In solution to HW 2 Prob. 1.4
Why is there no $3 x$ conversion to alpha-v (coming from dv/dt)p relation) from alpha-l (supplied in problem)?

You are correct, the thermal expansion coefficient that is needed in the final expression is alpha_V not alpha_L.

You can find alpha_V=3*alpha_L...
For a more detailed derivation of this relationship, please go to
Thermal Expansion

I will correct the solutions later...

