

## Session #24: Homework Problems

### Problem #1

To increase its corrosion resistance, chromium (Cr) is diffused into steel at 980°C. If during diffusion the surface concentration of chromium remains constant at 100%, how long will it take (in days) to achieve a Cr concentration of 1.8% at a depth of 0.002 cm below the steel surface? ( $D_0 = 0.54 \text{ cm}^2/\text{s}$ ;  $E_A = 286 \text{ kJ/mol}$ )

### Problem #2

By planar diffusion of antimony (Sb) into p-type germanium (Ge), a p-n junction is obtained at a depth of  $3 \times 10^{-3} \text{ cm}$  below the surface. What is the donor concentration in the bulk germanium if diffusion is carried out for three hours at 790°C? The surface concentration of antimony is held constant at a value of  $8 \times 10^{18} \text{ cm}^{-3}$ ;  $D_{790^\circ\text{C}} = 4.8 \times 10^{-11} \text{ cm}^2/\text{s}$ .

### Problem #3

You wish to dope a single crystal of silicon (Si) with boron (B). The specification reads  $5 \times 10^{16} \text{ boron atoms/cm}^3$  at a depth of 25  $\mu\text{m}$  from the surface of the silicon. What must be the effective concentration of boron in units of  $\text{atoms/cm}^3$  if you are to meet this specification within a time of 90 minutes? Assume that initially the concentration of boron in the silicon crystal is zero. The diffusion coefficient of boron in silicon has a value of  $7.23 \times 10^{-9} \text{ cm}^2 \text{ s}^{-1}$  at the processing temperature.

### Problem #4

A slab of plate glass containing dissolved helium (He) is placed in a vacuum furnace at a temperature of 400°C to remove the helium from the glass. Before vacuum treatment, the concentration of helium is constant throughout the glass. After 10 minutes in vacuum at 400°C, at what depth from the surface of the glass has the concentration of helium decreased to 1/3 of its initial value? The diffusion coefficient of helium in the plate glass at the processing temperature has a value of  $3.091 \times 10^{-6} \text{ cm}^2/\text{s}$ .

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