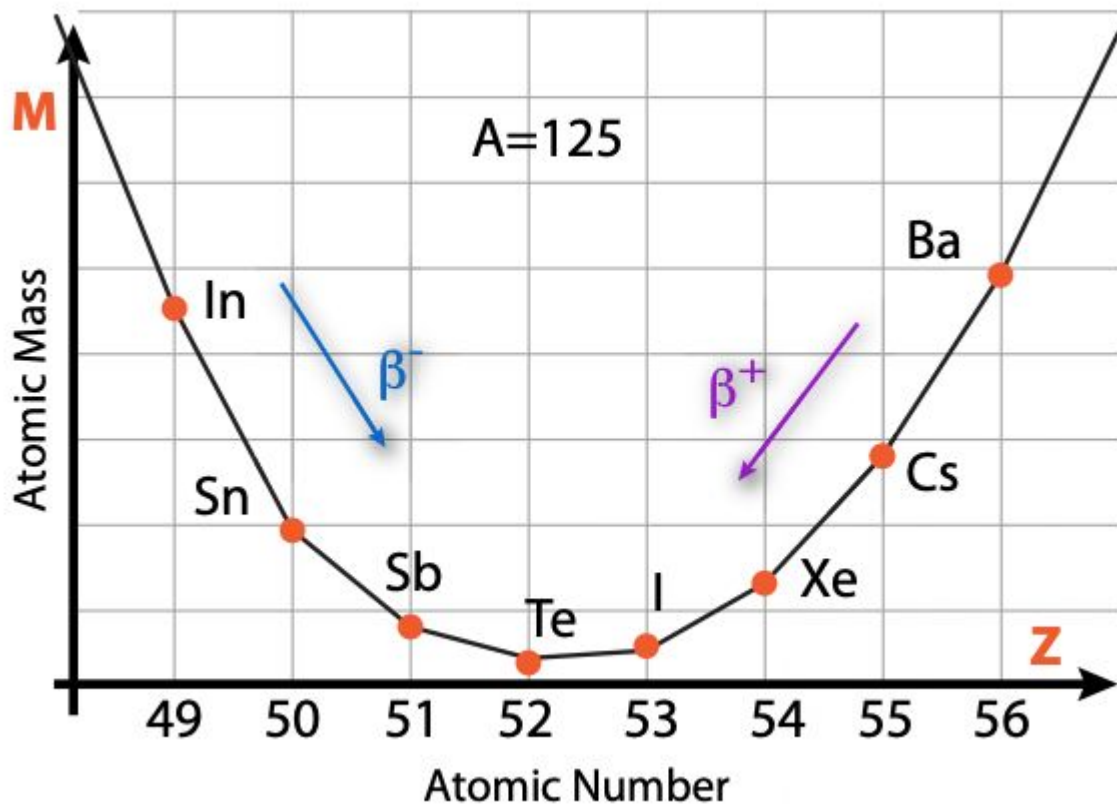
Semi-Empirical Mass Formula

$B(A, Z) =$



(4.10)





Activity

$$A(t) = A_0 e^{-\lambda t}$$

$$t_{1/2} = \ln(2)/\lambda$$

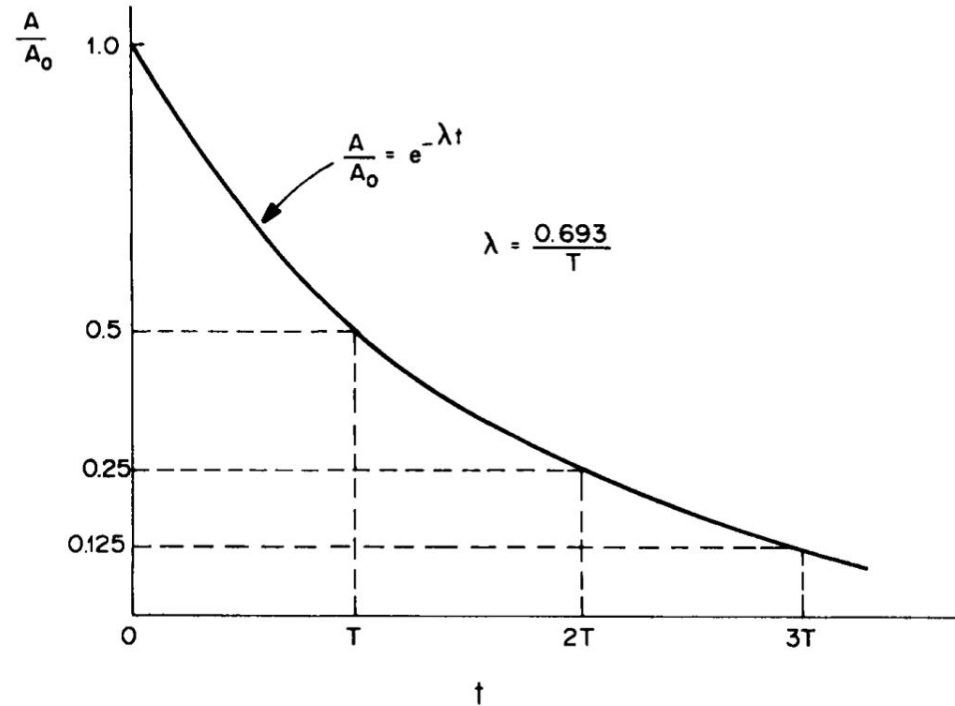


Fig. 4.1 Exponential radioactivity decay law, showing relative activity, A/A_0 , as a function of time t ; λ is the decay constant and T the half-life.

Burning isotopes in a reactor

1kg of Fe-55 is put into a reactor with thermal flux = $1E14$ [n/cm²/s]

- What is the removal rate of Fe-55? [atoms/s]
- How much Fe-55 will there be after 1 year? [kg]

▼ Quiz 1 Resources

 [2201-Study-Sheet.pdf](#)

 [22.01 F16 Quiz 1 Solutions.pdf](#)

 [2201-F17-Quiz-1.pdf](#)

 [2201-F18-Quiz-1-Solutions.pdf](#)

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Review 4-5pm Today

Office Hours 8-8.45am Monday

Office Hours 3-4pm Tuesday

Questions?

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<https://ocw.mit.edu>

22.01 Introduction to Nuclear Engineering and Ionizing Radiation

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