#### 5.61 Fall 2017 Problem Set #3

**1.** A. McQuarrie, page 120, #3-3 Show  $\hat{A}f(x) = \lambda f(x)$ , for  $\lambda$  constant. Find the eigenvalue  $\lambda$ .

**B.** McQuarrie, page 120, #3-4

**C.** McQuarrie, page 182, #4-11

## 2. McQuarrie, pages 121-122, #3-11. Continuity of $\psi'$

**3.** A. McQuarrie, page 123, #3-17

**B.** McQuarrie, page 127, #3-36

### 4. Particle in an infinite 1-D Well

A. McQuarrie, page 122, #3-12. Answer this problem qualitatively by drawing a cartoon for n = 2 and n = 3 states.

**B.** Is there a simple mathematical/physical reason why the probabilities are not 1/4 for all four regions:  $0 \le x \le a/4$ ,  $a/4 \le x \le a/2$ ,  $a/2 \le x \le 3a/4$ , and  $3a/4 \le x \le a$ ?

**[HINT**: where are the nodes in  $\psi_n(x)$ ?]

#### 5. Particle on a Ring

Solve for the energy levels of the particle confined to a ring as a crude model for the electronic structure of benzene. The two dimensional Schrödinger Equation, in polar coordinates, is

$$-\frac{\hbar^2}{2\mu} \left[ \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2}{\partial \phi^2} + U(r,\phi) \right] \psi = E \psi.$$

For this problem,  $U(r, \phi) = \infty$  for  $r \neq a$ , but when r = a,  $U(a, \phi) = 0$ .

**A.** This implies that  $\psi(r, \phi) = 0$  for  $r \neq a$ . Why?

**B.** If  $\psi(r, \phi) = 0$  for  $r \neq a$ , then  $\frac{\partial \psi}{\partial r} = 0$ . What is the simplified form of the Schrödinger Equation that applies when the particle is confined to the ring?

**C.** Apply the "periodic" boundary condition that  $\psi(a, \phi) = \psi(a, \phi + 2\pi)$  to obtain the  $E_n$  energy levels.

# 6. 1-Dimensional Infinite Wells with Steps

Consider the potential

$$V(x) = \infty \qquad x < 0, x > a$$
  

$$V(x) = 0 \qquad 0 \le x \le a/2$$
  

$$V(x) = V_0 = \frac{h^2}{8ma^2} (2)^2 \qquad a/2 < x \le a$$

(This is the energy of n = 1 of an infinite well of width a/2.)

А.

Sketch V(x) vs. x.

**B.** What are the boundary conditions for  $\psi(x)$  at x = 0 and x = a?

C. What requirements must be satisfied at x = a/2?

**D.** Solve for the n = 2 (one node) and n = 3 (two nodes)  $\psi_n(x)$  eigenfunctions of  $\hat{H}$  and  $E_n$  energy levels.

Hints:

- (i) For  $0 \le x \le a/2$ ,  $\psi_{\rm I}(x) = A \sin k_{\rm I} x$   $k_{\rm I} = [2mE/\hbar^2]^{1/2}$
- (ii) For  $a/2 < x \le a$ ,  $\psi_{\text{II}} = B \sin k_{\text{II}} (a x) k_{\text{II}} = [2m(E V_0)/\hbar^2]^{1/2}$

(iii) 
$$\psi_{\mathrm{I}}(a/2) = A \sin(k_{\mathrm{I}} a/2)$$
  
 $\psi_{\mathrm{II}}(a/2) = B \sin(k_{\mathrm{II}} a/2)$   
 $\frac{d\psi_{\mathrm{I}}}{dx}_{x=a/2} = Ak_{\mathrm{I}} \cos(k_{\mathrm{I}} a/2)$   
 $\frac{d\psi_{\mathrm{II}}}{dx}_{x=a/2} = +Bk_{\mathrm{II}} \cos(k_{\mathrm{II}} a/2)$ 

**E.** Compare your values of  $E_2$  and  $E_3$  to what you obtain from the de Broglie quantization condition

$$(n/2) = \frac{a/2}{\lambda_{n,\mathrm{I}}} + \frac{a/2}{\lambda_{n,\mathrm{II}}}$$
  
 $\lambda = h/p = 2\pi/k = h[2m(E - V(x))]^{-1/2}$ 

**F.** For the n = 2 and n = 3 energy levels, what are the probabilities,  $P_2$  and  $P_3$ , of finding the particle in the  $0 \le x \le a/2$  region?

**G.** (optional) Will the n = 2 and n = 3 energy levels of the  $V_1(x)$  and  $V_2(x)$  potentials (defined below) be identical, as suggested by part **E**? Why?

$$\begin{array}{cccc} V_1(x): & V_1(x) = \infty & x < 0, x > a \\ & V_1(x) = 0 & 0 \le x \le a/2 \\ & V_1(x) = V_0 & a/2 < x \le a \end{array} \end{array} \text{ barrier on right side }$$

versus

$$\begin{array}{cccc} V_2(x): & V_2(x) = \infty & x < 0, x > a \\ & V_2(x) = 0 & 0 \le x \le a/4, 3a/4 \le x \le a \\ & V_2(x) = V_0 & a/4 < x \le 3a/4 \end{array}$$
 barrier in the center

**H.** Solve for  $n = 1 \ \psi_1(x)$  and  $E_1$  for  $V_1$ . HINTS: For  $a/2 < x \le a$ ,

$$\psi_{\rm II}(x) = Be^{\kappa_{\rm II}(a-x)} + Ce^{-\kappa_{\rm II}(a-x)}$$
$$\kappa_{\rm II} = [2m(V_0 - E)/\hbar^2]^{1/2}$$

**I.** (optional) Is  $E_1$  for  $V_1$  larger or smaller than  $E_1$  for  $V_2$ ? Why? A cartoon would be helpful.

MIT OpenCourseWare <u>https://ocw.mit.edu/</u>

5.61 Physical Chemistry Fall 2017

For information about citing these materials or our Terms of Use, visit: <u>https://ocw.mit.edu/terms</u>.