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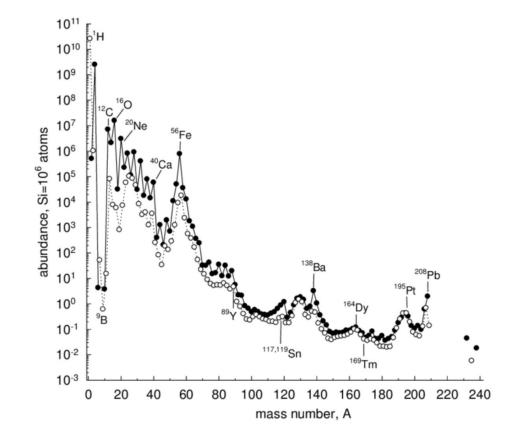
Introduction to Nuclear and Particle Physics

Markus Klute - MIT

9. Nuclear Physics

9.2 Binding Energy

Nuclear Abundance in the Solar System



Binding Energy

_ _ _

$$B(Z, A) = [Z(M_{p}+m_{e}) + (A-Z)M_{N} - M(A,Z)] c^{2}$$

$$M_{p} = 938.272 \text{ MeV/c}^{2} = 1836.149 m_{e}$$

$$M_{N} = 939.566 \text{ MeV/c}^{2} = 1838.679 m_{e}$$

$$m_{e} = 0.511 \text{ MeV/c}^{2}$$
Apart from the lightest elements, binding energy per nuclean is about 7.5-9 MeV

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Parametrization of Binding Energies

Weizäcker (semi-empirical mass) formula and liquid-drop model

$$M(A,Z) = NM_n + ZM_p + Zm_e - a_VA + a_sA^{2/3} + a_c\frac{Z^2}{A^{1/3}} + a_a\frac{(N-Z)^2}{4A} + \frac{\delta}{A^{1/2}}$$

with $N = A - Z$.

 $a_V = 15.67 \text{ MeV/c}^2$, $a_s = 17.23 \text{ MeV/c}^2$, $a_c = 0.714 \text{ MeV/c}^2$, $a_a = 93.15 \text{ MeV/c}^2$ and $\delta = -11.2, 0, +11.2 \text{ MeV/c}^2$ for even Z and Z, odd A, or odd Z and N, respectively.

Volume term: $a_V A$

Dominates binding energy, proportional to the number of nucleon which contributes with about 16 MeV each.

Nuclear force must be short range, corresponding roughly to the distance between two nucleons.

As a result, the central density is ~0.17 nucleons / fm^3 and the average distance ~1.8fm.

Surface term: $a_s A^{2/3}$

Nucleons at the surface of the nucleus are surrounded by fewer nucleons and the binding energy is reduced

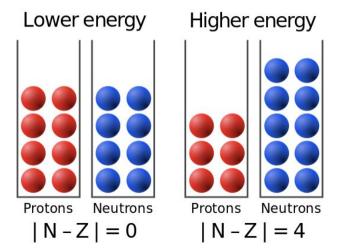
Coulomb term: $a_c \frac{Z^2}{A^{1/3}}$

Repulsive force acting between protons reduces the binding energy

Asymmetry term:

Pauli exclusion principle allows two identical fermions to occupy the same state. Additional particles will occupy higher energy levels. Hence, an asymmetry in number of protons and neutrons reduces the binding energy.

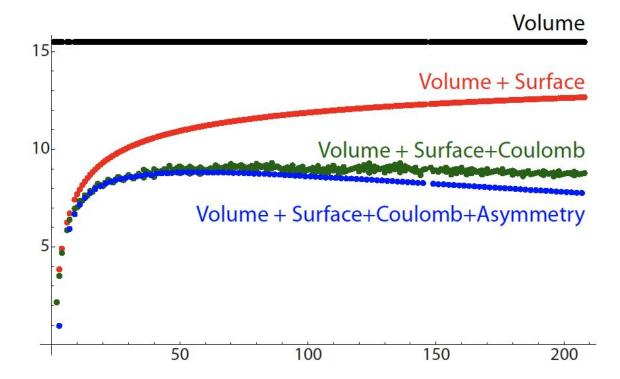
A = 16



Pairing term: $\frac{\delta}{A^{1/2}}$

Again, the Pauli exclusion principle leads to a reduced binding energy when Z and N are not even

$$\delta(A,Z) = egin{cases} \cdot -\delta_0 & Z, N ext{ even} \ (A ext{ even}) \ 0 & A ext{ odd} \ +\delta_0 & Z, N ext{ odd} \ (A ext{ even}) \end{cases}$$



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