3.40 LECTURE SUMMARY - Nov 9th

11/ 16/ 2009

INTRODUCTION

Precipitation Hardening

 It relies on changes in solid solubility with temperature to produce a second phase

Hardening Mechanisms

- Modulus Mismatch
 - Changes the dislocation energy
- Ordered Precipitates
- Size Effect
 - Cutting to Bowing Transition
 - Coherent and Incoherent Precipitates

2 COMPONENT PHASE DIAGRAMS



Fig. 1: System forming 2 Solid Solutions [http://en.wikipedia.org/wiki/Phase_diagram] Fig. 2: System forming an Intermetallic [http://www.substech.com/]

ANTI-PHASE BOUNDARIES

Fig. 3: Dislocation cutting through an ordered precipitate forming an APB

• Additional energy is required to create an APB

$$\Delta \tau_{APB} = \frac{\pi}{2b} \gamma_{APB} \frac{r}{L} = \frac{\pi}{2b} \gamma_{APB} V_f$$

PRECIPITATION HARDENING

- Contributions to Precipitation Hardening
 - Anti-Phase Boundaries
 - Interfaces
 - Coherency Strains
 - Modulus Effect

Decreasing effect on Strength

$$\Delta \tau_{tot} = \Delta \tau_{APB} + \Delta \tau_{int} + \Delta \tau_{coh} + \Delta \tau_{mod}$$

Fig. 4: Schematic of a sheared precipitate

APB

Interface

SIZE EFFECTS

- As the size of the second phase particle increases
 - Cutting through it becomes more difficult

 $\Delta \tau_{cutting} \propto r$ $\text{ Dislocations tend to bow} \\ \text{ around the particle by} \\ \text{ Orowan Looping} \\ \Delta \tau_{bowing} \propto \frac{1}{r}$



Critical radius ~ 5 - 30 nm Fig. 5: Shear Strength vs. Particle Radius [http://en.wikipedia.org/wiki/Precipitation_strengthening]

SIZE EFFECTS

- As the size increases, coherent particles transform to incoherent precipitates
- The strain energy is reduced and the strengthening effect is weakened
- The critical radius for this transition ~ 10 nm 1 μ m



Fig. 6: Different types of interfaces [http://www.doitpoms.ac.uk/tlplib/solid-solutions/]

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PRECIPITATE FREE ZONES

- Heterogeneous nucleation at grain boundaries, dislocations
- Particles nucleate at grain boundaries
- Regions surrounding the grain boundary are depleted of solutes, forming PFZs



Fig. 7: PFZ near grain boundaries in Al -4 wt% Cu [http://www.msm.cam.ac.uk/phasetrans/abstracts/pfz. html]

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SUMMARY

- Precipitation Hardening provides a greater increase in strength than solid solution strengthening
- Intermetallics are better obstacles to dislocation motion due to the creation of Anti-Phase Boundaries
- 2 critical radii exist:
 - Cutting to bowing: $r_c \sim 10 \text{ nm}$
 - Coherent to Incoherent: $r_{c/i} \sim 10 \text{ nm} 1 \mu \text{m}$

QUESTIONS?

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