3.020 Lecture 23

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1 Taut rope construction

• Consider lens diagram, e.g. Si-Ge



2 Eutectic reactions

 $\mathbf{L} \qquad \leftrightarrow \qquad \mathbf{Solid} \ \mathbf{1} + \mathbf{Solid} \ \mathbf{2}$

- Eutectic point is a local minimum of the liquidus, at temp T_E Liquidus = locus of points on binary phase diagram above which only liquid phase is stable
- DoF = C Ph + 2 = 1
- Eutectic point on binary phase diagram becomes a line in (T, P, X_2) higher-dimensional sapce
- Eutectoid reactions



• Free energy-composition diagrams for eutectic systems



 $-\ T < T_E$: liquid is unstable, and system has two immiscible solid phases

 $-\ T > T_E$: liquid solution stable over a finite composition range

or + L

1 B+L



• Plotting ΔG_{mix} or G? It doesn't change the resulting taut-rope construction and phase diagram



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