Mass Transport in liquids

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Outline

> Chemical potential

> Species conservation including convection

- > H-filter design & eigenfunction expansion
- > Taylor dispersion, the microfluidicist's enemy
- > Mixing

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Chemical potential

- It comes from thermodynamics
- > Chemical potential gradients are the driving force for the movement of molecules
- It is the electron Fermi level in semiconductors
- > At equilibrium, there are no gradients in μ

$$\mu = \left(\frac{\partial W}{\partial N}\right)_{T,V}$$

For an ideal solution:

$$\mu_i(x) = \mu_i^0 + k_B T \ln \frac{c_i(x)}{c_i^0}$$

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Chemical potential

- > We can derive Fick's first law from the chemical potential
- > First, note that there are two concentration units
- > Relate flux to velocity
- Then relate the velocity to a force *f*, using a mobility *M*
- Then the force to a potential (\$\vert\$) gradient

$$\begin{bmatrix} \mathbf{m}^{2}/\mathbf{V} \cdot \mathbf{s} \end{bmatrix} \begin{bmatrix} \mathbf{s}/\mathbf{kg} \end{bmatrix}$$

$$U = \mu_{n}E = \frac{\mu_{n}}{q_{e}} \begin{pmatrix} q_{e}E \end{pmatrix} = -\frac{\mu_{n}}{q_{e}} (\nabla q_{e}\phi)$$

$$\begin{pmatrix} \uparrow & \uparrow & q_{e} \end{pmatrix} \begin{bmatrix} \mathbf{v}/\mathbf{m} \end{bmatrix} \begin{bmatrix} \mathbf{N} \end{bmatrix}$$

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JV: 2.372J/6.777J Spring 2007, Lecture 16 - 4

 $c_i = N_A C_i$

 $\left[\frac{\#}{m^3}\right] = \left[\frac{\#}{mol}\right] \left[\frac{mol}{m^3}\right]$

 $J_i = c_i U_i = N_A C_i U_i$

 $U_i = Mf = -M \frac{\partial \mathcal{P}}{\partial r}$

Chemical potential

> Finally, find flux due to a chemical potential gradient

$$\mu_{i}(x) = \mu_{i}^{0} + k_{B}T \ln \frac{c_{i}(x)}{c_{i}^{0}}$$

> Can relate diffusivity to mobility

 $k_{B}T = \frac{D}{M}$

Einstein Relation

$$J_{i} = -c_{i}M\frac{\partial\mu_{i}}{\partial x} = -c_{i}Mk_{B}T\frac{\partial}{\partial x}\left(\ln\frac{c_{i}(x)}{c_{i}^{0}}\right)$$
$$J_{i} = -c_{i}Mk_{B}T\frac{\partial}{\partial x}\left(\ln c_{i}(x) - \ln c_{i}^{0}\right)$$
$$J_{i} = -Mk_{B}Tc_{i}\frac{\partial}{\partial x}\left(\ln c_{i}(x)\right)$$
$$J_{i} = -Mk_{B}Tc_{i}\frac{1}{c_{i}}\frac{\partial c_{i}}{\partial x}$$
$$J_{i} = -Mk_{B}T\frac{\partial c_{i}}{\partial x} = -D\frac{\partial c_{i}}{\partial x}$$

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Species conservation equation

- > One more conservation equation...
- > Flux now includes convection and diffusion
- Incompressible flow

$$\frac{\frac{d}{dt}\int b dV = -\int \mathbf{F} \cdot \mathbf{n} dS + \int g dV}{\frac{\partial b}{\partial t}} = -\nabla \cdot \mathbf{F} + g$$

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$$\frac{\partial c_i}{\partial t} = -\nabla \cdot \mathbf{J}_i + R_{Vi}$$

convection $\mathbf{J}_i = -D_i \nabla c_i + c_i \mathbf{U}_i$ diffusion

$$\frac{\partial c_{i}}{\partial t} = -\nabla \cdot \mathbf{J}_{i} + R_{Vi}$$

$$\frac{\partial c_{i}}{\partial t} = D_{i} \nabla^{2} c_{i} - c_{i} \nabla \cdot \mathbf{U}_{i} - \mathbf{U}_{i} \cdot \nabla c_{i} + R_{Vi}$$

$$\frac{\partial c_{i}}{\partial t} = D_{i} \nabla^{2} c_{i} - C_{i} \nabla \cdot \mathbf{U}_{i} - \mathbf{U}_{i} \cdot \nabla c_{i} + R_{Vi}$$

$$\frac{\partial c_{i}}{\partial t} + \mathbf{U}_{i} \cdot \nabla c_{i} = D_{i} \nabla^{2} c_{i} + R_{Vi}$$

$$\frac{\partial c_{i}}{\partial t} = -\nabla \cdot \left(-D_{i} \nabla c_{i} + c_{i} \mathbf{U}_{i} \right) + R_{Vi}$$

$$\frac{\partial c_{i}}{\partial t} = D_{i} \nabla^{2} c_{i} + R_{Vi}$$

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Convective term

- > We have seen this equation before
- > We can compare the convective to diffusive flux terms, and get a Peclet number again
 - Now for diffusive vs. convective mass transport
- For BSA (66 kDA) in microscale flows, L~100 μm, U~1 mm/s, D~7x10⁻¹¹ m²/s
- > Convection is important because molecular diffusivity is 10⁷ times slower than heat diffusivity and 10⁵ times slower than momentum diffusivity

$$\frac{\partial c_i}{\partial t} + \mathbf{U}_i \cdot \nabla c_i = D_i \nabla^2 c_i + R_{Vi}$$

$$\frac{convection}{diffusion} \sim \frac{\mathbf{U}_{i} \cdot \nabla c_{i}}{D_{i} \nabla^{2} c_{i}} \sim \frac{U c/L}{D c/L^{2}} \sim \frac{LU}{D}$$

$$Pe = \frac{LU}{D} = \frac{(10^{-4} m)(10^{-3} m/s)}{7 \cdot 10^{-11} m^2/s} \sim 10^3$$

$$D_{heat}$$
 ~10⁻⁴ m²/s for water
 $D_{momentum}$ ~10⁻⁶ m²/s for water

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Diffusivities

- > How can we get diffusivities for different objects?
- > Use mobility due to Stokes drag
- > Result is Stokes-Einstein relation
- > Larger particles have smaller diffusivity
- > Often used to get an effective radius (R_h) for a species

$$D = Mk_{B}T = \frac{U_{i}}{f}k_{B}T$$
$$f = 6\pi\eta RU_{i} \Rightarrow \frac{U_{i}}{f} = \frac{1}{6\pi\eta R}$$
$$D = Mk_{B}T = \frac{k_{B}T}{6\pi\eta R}$$

R=45 nm R_h=44.8 nm

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- > What are the minimum diffusivity differences that we can separate?
- > How to choose channel width, length, flowrate



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Courtesy of Paul Yager, Thayne Edwards, Elain Fu, Kristen Helton, Kjell Nelson, Milton R. Tam, and Bernhard H. Weigl. Used with permission. Please see:

Yager, P., T. Edwards, E. Fu, K. Helton, K. Nelson, M. R. Tam, and B. H. Weigl. "Microfluidic Diagnostic Technologies for Global Public Health." *Nature* 442 (July 27, 2006): 412-418.

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- > First, let's try a quick and dirty diffusion calculation
- > Assume 1-D diffusion across width of channel
- > Ignore convection effects along length of channel
- > No generation terms
- > Result suggests that separation will go as \sqrt{D}



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- > Can we do better?
- > Yes, using eigenfunction analysis

> Assumptions

- Ignore convection
- No generation
- No concentration gradients along channel height or length
 - » 1-D diffusion
- One dilute component in solvent



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Initial condition

- Solute initially fills part of channel
- > Boundary condition
 - No solute flux through walls





Initial condition:

$$c(x,0) = \begin{cases} c_0 & \text{for } 0 < x < d \\ 0 & \text{for } d < x < W \end{cases}$$

Boundary condition:

$$\left. \frac{\partial c}{\partial x} \right|_{x=0 W} = 0 \text{ for all } t$$

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> First, separate variables



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> Sine does not meet BCs



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> Finally, use eigenfunction expansion to meet initial concentration profile ____

$$c(x,t) = a_0 + \sum_{n=1}^{\infty} B_n \sqrt{\frac{2}{W}} \cos k_n x \cdot e^{-\alpha_n t}$$

$$\int \mathbf{t} = \mathbf{0}$$

$$c(x,0) = a_0 + \sum_{n=1}^{\infty} B_n \sqrt{\frac{2}{W}} \cos k_n x = \begin{cases} c_0 & \text{for } 0 < x < d \\ 0 & \text{for } d < x < W \end{cases}$$

multiply both sides by eigenfctn & integrate

$$\int_{0}^{W} c(x,0)\sqrt{\frac{2}{W}}\cos k_{m}xdx = \int_{0}^{W} a_{0}\sqrt{\frac{2}{W}}\cos k_{m}xdx + \sum_{n=1}^{\infty}\int_{0}^{W} B_{n}\sqrt{\frac{2}{W}}\cos k_{n}x\sqrt{\frac{2}{W}}\cos k_{m}xdx$$

extract coefficient
$$B_{n} = \int_{0}^{W} c(x,0)\sqrt{\frac{2}{W}}\cos(k_{n}x)dx$$

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> Get coefficients and DC term

$$B_{n} = \int_{0}^{W} c(x,0) \sqrt{\frac{2}{W}} \cos(k_{n}x) dx$$

$$B_{n} = \sqrt{\frac{2}{W}} \left[\int_{0}^{d} c_{0} \cos(k_{n}x) dx + \int_{d}^{W} 0 \cos(k_{n}x) dx \right]$$

$$B_{n} = \sqrt{\frac{2}{W}} \frac{c_{0}}{k_{n}} \sin(k_{n}x) \Big|_{0}^{d}$$

$$B_{n} = \sqrt{\frac{2}{W}} \frac{c_{0}}{k_{n}} \sin(k_{n}d) \quad \text{for } n = 1, 2, 3, ...$$

$$a_{0} = \frac{1}{W} \int_{0}^{W} c(x,0) dx = \frac{c_{0}d}{W}$$

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- > Can plot time evolution for d=W/2
- > Lowest-order mode (n=1) is dominant





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- > Only focus on 1st mode
 - Simplifies math
 - Is dominant mode
- > First mode has error at t=0
 - Need other terms to meet I.C.

$$c(x,t) \approx \frac{2c_0}{\pi} \cos\left(\frac{\pi x}{W}\right) \cdot e^{-\left(\frac{\pi}{W}\right)^2 Dt} + \frac{c_0}{2}$$

$$c_{out} = \frac{2}{W} \int_{W/2}^{W} \left(\frac{2c_0}{\pi} \cos\left(\frac{\pi x}{W}\right) \cdot e^{-\left(\frac{\pi}{W}\right)^2 Dt} + \frac{c_0}{2}\right) dx$$

$$c_{out} = \frac{2}{W} \left[\frac{2c_0 W}{\pi^2} \sin\left(\frac{\pi x}{W}\right)\Big|_{W/2}^{W} \cdot e^{-\left(\frac{\pi}{W}\right)^2 Dt} + \frac{c_0 W}{4}\right]$$

$$c_{out} = \frac{2}{W} \left[\frac{-2c_0 W}{\pi^2} \cdot e^{-\left(\frac{\pi}{W}\right)^2 Dt} + \frac{c_0 W}{4}\right]$$

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- > Can also look at all modes at short time
- > Result is that increases as √Dt for short times

$$c(x,t) = \sum_{n=1}^{\infty} \frac{2c_0}{n\pi} \sin\left(\frac{n\pi}{2}\right) \cos\left(\frac{n\pi x}{W}\right) \cdot e^{-\alpha_n t} + \frac{c_0}{2}$$

↓Take average over output channel

$$c_{out} = \frac{c_0}{2} - \frac{4c_0}{\pi^2} \sum_{n \text{ odd}} \frac{1}{n^2} e^{-\alpha_n t}$$

$$\downarrow ?$$

$$c_{out} \approx 1.1 \sqrt{\frac{Dt}{W^2}}$$

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- Since c_{out} scales with both D and t, c_{out,1}/c_{out,2} will be independent of time at short times
- If D₁>>D₂, then increasing time and decreasing W helps



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Taylor dispersion

- > Was ignoring convection OK?
 - Not really
- > One can solve the 1-D convection-diffusion problem
- > This is called Taylor Dispersion
 - Axial convection + transverse diffusion
- > The result is that the plug spreads out faster than from simple diffusion
- > The apparent diffusivity is *K*
- > EOF does NOT suffer from Taylor dispersion
 - Uniform flow field

$$K_i = D_i + \frac{U^2 h^2}{210D_i} = D_i \left(1 + \frac{Pe^2}{210}\right)$$

Parallel-plate flow channel

$$K_i = D_i + \frac{U^2 h^2}{210D_i} f\left(\frac{h}{W}\right)$$

Rectangular flow channel

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Taylor dispersion

- Can determine K_i for rectangular channels
- > As h/W→0, f(h/W)→ ~7.95NOT 1
 - Because of 2-D profile at wall
- > This implies that for a given h, bigger h/W is better → area small
- > But this means a smaller channel cross-section and higher U, therefore possibly more dispersion



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Convection, diffusion, and mixing

- > We can use convection for good as well as evil
- > At steady state, fluid mixing time turns into distance
- > Short distances from inlet, two fluids appear not to mix

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- > Mixing is driven by diffusion
- Macroscale mixing uses turbulence (e.g., stirring) to reduce length for diffusive mixing
- In liquid microfluidics, there is no turbulence to decrease mixing lengths

THEREFORE,

- > Microfluidic mixing is EASY
- > Microfluidic mixing is HARD
- > Mixing length scales with Pe

 $L \sim U \frac{W^2}{D} \sim Pe \cdot W \qquad \tau \approx 2.5 \text{ s for a 50 } \mu \text{m channel} (D = 10^{-5} \text{ cm}^2/\text{s})$ $\tau \approx 40 \text{ s for a 200 } \mu \text{m channel} (D = 10^{-5} \text{ cm}^2/\text{s})$

JV: 2.372J/6.777J Spring 2007, Lecture 16 - 29

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- > How does one define mixing?
- > No universal definition
- > One definition:
 - When concentration profile is uniform to within 1% (or 5%)
- For our rectangular channel, concentration difference is biggest between x=0 and x=W

$$\Delta c_{\max} = c(0,t) - c(W,t)$$

$$= \sum_{n=1}^{\infty} \frac{2c_0}{n\pi} \sin\left(\frac{n\pi d}{W}\right) \left[1 - (-1)^n\right] \cdot e^{-\alpha_n t}$$

$$= \sum_{n \text{ odd}} \frac{4c_0}{n\pi} \sin\left(\frac{n\pi d}{W}\right) \cdot e^{-\alpha_n t}$$

$$\Delta c_{\max} \approx \frac{4c_0}{\pi} \sin\left(\frac{\pi d}{W}\right) \cdot e^{-\left(\frac{\pi}{W}\right)^2 Dt}$$

$$\frac{\Delta c_{\max}}{c_0} = 0.01$$

$$c(x,t) = \sum_{n=1}^{\infty} \frac{2c_0}{n\pi} \sin\left(\frac{n\pi d}{W}\right) \cos\left(\frac{n\pi x}{W}\right) \cdot e^{-\alpha_n t} + \frac{c_0 d}{W} \qquad T_{mix} = \left(\frac{W^2}{D}\right) \frac{1}{\pi^2} \ln\left[\frac{400}{\pi} \frac{W}{d} \sin\left(\frac{\pi d}{W}\right)\right]$$
$$\alpha_n = \left(\frac{n\pi}{W}\right)^2 D$$

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> Mixing time scales as expected for semi-infinite diffusion



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- > At the microscale various approaches exist for reducing diffusion lengths
 - Depends on how fast you need to mix
- > Approaches trade off fabrication complexity, generality, mixing time, etc.
- > All find ways to laminate two fluids



Figure 2 on p. 249 in Miyake, R., T. S. J. Lammerink, M. Elwenspoek, and J. H. J. Fluitman. "Micro-Mixer with Fast Diffusion." In *Micro Electro Mechanical Systems, 1993, MEMS '93: An Investigation of Micro Structures, Sensors, Actuators, Machines and Systems, February 7-10, 1993.* New York, NY: Institute of Electrical and Electronics Engineers, 1993, pp. 248-253. ISBN: 9780780309579. © 1993 IEEE.

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- > 3-D split and recombine lamination
- > Complicated fab
- > Typical of early designs that focused on Si



Figure 1 on p. 442 in Branebjerg, J., P. Gravesen, J. P. Krog, and C. R. Nielsen, C.R. "Fast Mixing by Lamination." In *Micro Electro Mechanical Systems, 1996, MEMS '96: An Investigation of Micro Structures, Sensors, Actuators, Machines and Systems, February 11-15, 1996.* New York, NY: Institute of Electrical and Electronics Engineers, 1996, pp. 441-446. ISBN: 9780780329850. © 1996 IEEE.

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Laminate in one level of channels by moving complexity from fab to packaging

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Figures 2 and 4 on pp. 266-267 in Jackman, R. J., T. M. Floyd, R. Ghodssi, M. A. Schmidt, and K. F. Jensen. "Microfluidic Systems with On-line UV Detection Fabricated in Photodefinable Epoxy." *Journal of Micromechanics and Microengineering* 11, no. 3 (May 2001): 263-269.

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Passive chaotic micromixer

- > Fairly simple to make
- > Uses simple pressuredriven flow
- > Anisotropic boundary induces anisotropic flow
- > Stroock et al., Science 295(2002):647



Image by MIT OpenCourseWare. Adapted from Figure 2 on p. 648 in Stroock, A. D., S. K. W. Dertinger, A. Ajdari, I. Mezic, H. A. Stone, and G. M. Whitesides. "Chaotic Mixer for Microchannels." *Science* New Series, 295, no. 5555 (January 25, 2002): 647-651.

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Passive chaotic micromixer

- > Mixing length scales with ln(Pe)
 - Rather than Pe in pure diffusive mixing





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More info

- > Microfluidic flow
 - Viscous fluid flow, F. White
 - Low Reynolds Number Hydrodynamic, Happel & Brenner
 - Gravesen et al., "Microfluidics, A Review", JMME 3(1993) 168
 - » Includes lumped resistances for turns, constrictions, etc.
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