

Review of Last Lecture

- Seebeck effect
- Peltier effect
- Thomson effect
- Device analysis
- Figure of merit ZT
- Applications

Thermoelectric Figure of Merit

$$ZT = \frac{\sigma S^2 T}{k} = \frac{\sigma S^2 T}{k_p + k_e}$$

Electrical Conductivity

Seebeck Coefficient

Thermal Conductivity

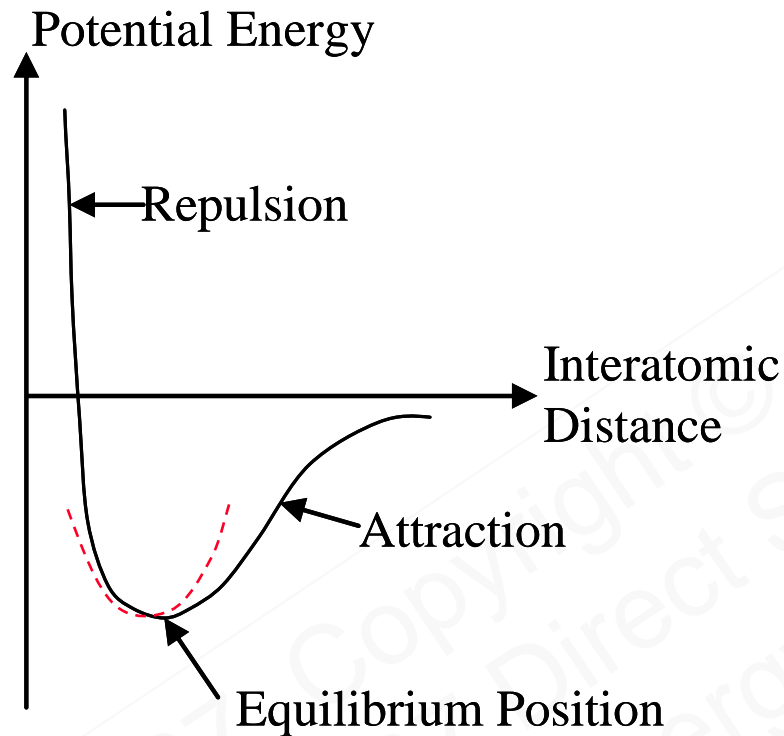
Phonon Contribution

Electron Contribution

Microscopic Formulation of Thermoelectric Properties

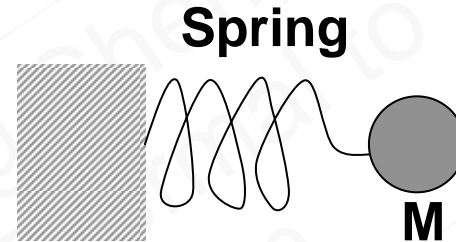
- Review of basic concepts in solid states
- Simple kinetic formulation
- Results from formal transport theory

Atomic Vibration



$$U \approx U_o + K(x - x_o)^2$$
$$F = -\frac{dU}{dx} = -K(x - x_o)$$

- **Classical Oscillator**



- **Natural Frequency**

$$\nu = \frac{1}{2\pi} \sqrt{\frac{K}{M}}$$

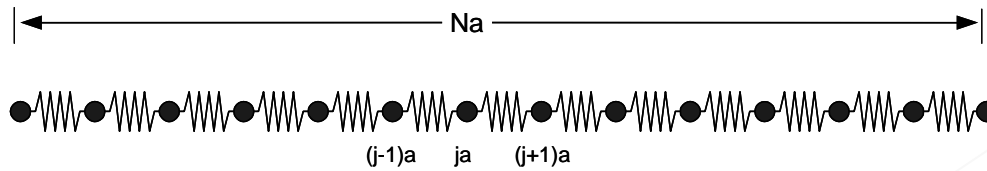
- **From Quantum Mechanics**
Energy of Mode

$$E = \left(n + \frac{1}{2}\right) h\nu \quad n = 0, 1, 2, \dots$$

- **Basic vibrational energy quanta $h\nu$ is called a phonon**

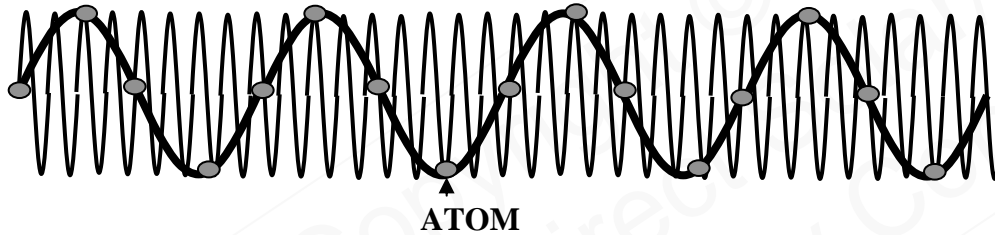
1D Atomic Chain

Monatomic



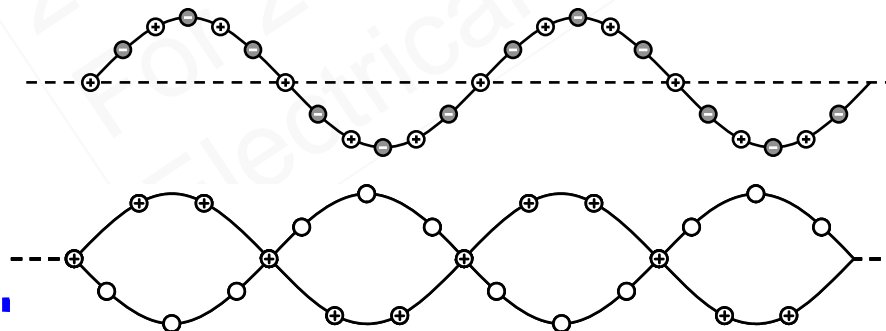
Allowable wavelength $Na = \frac{\lambda}{2}, 2\frac{\lambda}{2}, \dots$

$k = \frac{2\pi}{\lambda} = \frac{2\pi}{2Na}, \dots, \frac{(N-1)a\pi}{2Na}, \frac{\pi}{2a}, \dots$ **Standing Wave Picture**



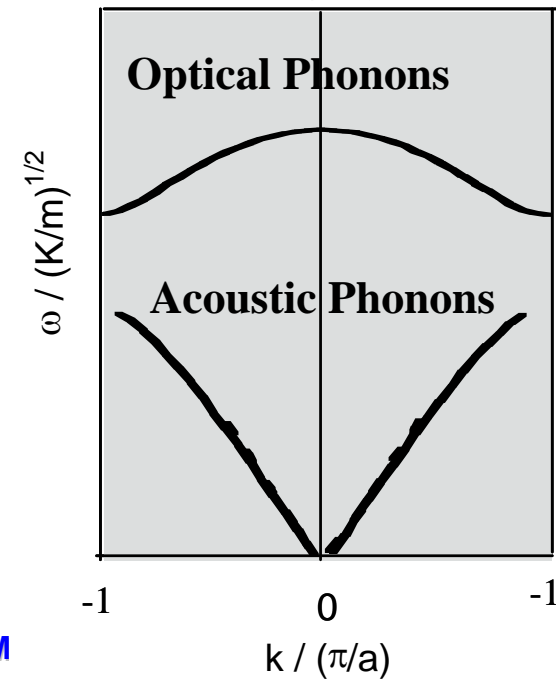
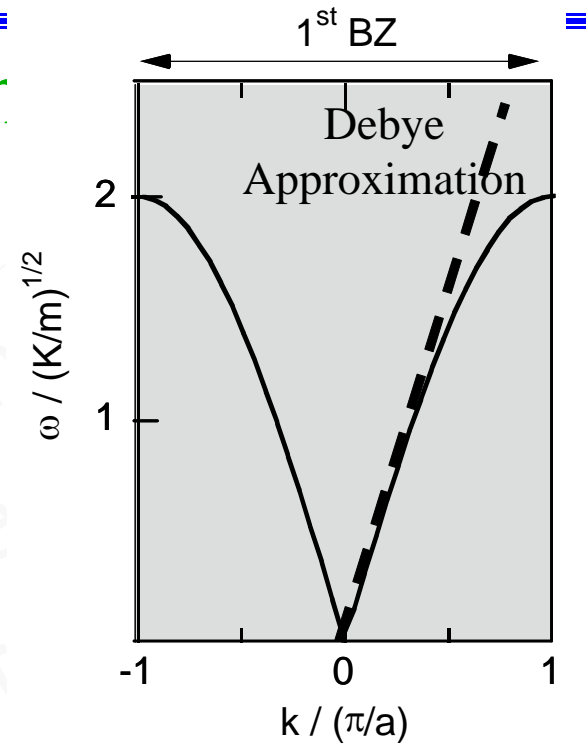
$k = \frac{2\pi}{\lambda} = -\frac{\pi}{a}, -\frac{(N-1)a\pi}{Na}, \dots, \frac{(N-1)a\pi}{Na}, \frac{\pi}{a}$

Diatomic



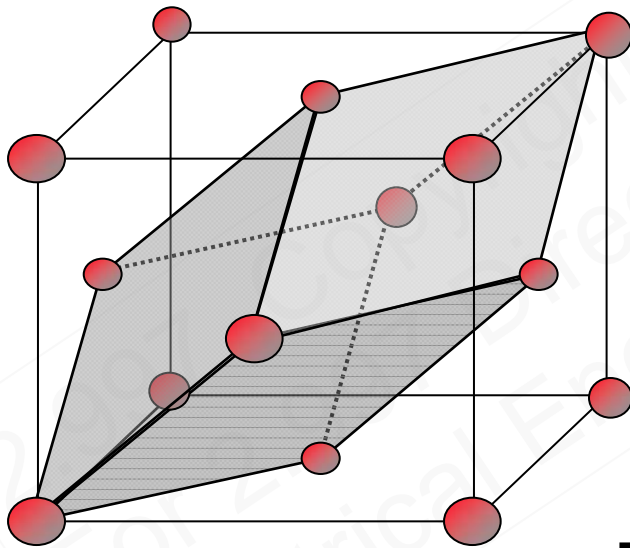
Na

LOW HEAT AND M



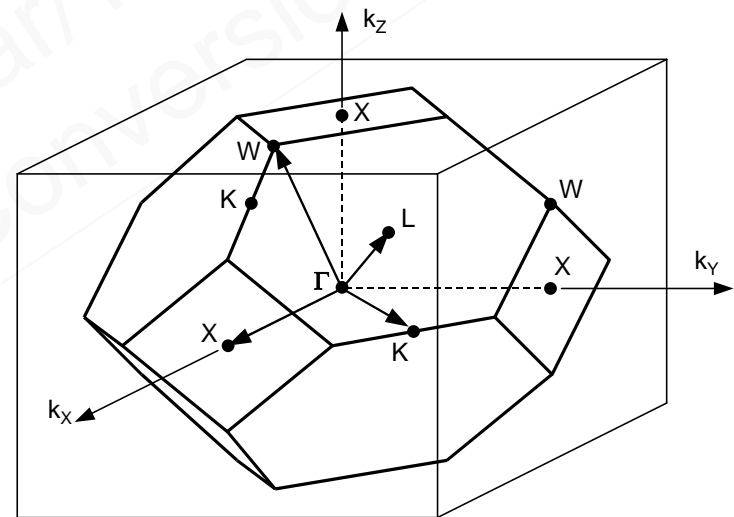
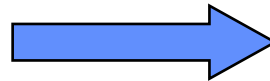
Unit Cell in Real and Reciprocal Spaces

- Periodic signal in time with period T , Fourier transform gives a frequency $\nu=2\pi/T$,
- Periodic signal in space with wavelength λ , Fourier transform gives $2\pi/\lambda$.



**Crystal unit cell
in real space**

Fourier
Transform



Reciprocal Space

Phonons Dispersion in Crystals

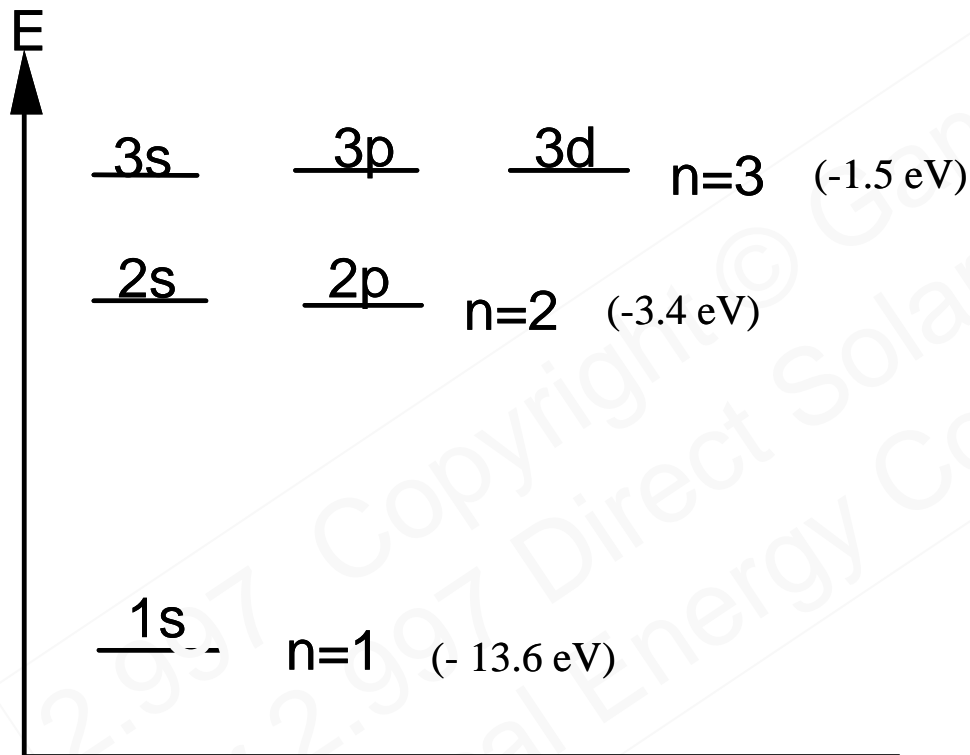
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Please see Fig. 1a and 2a in Giannozzi, Paolo, et al.

"*Ab initio* Calculation of Phonon Dispersions in Semiconductors."

Physical Review B 43 (March 1991): 7231-7242.

Electronic Energy Levels



Hydrogen Atom

- **Wavefunction**

$$\Psi_{nlms}(r, \theta, \varphi)$$

$$n = 1, 2, 3, \dots$$

$$l < n$$

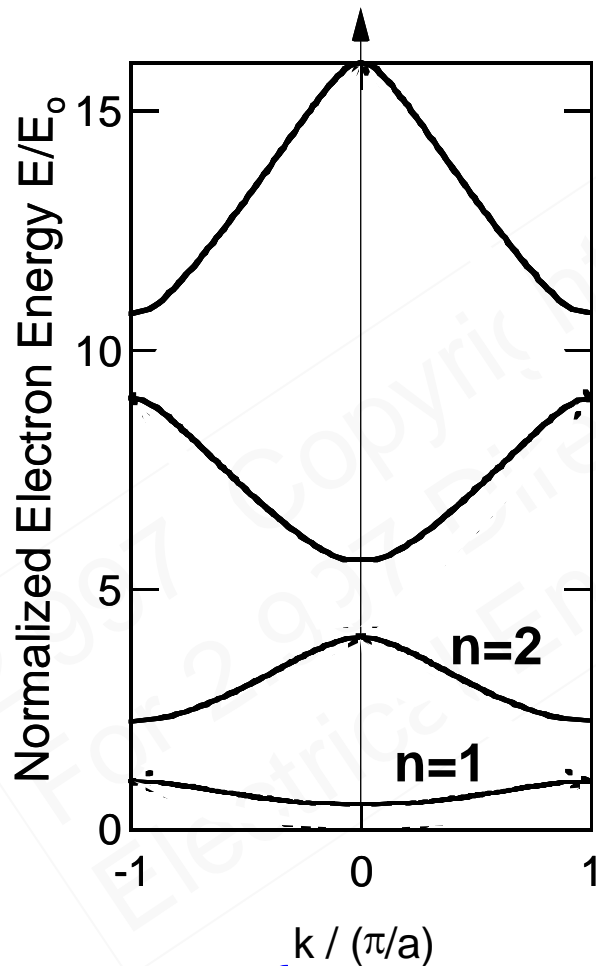
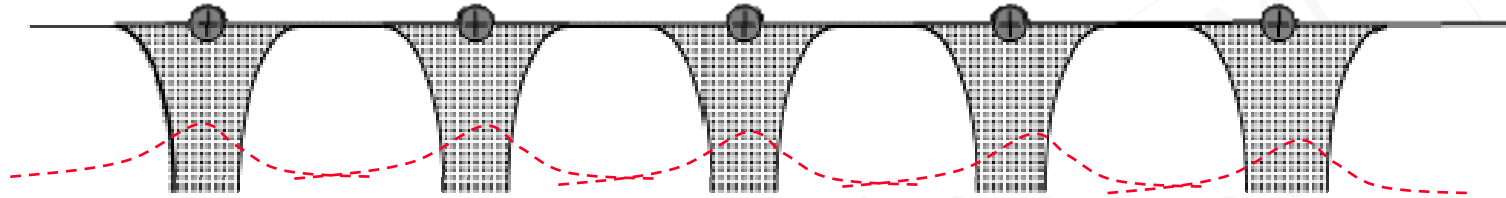
$$|m| \leq l$$

$$s = \pm \frac{1}{2}$$

- **Degeneracy**

$$D = 2n^2$$

Electrons in an Atomic Chain

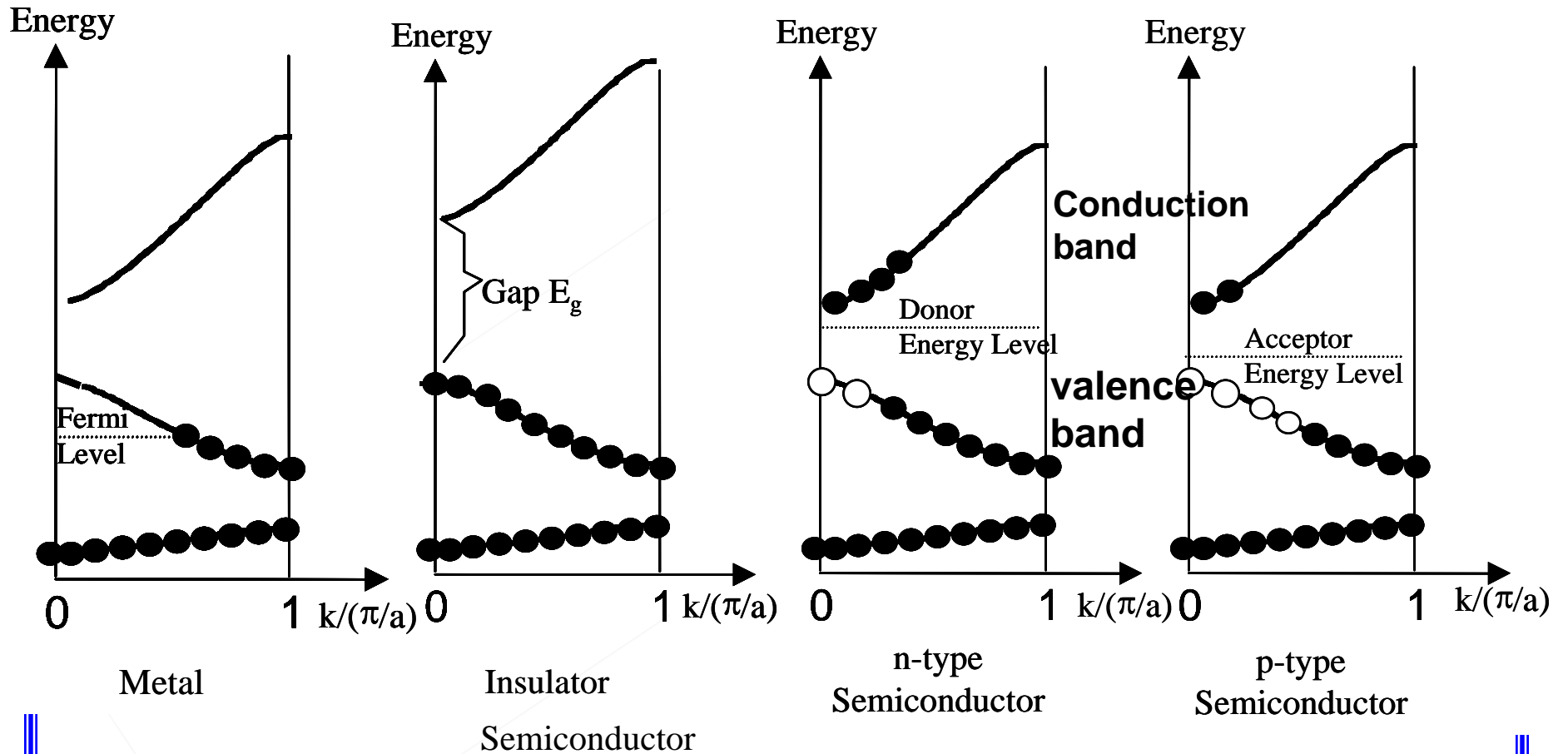


$$\Psi_{ns}(k, x)$$

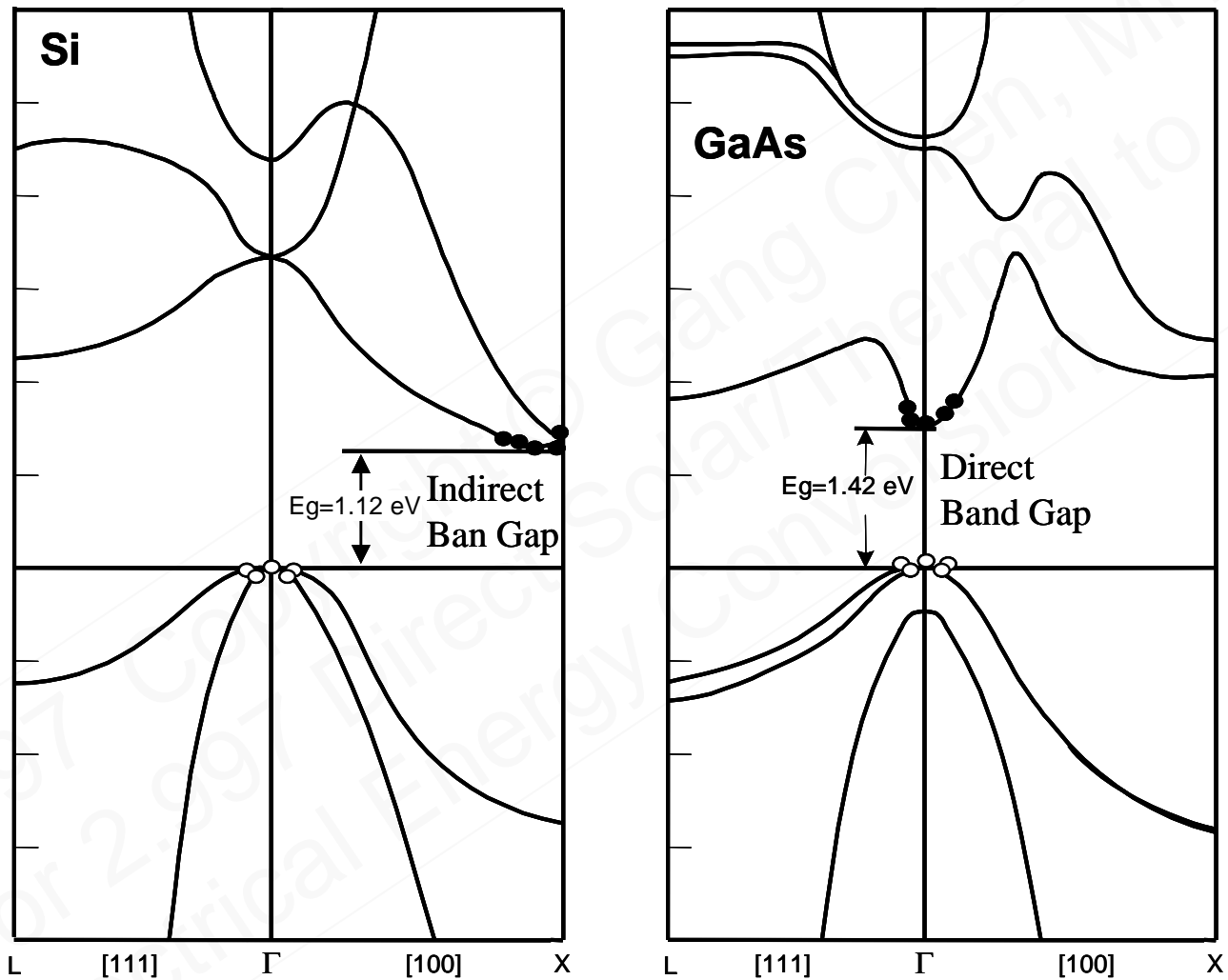
$$k = \frac{2\pi}{\lambda} = -\frac{\pi}{a}, -\frac{(N-1)a\pi}{Na}, \dots, \frac{(N-1)a\pi}{Na}, \frac{\pi}{a}$$

**k has N discrete values
between $(-N/2, N/2)$**

Different Solids



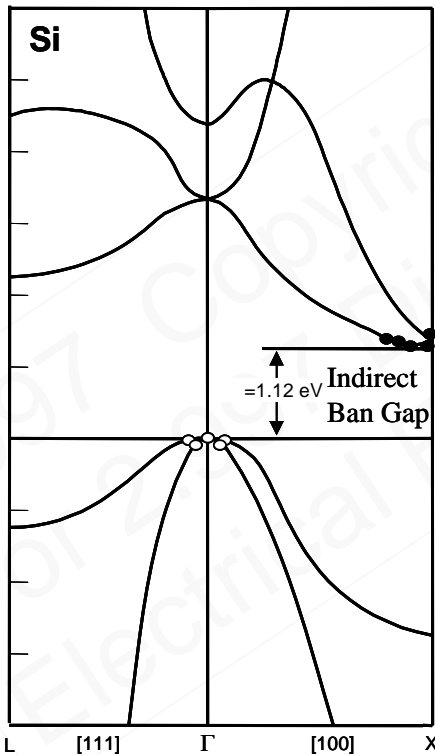
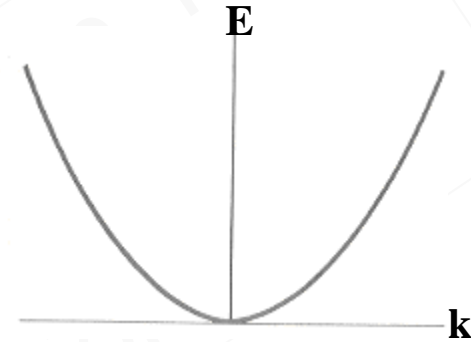
Electronic Band Structures of Real Crystals



Parabolic Band Approximation

- Free Electrons

$$E = mv^2/2 = p^2/2m = \hbar\mathbf{k}^2/2m$$



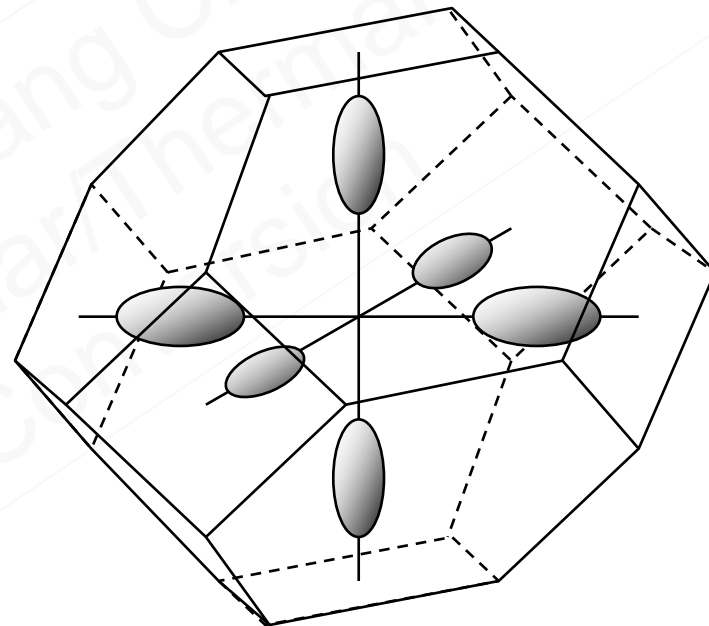
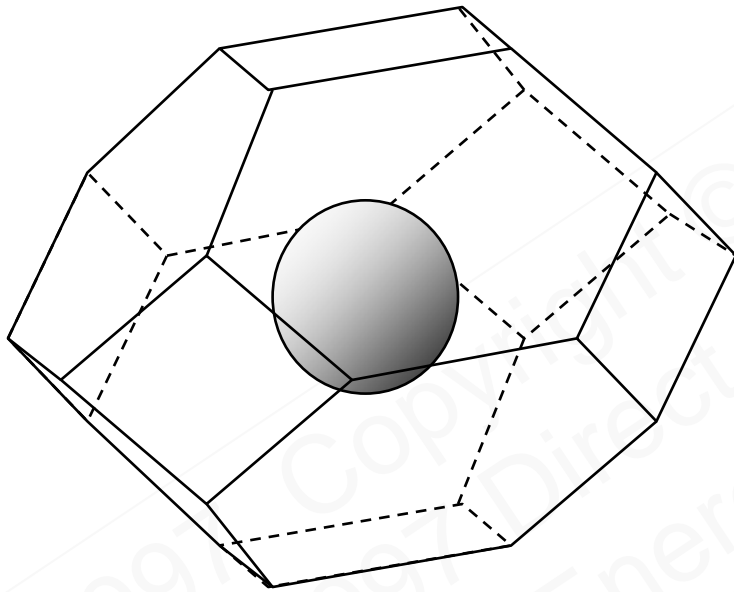
- Near Minimum (Maximum)

$$E - E_c = \frac{\hbar^2}{2} \left(\frac{k_x^2}{m_{11}} + \frac{k_y^2}{m_{22}} + \frac{k_z^2}{m_{33}} \right)$$

Effective mass

$$m_{ij} = \hbar^2 / \left(\partial^2 E / \partial k_i \partial k_j \right)$$

Constant Energy Surface



Statistical Distributions

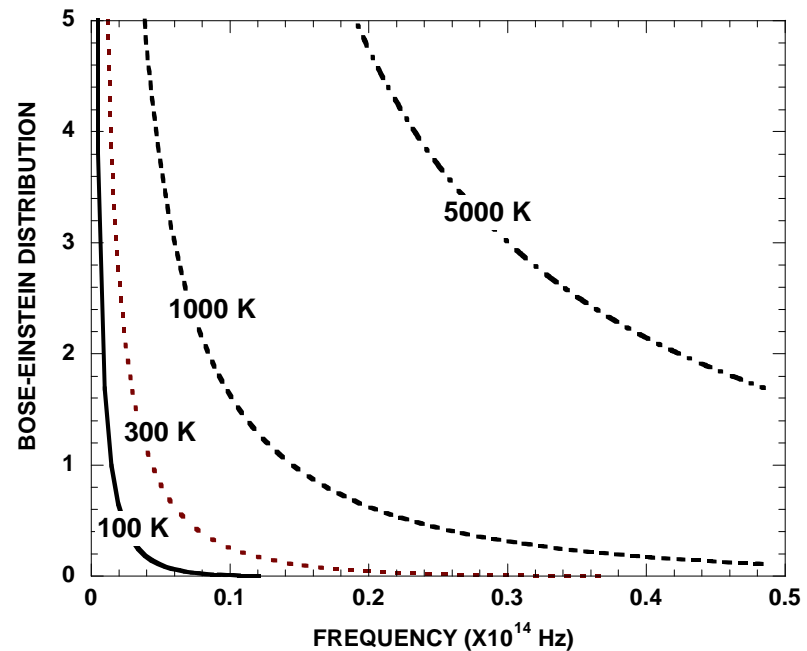
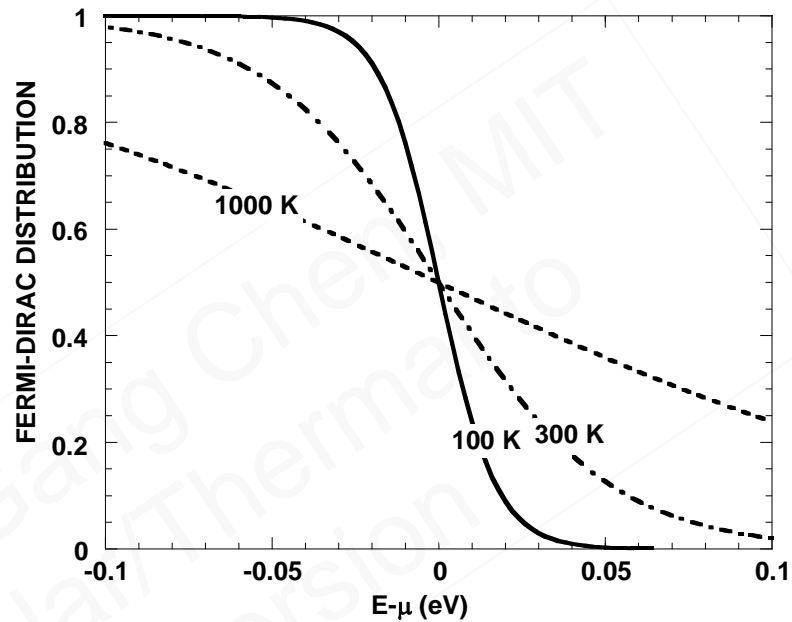
Average Number of Particles in a Quantum State

Fermi-Dirac

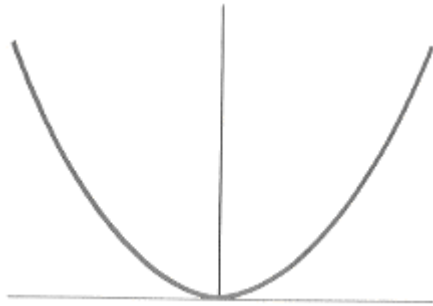
$$f = \frac{1}{\exp\left(\frac{E - \mu}{k_B T}\right) + 1}$$

Bose-Einstein

$$f = \frac{1}{\exp\left(\frac{E - \mu}{k_B T}\right) - 1}$$



Electron Density



$$E - E_c = \frac{\hbar^2 (k_x^2 + k_y^2 + k_z^2)}{2m}$$

$$N = 2 \sum_{-N_x/2}^{N_x/2} \sum_{-N_y/2}^{N_y/2} \sum_{-N_z/2}^{N_z/2} f(E, T)$$

$$= 2 \int_{-\pi/a}^{\pi/a} \frac{dk_x}{(2\pi/L_x)} \int_{-\pi/a}^{\pi/a} \frac{dk_y}{(2\pi/L_y)} \int_{-\pi/a}^{\pi/a} \frac{dk_z}{(2\pi/L_z)} f(E, T)$$

$$= \frac{2V}{8\pi^3} \int_{-\pi/a}^{\pi/a} \int_{-\pi/a}^{\pi/a} \int_{-\pi/a}^{\pi/a} dk_x dk_y dk_z \exp\left[-\frac{E - \mu}{k_B T}\right]$$

Electron Density

$$\begin{aligned}n &= \frac{N}{V} = \frac{2V}{8\pi^3} \int_{E_c}^{\infty} 4\pi k^2 dk \exp\left[-\frac{E - \mu}{k_B T}\right] \\&= \frac{1}{\pi^2} \int_{E_c}^{\infty} \left(\frac{2m(E - E_c)}{\hbar^2}\right) d\sqrt{\left(\frac{2m(E - E_c)}{\hbar^2}\right)} \exp\left[-\frac{E - \mu}{k_B T}\right] \\&= \int_{E_c}^{\infty} \left(\frac{\sqrt{2m}^{3/2} \sqrt{E - E_c}}{\pi^2 \hbar^3}\right) \exp\left[-\frac{E - \mu}{k_B T}\right] dE = \int_{E_c}^{\infty} D(E) \exp\left[-\frac{E - \mu}{k_B T}\right] dE \\&= 2 \left(\frac{2\pi m^* \kappa_B T}{\hbar^2}\right)^{3/2} \exp\left(-\frac{E_c - \mu}{k_B T}\right) \\&= N_c \exp\left(-\frac{E_c - \mu}{k_B T}\right)\end{aligned}$$

↑
**Density of States D(E):
Number of quantum
states per unit volume
and per energy interval**

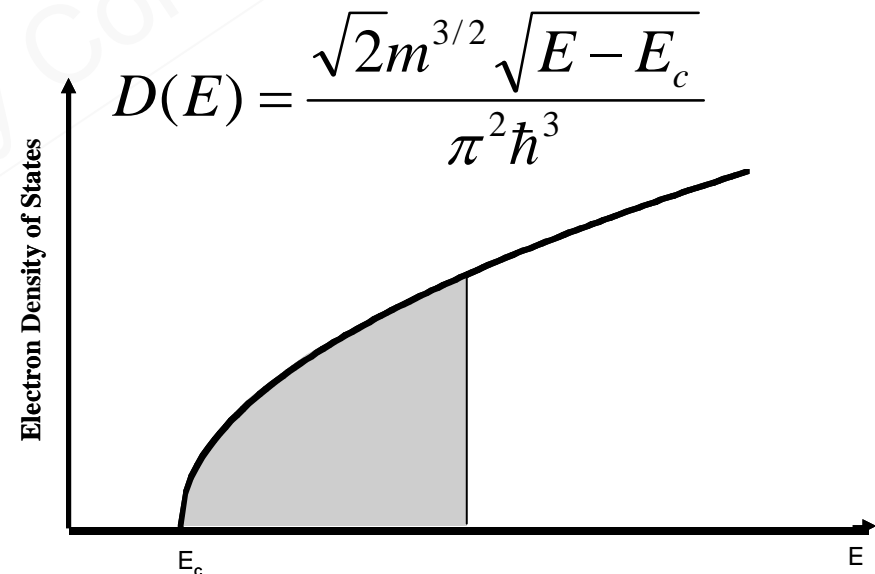
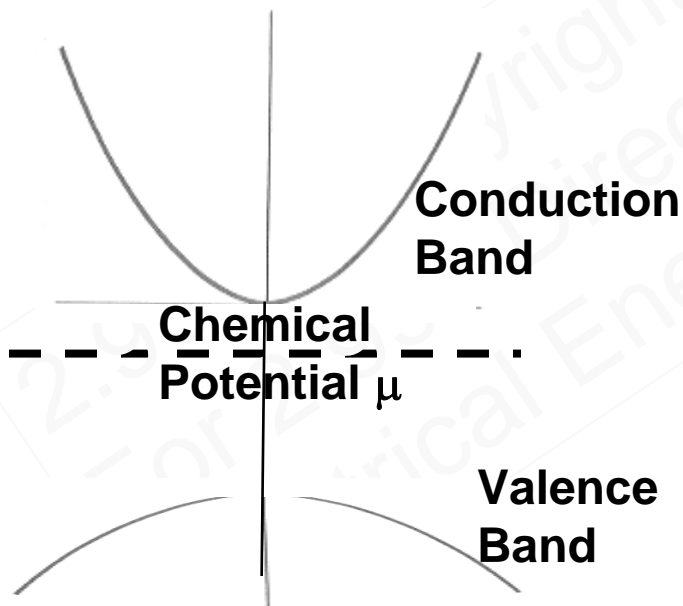
Electron Density

General:

$$n = \int_{E_c}^{\infty} D(E) f(E, \mu, T) dE$$

$$n = 2 \left(\frac{2 \pi m^* \kappa_B T}{h^2} \right)^{3/2} \exp\left(-\frac{E_c - \mu}{k_B T} \right) = N_c \exp\left(-\frac{E_c - \mu}{k_B T} \right)$$

Under Boltzmann Statistics



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